

# Fast Hydrolysis for Determination of Lipopeptides on Solid Surface

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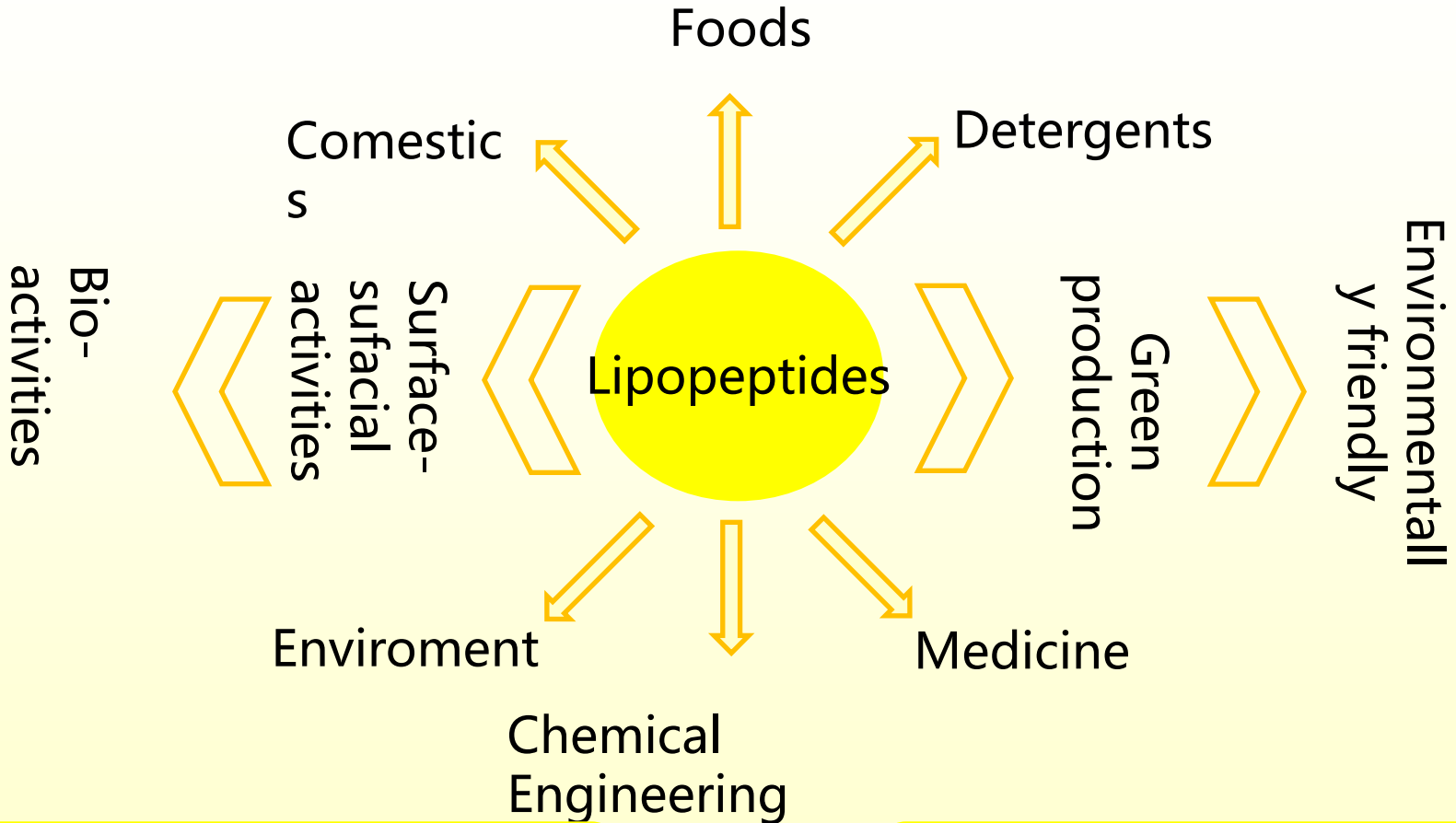
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- 1 Introduction**
- 2 Experiment**
- 3 Results and Discussion**
- 4 Conclusion**



# 1 Introduction



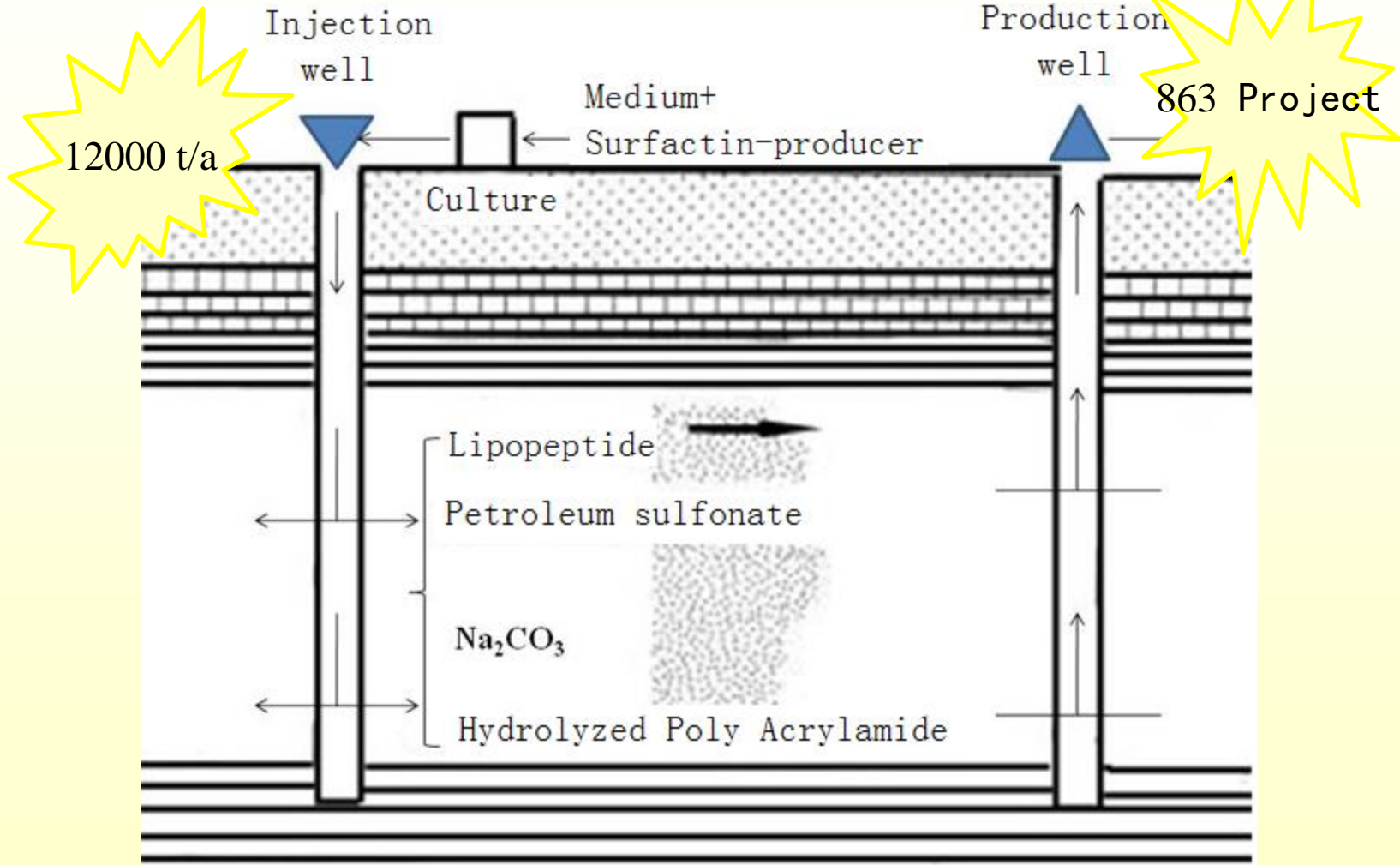
Low contents in samples

Mixture

No sensitive methods

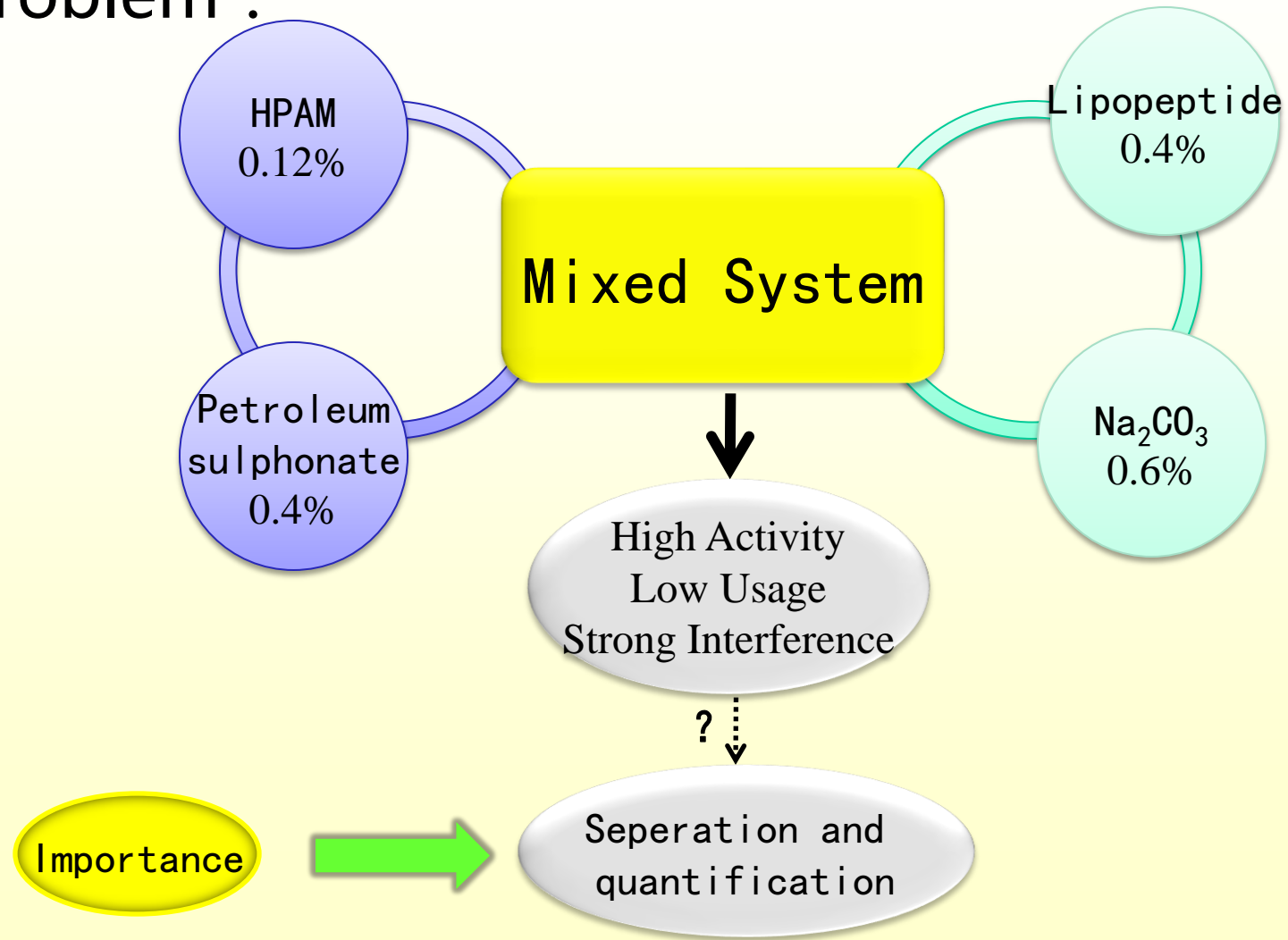
# 1 Introduction

Surfactin, a lipopeptide was used in EOR



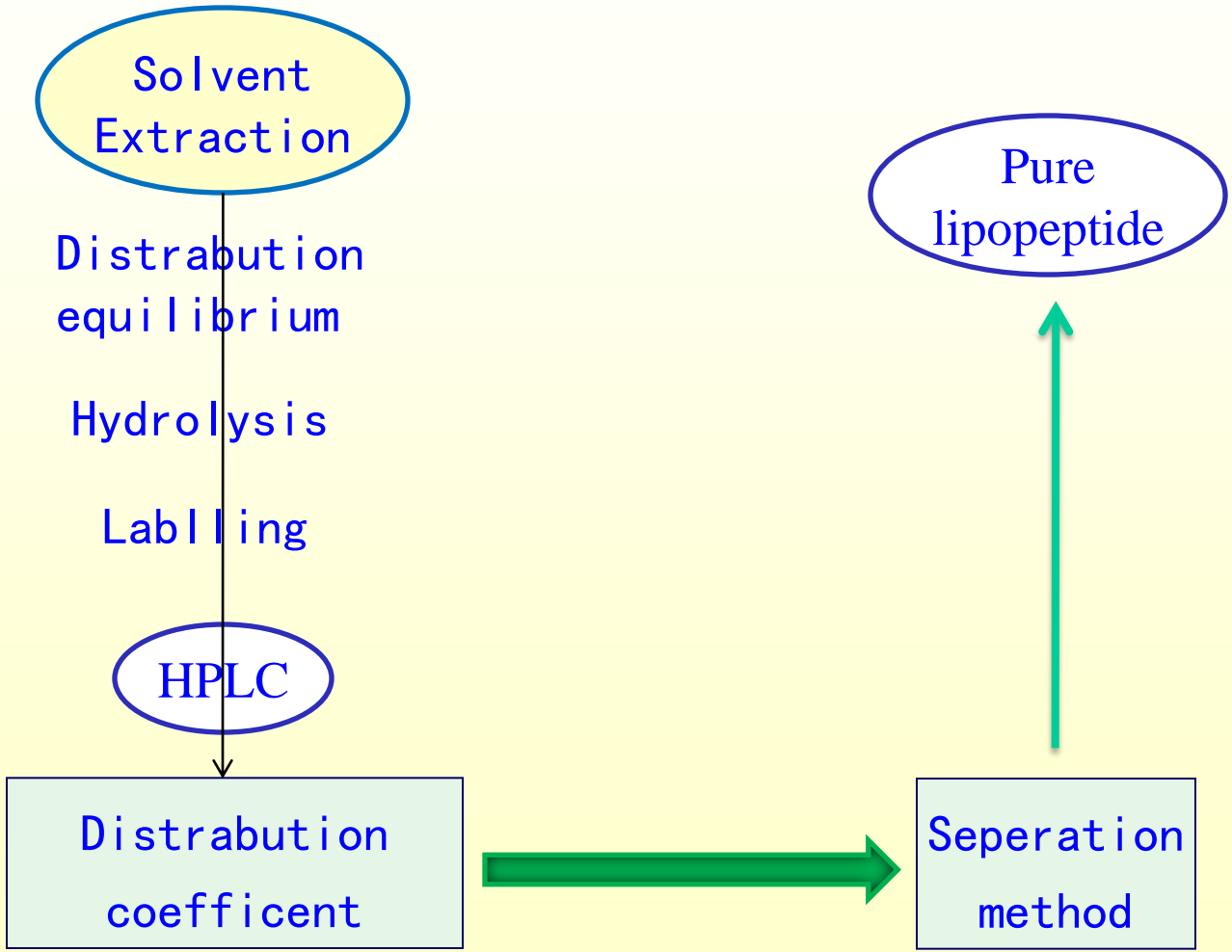
# 1 Introduction

Problem :



# 1 Introduction

Seperation method was established



# 1 Introduction

## 1.2 Lipopeptide determination

### Seperation

- Acid precipitation
- Sovent extraction
- Chromatography

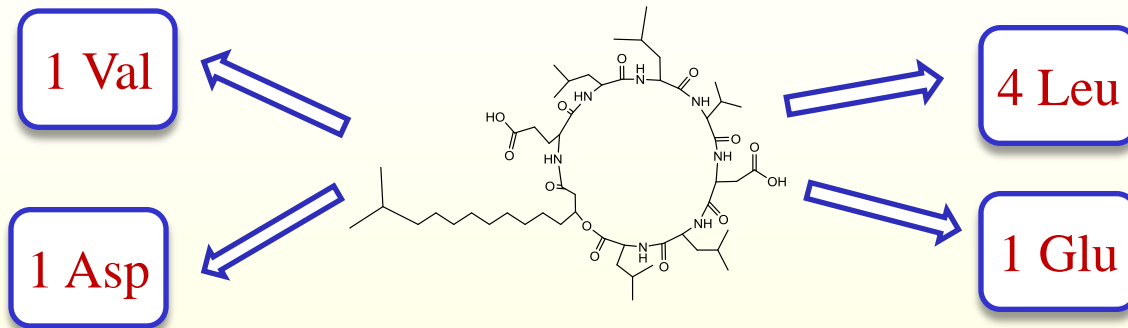
### quantification

- **HPLC**
  - Natuatural ultraviolet absorbence
  - Low sensitivity
  - Needing pure standard
- **Spectrometry of Amino acid**
  - Solvent seperation
  - Hydrolysis
    - Total gross
- **GC-MS**
  - Hydrolysis
    - Trimethylsilylation

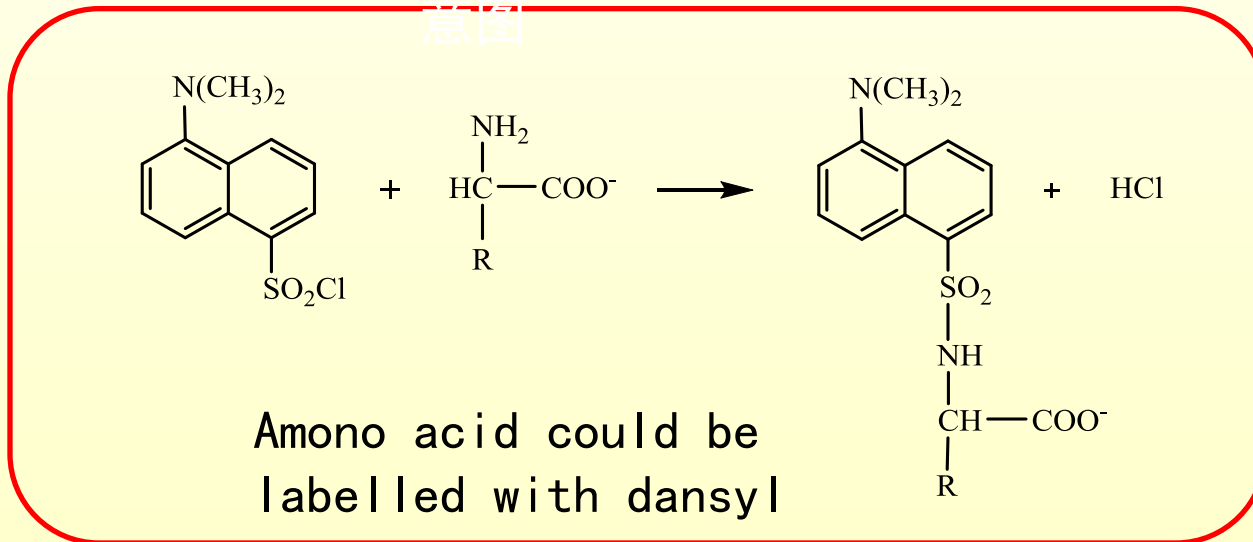


# 1 Introduction

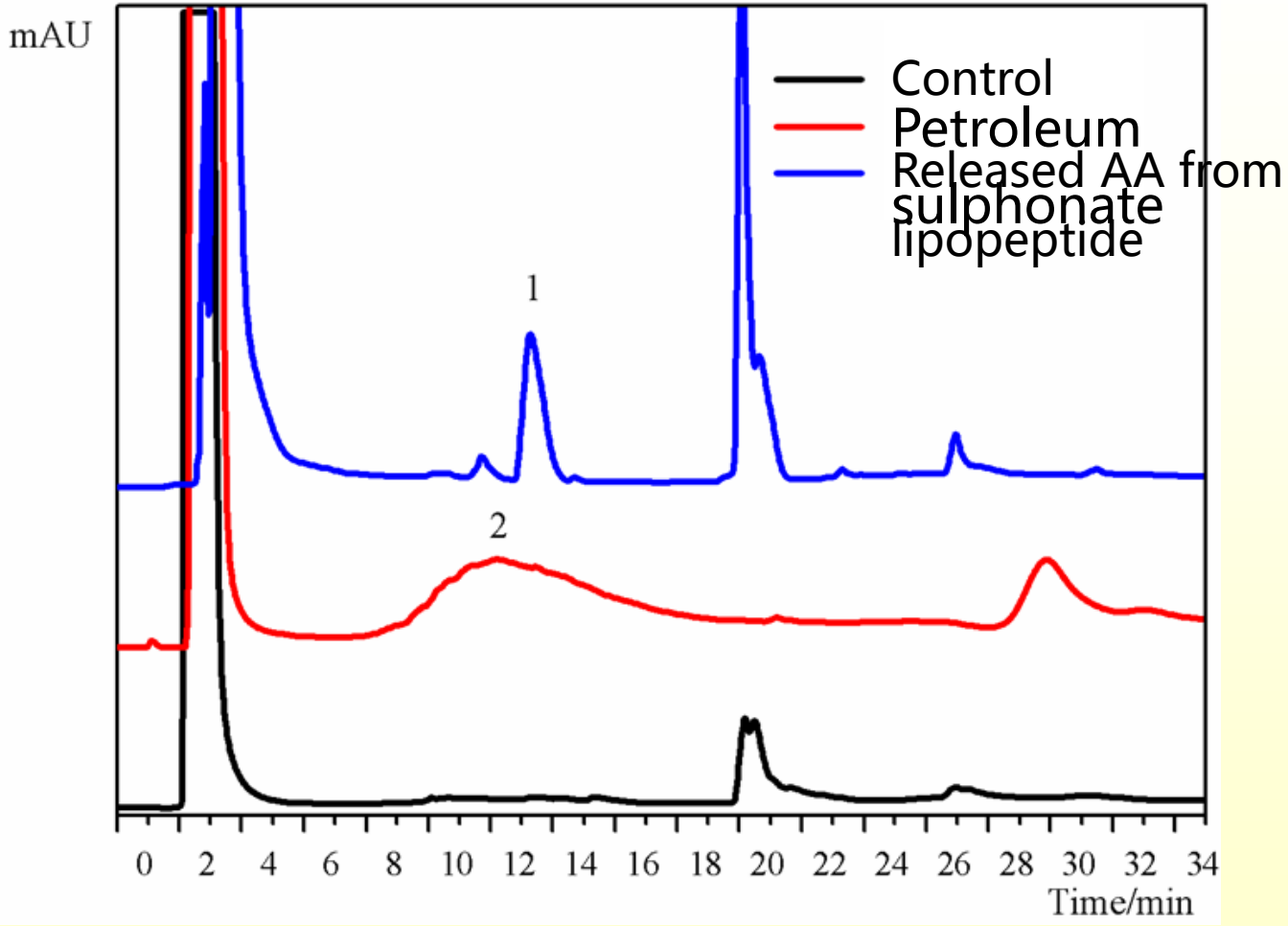
## Object



脂肽水解成氨基酸示意图



# 1 Introduction



HPLC of Lablled Amino Acids Released by Hydrolysis of Lipopeptide



# 1 Introduction

## 1.3 Lipopeptide Hydrolysis

Traditional method stems from protein hydrolysis

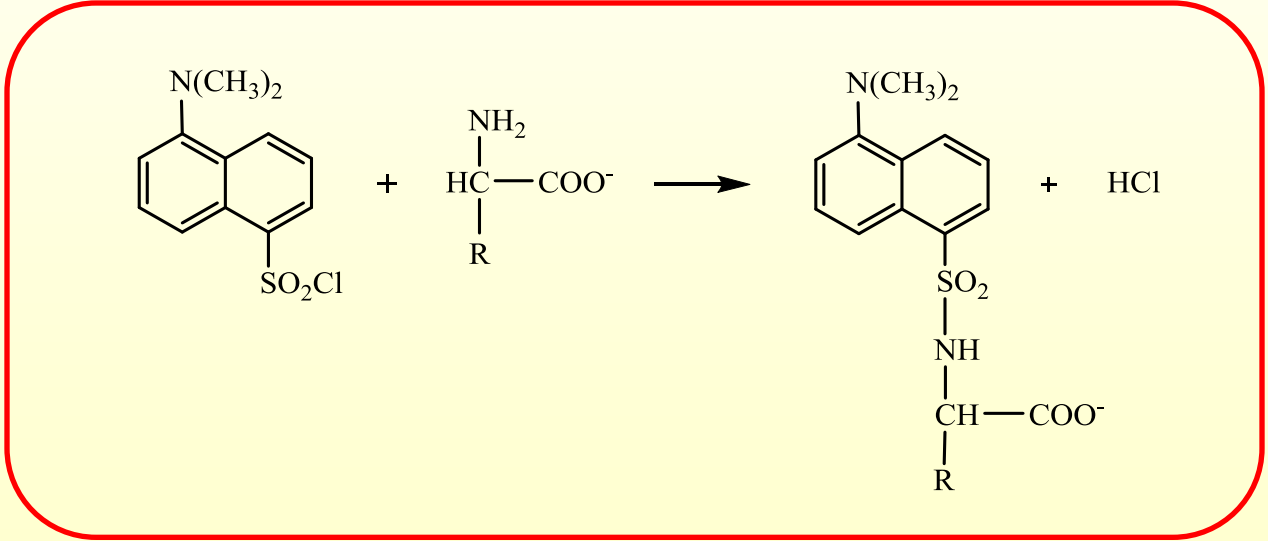
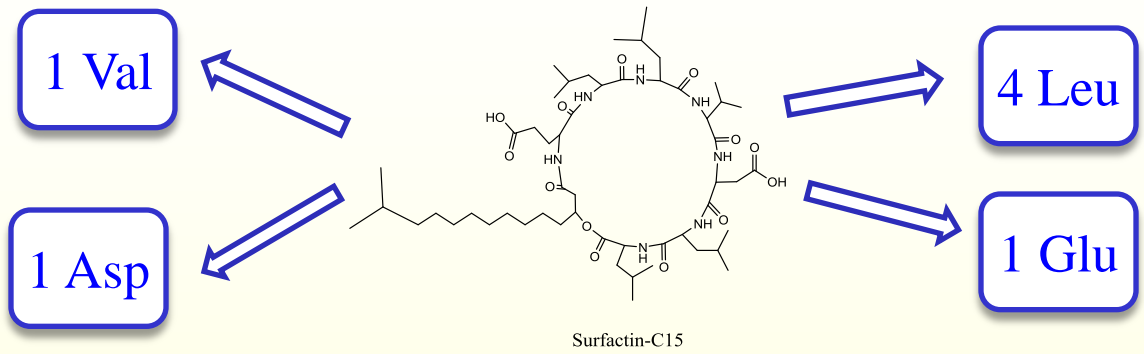
Hydrolysis Condition: in 6 M HCl at 110°C for 24 h

It has been improved to

Hydrolysis Condition: in 6M HC at 90°C for 24 h

At relative low temperature, but for **long time** and needing **post treatment** of evaporating the acid water solution

# 2 Experiment



Amino acids will be **labeled** by dansyl

# 2 Experiment

Stainless Steel Reaction Kettle

lined PTFE

Sample was stained

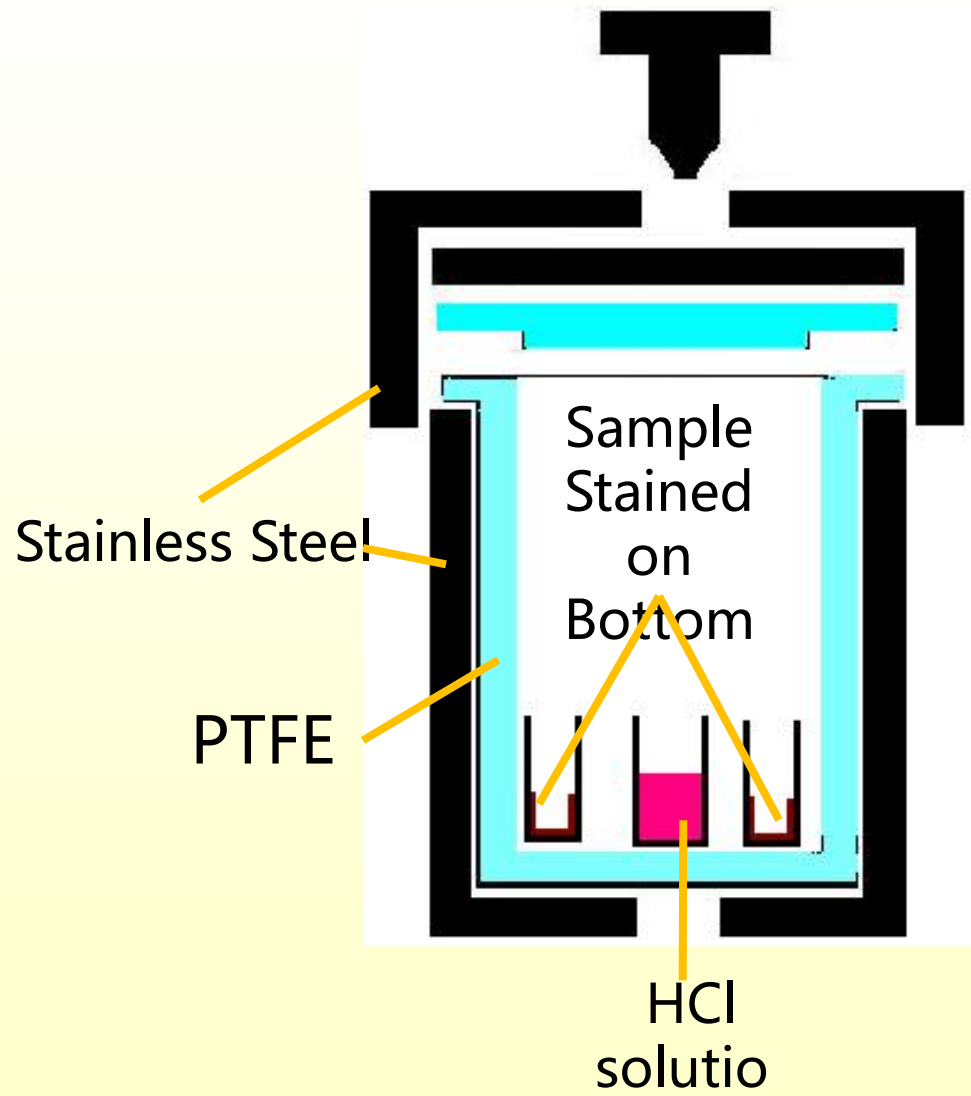
HCl solution was filled separately

Sealed

Heated

Lablled directly

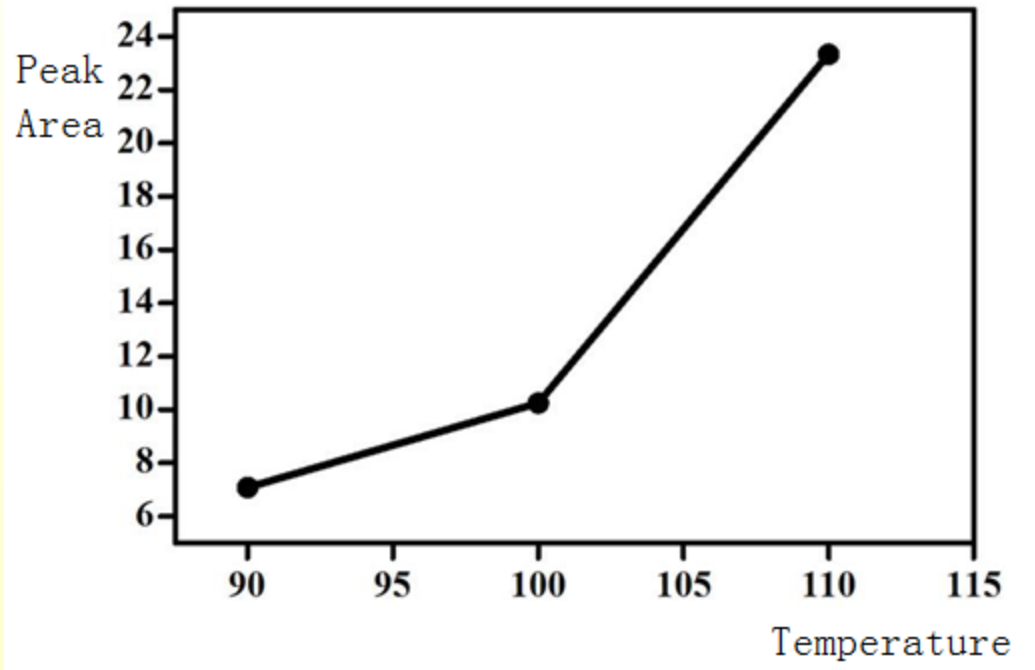
Determined



# 3 Results and Discussion

Orthogonal experimental design for optimal condition

## 3.1 Effect of Temperature on Peak Area

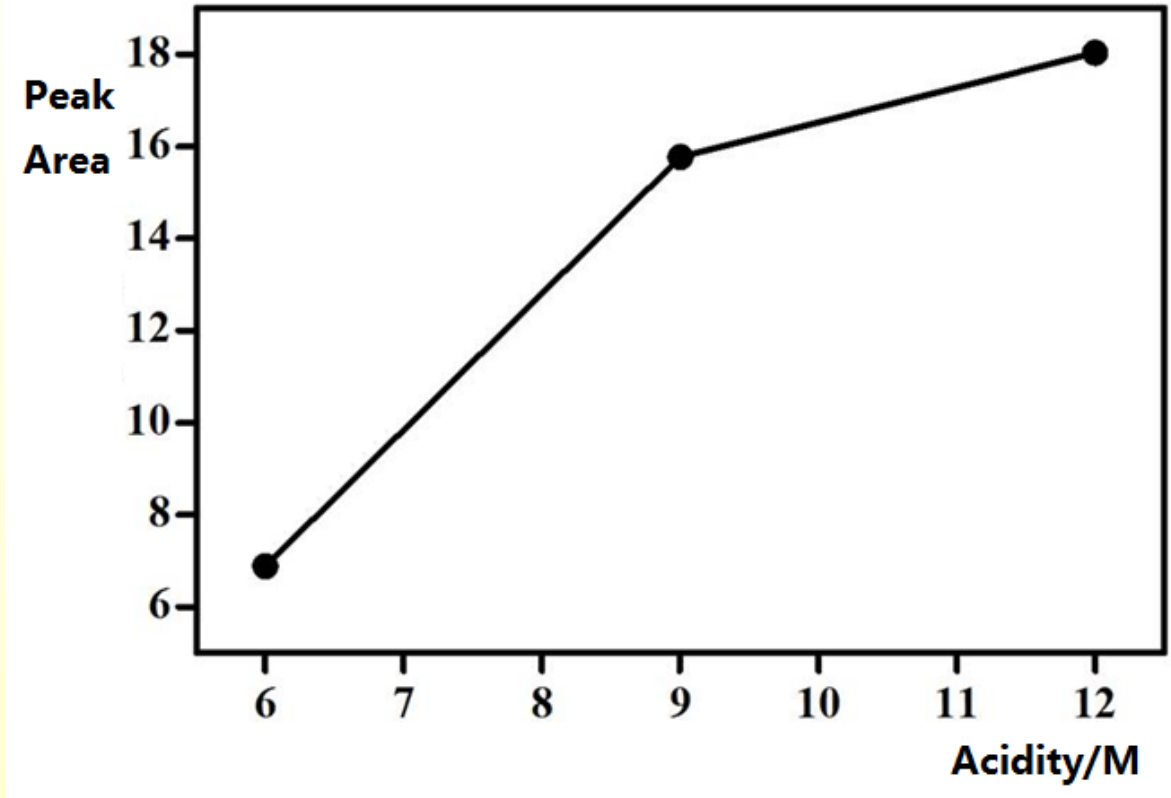


Index increased sharply with temperature

increase

# 3 Results and Discussion

## 3.2 Effect trend of acid concentrations

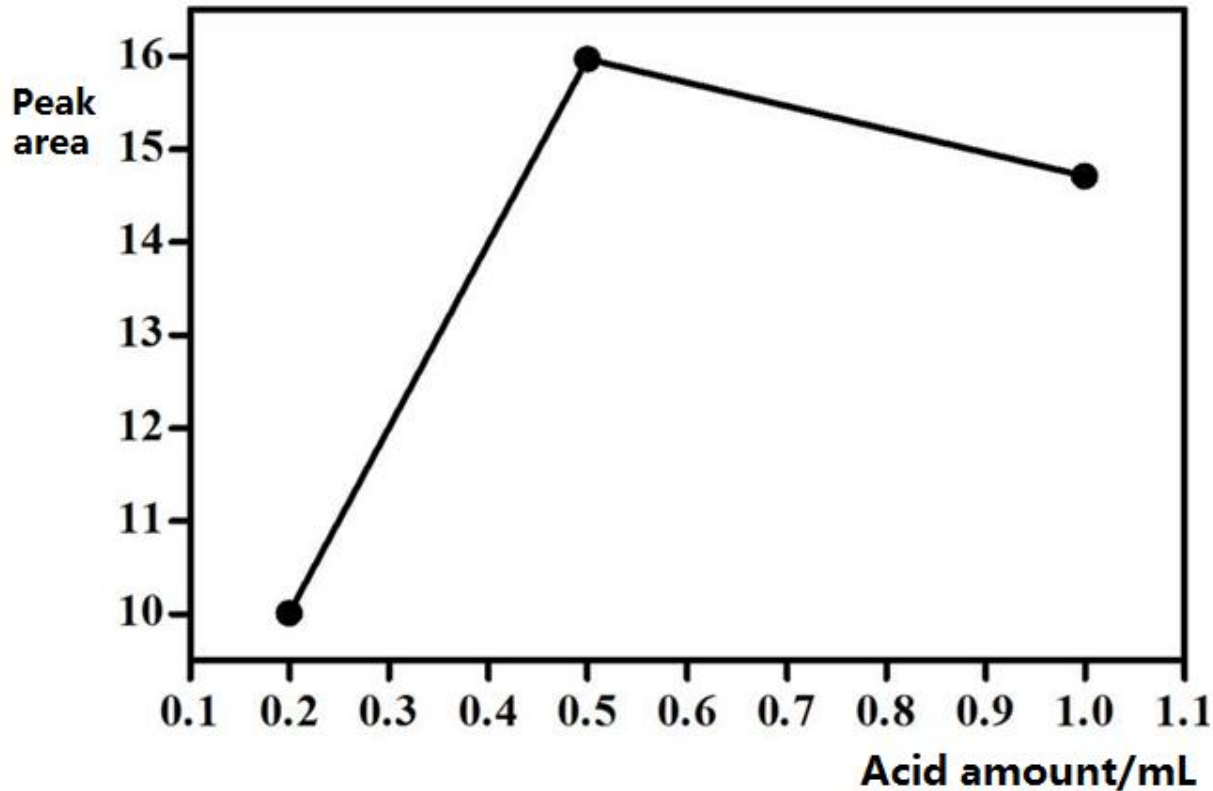


Index increased sharply then slowly with acidity

increase

# 3 Results and Discussion

## 3.3 Effect trend of acid quantities



Index increased sharply then decreased slowly with acid amount





## 3 Results and Discussion

The deviations were 18, 12 and 6 for **temperature**, **acidity** and acid amount, respectively.

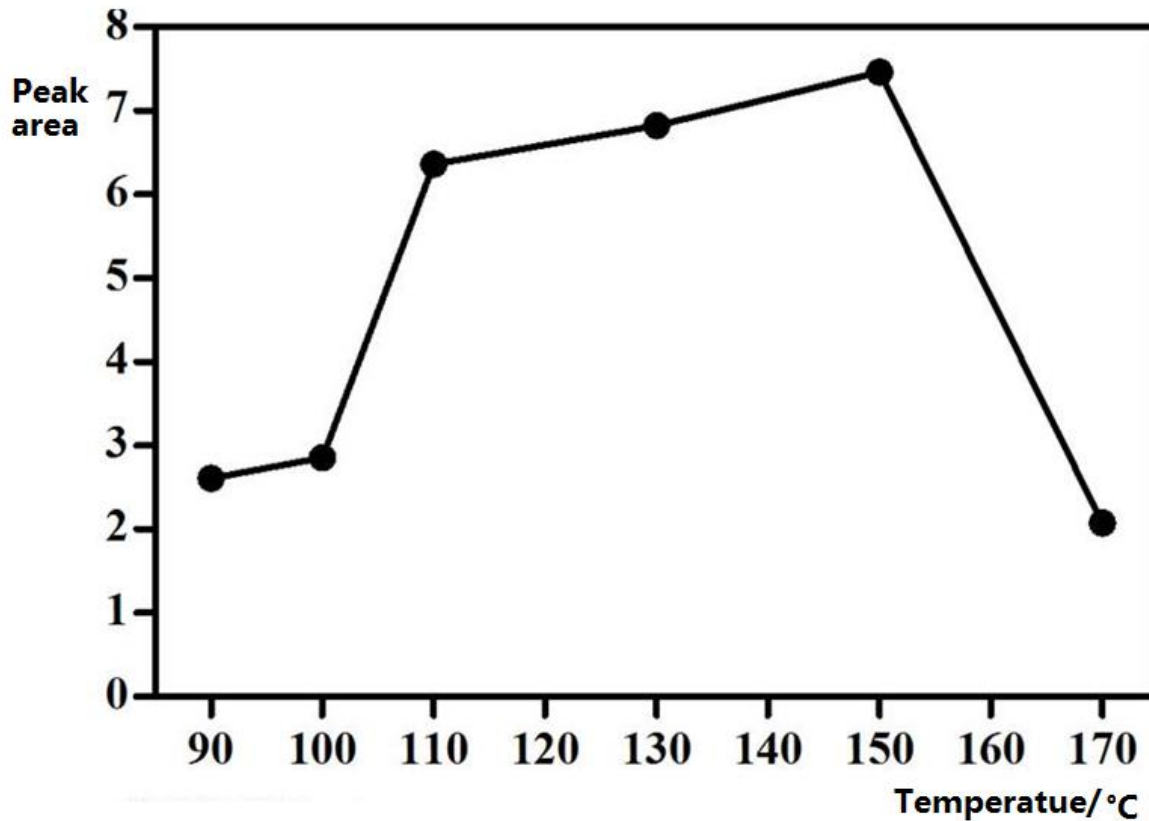
The acidity was 12 M, could not increase.

The acid amount should be fixed at 0.5 mL.

The hydrolysis temperature and time will be **modified for good results**

# 3 Results and Discussion

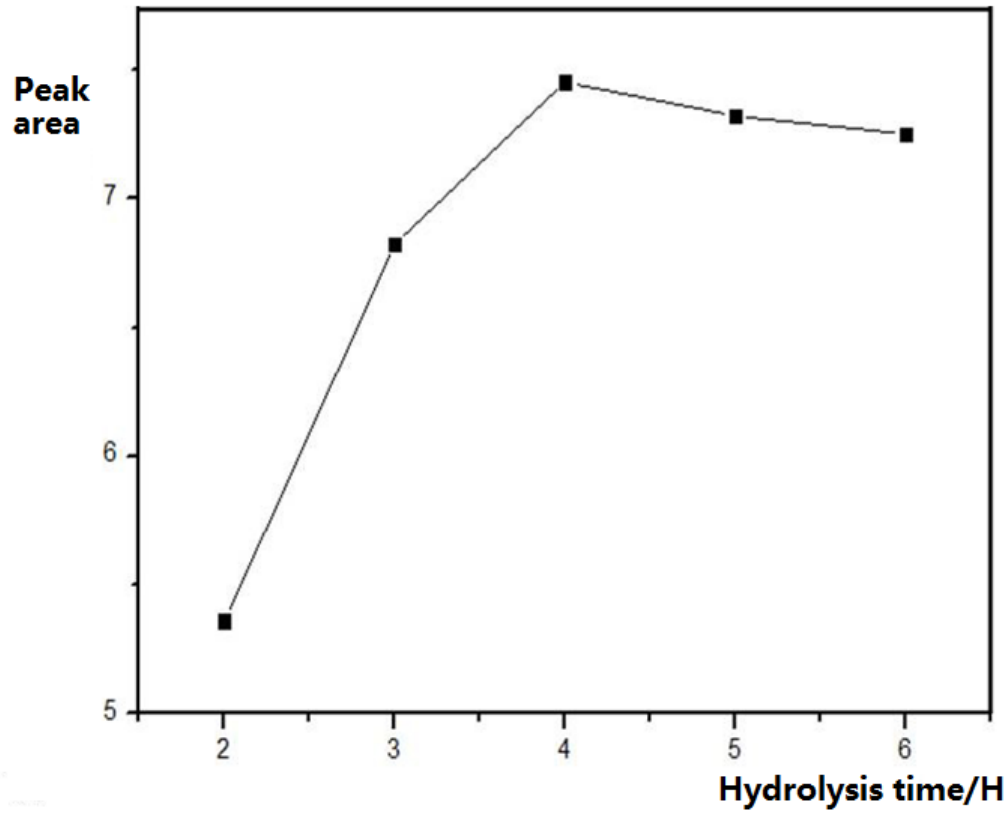
At different temperature using 0.50 mL 12 M HC for 4 H



Index was allmost stable between 110-150 celsius degree

# 3 Results and Discussion

For defferent time at 150 °C using 0.50 mL 12 M HC



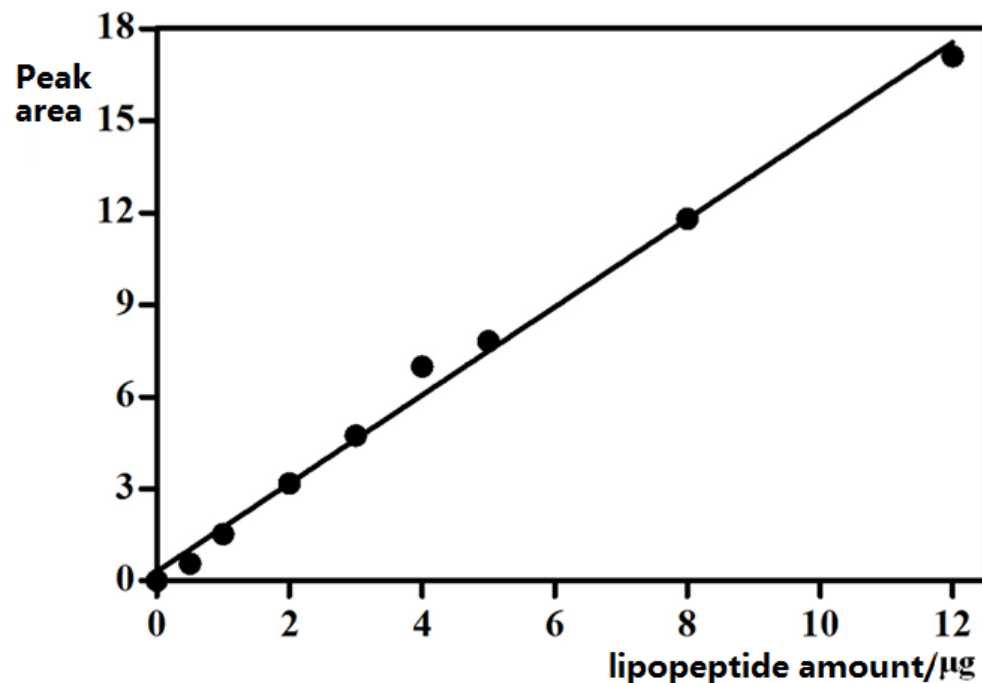
Index was allmost stable after 4 h

hydrolysis

# 3 Results and Discussion

## 3.4 good results at optimal condition

0.5 mL of 12 M HCl , hydrolysis at 150°C for 4 h



Working curve

Good linear relationship  
could be obtained

Linear equation:  $y=12.15c$

Linear coefficient:  $r^2=0.9940$



# 3 Results and Discussion

## 3.5 Results of Determining Samples

Hydrolysis was applied in determining samples

Determination results using hydrolysis

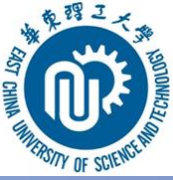
method	
Parallel samples	$n=5$
Mean value	7.05
S	0.47
RSD(%)	6.63
Working curve	$y=12.15c$
Recovery (%)	92.1



## 4 Conclusion

Acidity, acid amount, temperature and time were evaluated through orthogonal experimental design to resolve the long time and post-treatment of lipopeptide hydrolysis

The optimal condition of hydrolysis was determined: **at 150°C and 0.50 mL 12 M HCl for 4 h**, conducted on gas-solid surface



The hydrolysis **time** of lipopeptides was shortened from 24 to **4 h**

The hydrolysis **post-treatment**, such as evaporating solvent, was **eliminated**

# Thank You!

敬请指正!



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Engineering Research Center for MEOR, MOE

