

# Green Chemistry Innovation in the Synthesis of Medicines

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International Summit on Past and Present Research Systems  
of Green Chemistry  
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**Hilton Philadelphia Airport, USA**


<http://www.drreddys.com/products/green-chemistry.html> (Green Chemistry Website)

*Nature Medicine* **2013**, 19, 1200-1203 (Finding Right Chemistry)

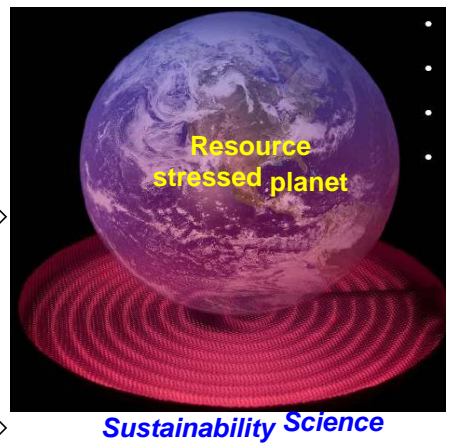
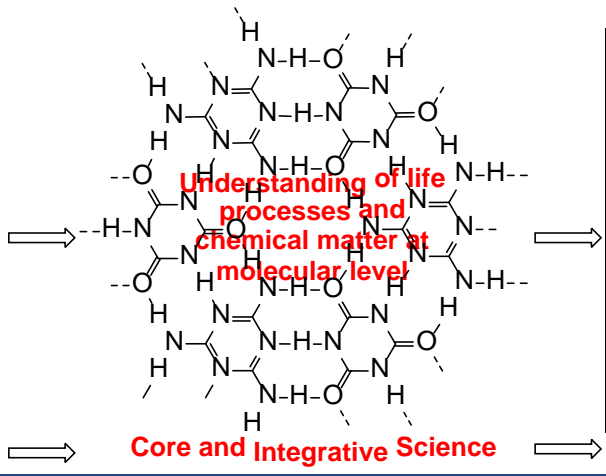
# Significance of Chemistry

- Whatever you hear, see, smell, taste, and touch involves chemistry and chemicals (matter).
- And all these processes involve intricate series of chemical reactions and interactions in the biological system.
- With such an enormous range of biological actions which are governed by chemistry therefore it is essential to know about this subject at some level in order to understand the world around us.

Green Tea   
>200

Coffee   
>1000  
Chemicals

Cigarette   
>7000



-To quote Linus Pauling, Nobel Laureate in chemistry, from a 1983 UC Berkeley lecture:

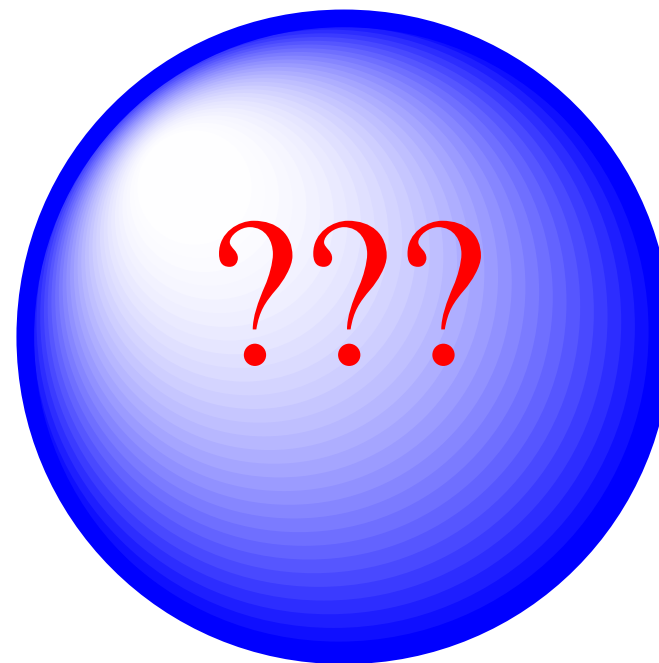
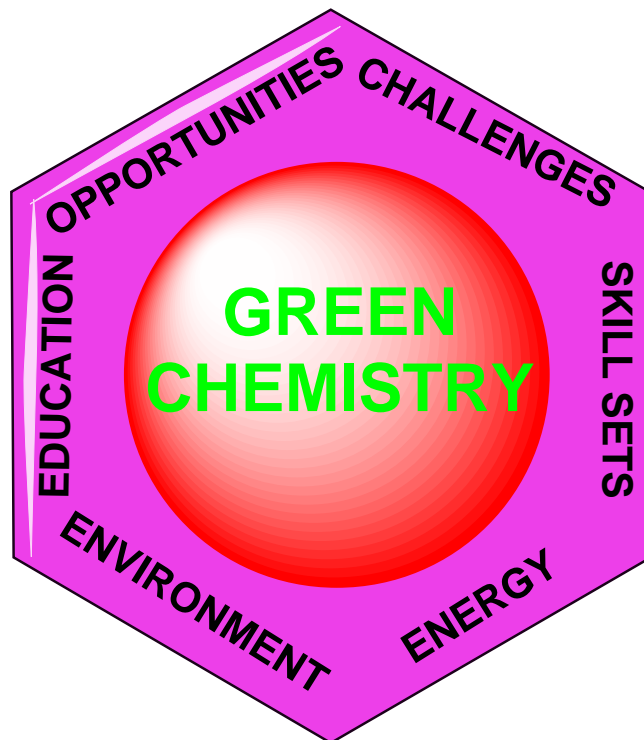
***Chemistry is wonderful! I feel sorry for people who don't know anything about chemistry. They are missing an important source of happiness -- that of satisfying one's intellectual curiosity. The world is wonderful. Chemistry is an important part of it.***

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## **Chemistry Signifies Love and Hate Relationship**

*-can't live without it but can't accept everything that it has*

- Evolving discipline it does mean that the definition of green may change tomorrow e.g. Grignard reaction was considered to be one of the best reactions but today it is being replaced with greener metal catalyzed transformations.
- It is a subject that deals with prevention of waste in any activity around us by design.





Energy alternatives

Textile, printing, agro and construction



Electronics and semiconductors

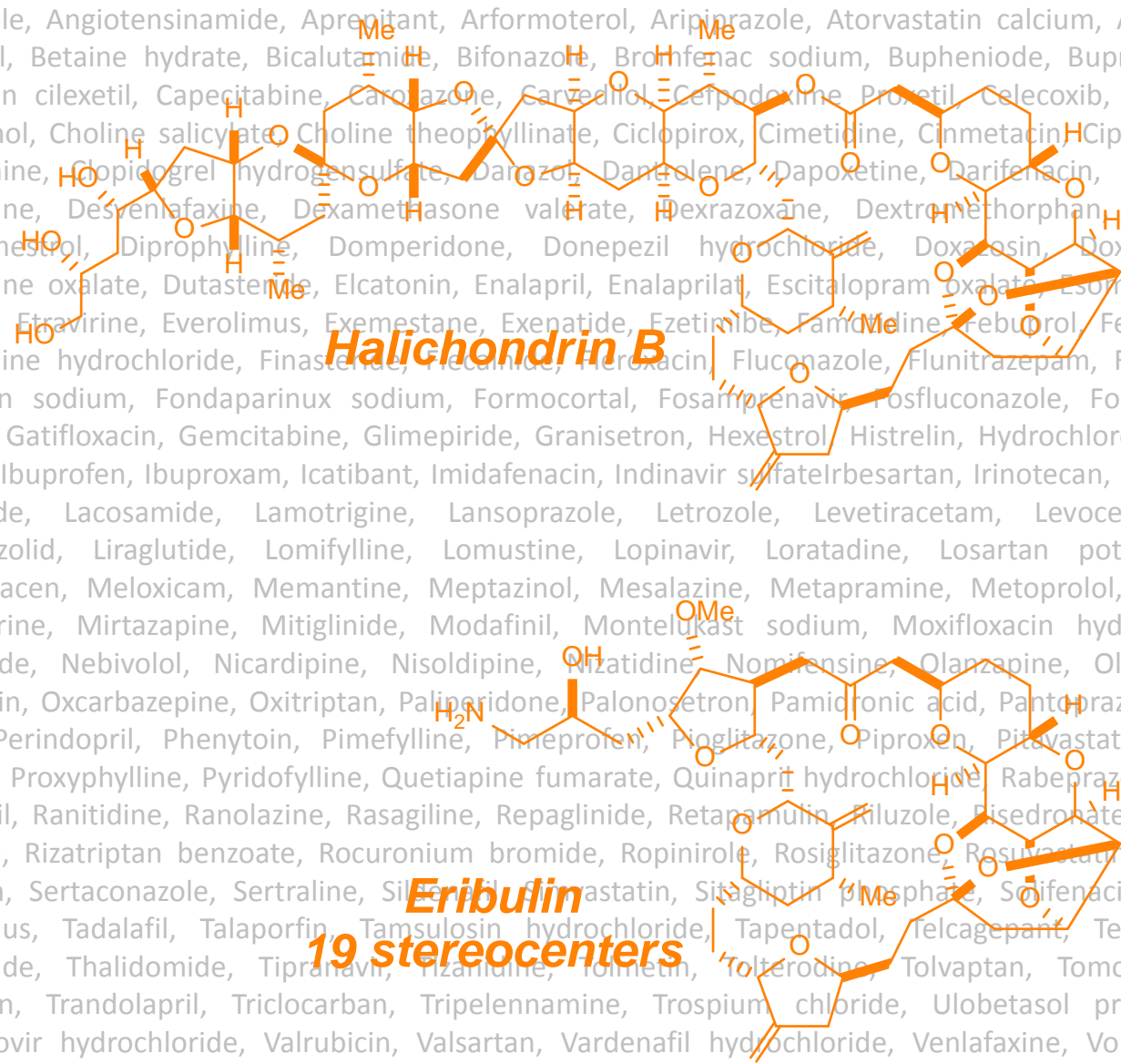
**Pharmaceutical**, medical and biotech



Cosmetics

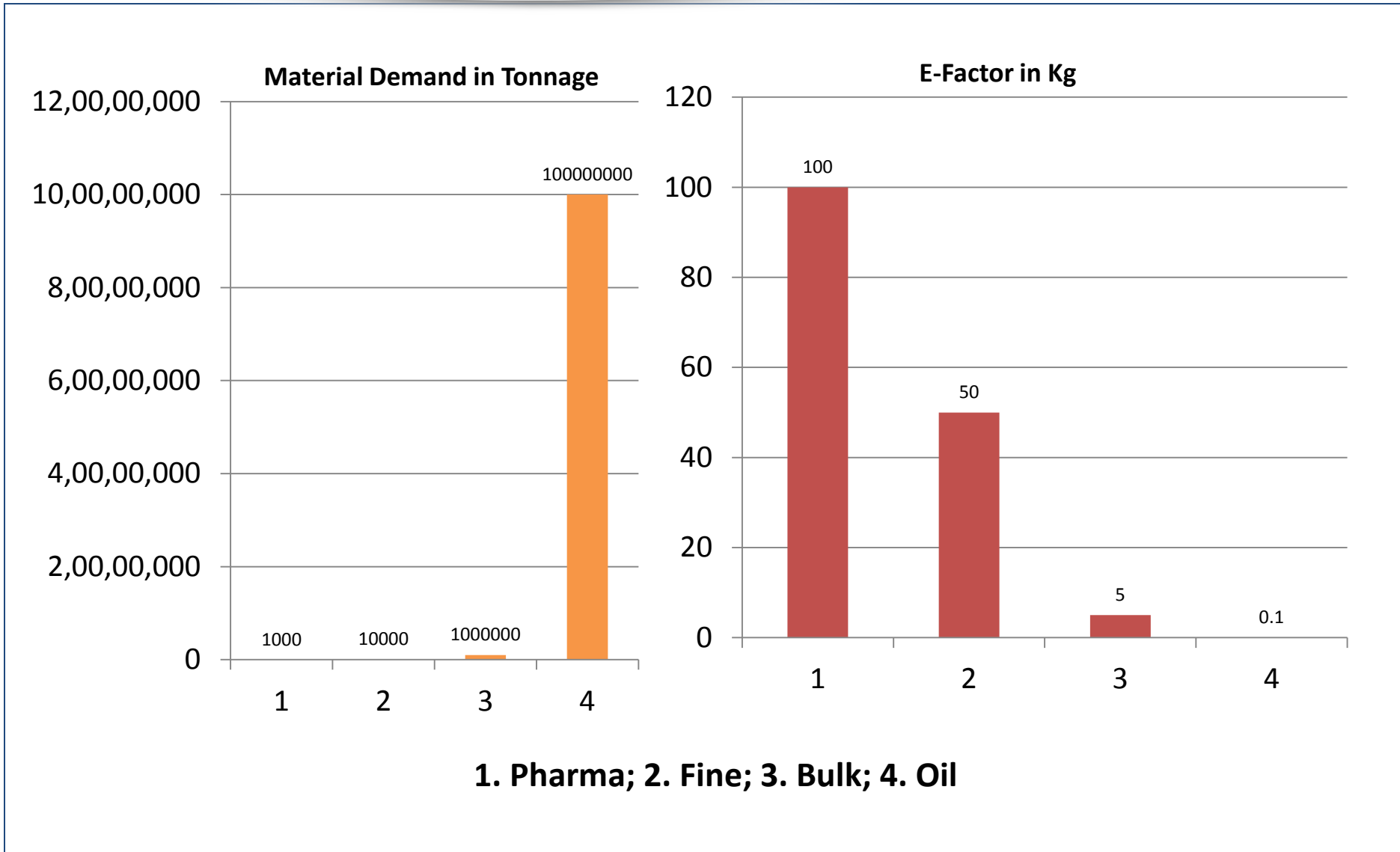
Retail and everyday commodity



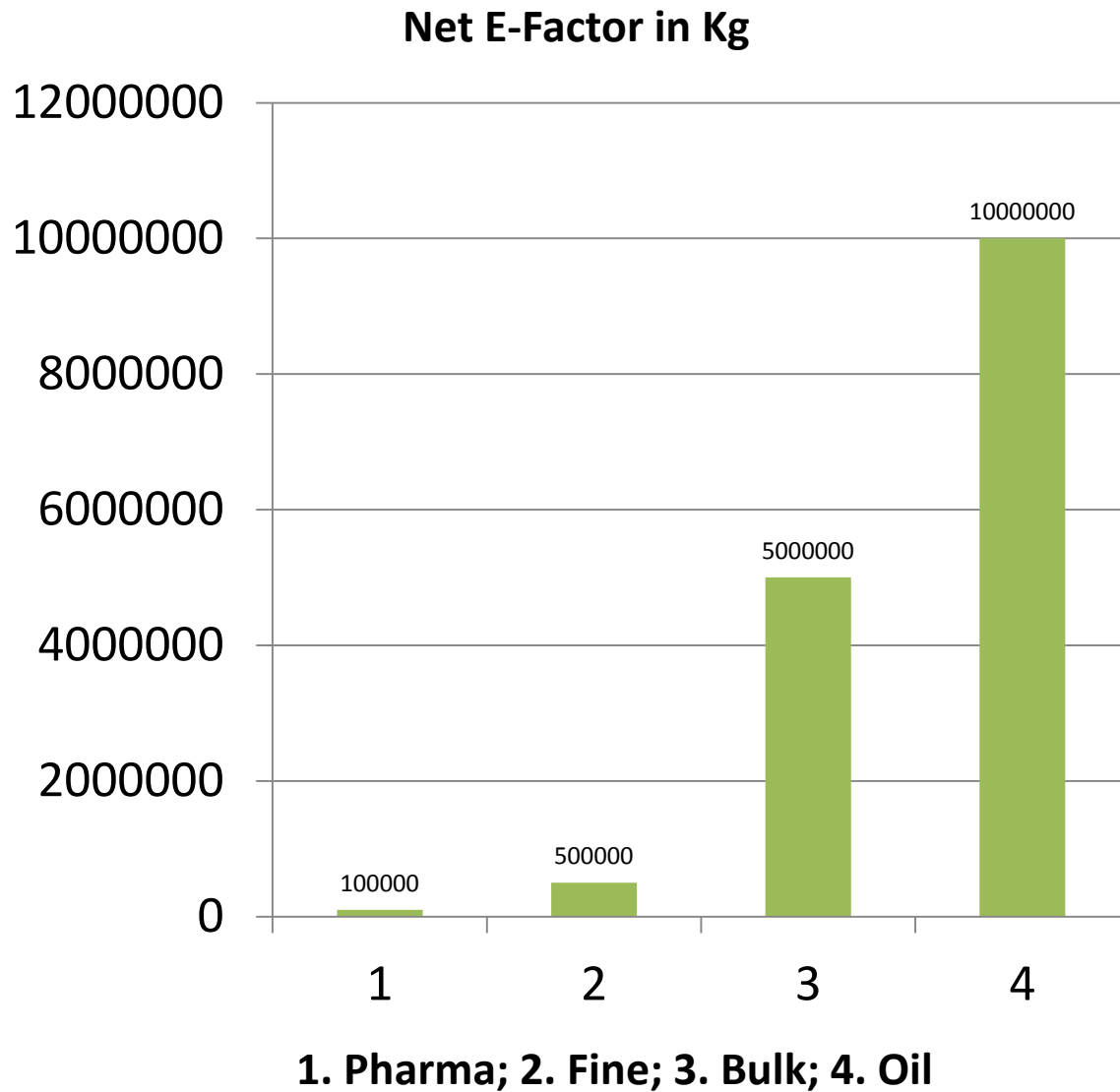


Abacavir, Acefylline, Acrivastine, Alendronate sodium, Alfuzosin, Almotriptan, Altretamine, Alvimopan, Amifostine, Amlodipine, Amprenavir, Anastrozole, Angiotensinamide, Aprepitant, Arformoterol, Aripiprazole, Atorvastatin calcium, Azacitidine, Azelastine, Bamifylline, Benazepril, Betaine hydrate, Bicalutamide, Bifonazole, Bromfenac sodium, Bupheniode, Buprenorphine, Cafedrine, Calcitonin, Candesartan cilexetil, Capecitabine, Capecitabine, Carvedilol, Ceriponidine, Cefepime, Celecoxib, Cetalkonium chloride, Cetirizine, Chlorobutanol, Choline salicylate, Choline theophyllinate, Ciclopirox, Cimetidine, Cinmetacin, Ciprofloxacin, Citalopram, Clomifene, Clomipramine, Clopidogrel hydrogensulfate, Darifenacin, Danrolene, Dapoxetine, Darifenacin, Darunavir, Decitabine, Degarelix, Desloratadine, Desvenlafaxine, Dexamethasone valerate, Dexrazoxane, Dextromethorphan, Dezocine, Diazepam, Diethylstilbestrol, Dimethyl, Diprophyllyne, Domperidone, Donepezil hydrochloride, Doxazosin, Doxofylline, Doxylamine, Dronedarone, Duloxetine oxalate, Dutasteride, Elcatonin, Enalapril, Enalaprilat, Escitalopram oxalate, Esomeprazole, Eszopiclone, Ethinamate, Etofylline, Etravirine, Everolimus, Exemestane, Exenatide, Ezetimibe, Famotidine, Fentanyl, Felodipine, Fenclofenac, Fenetylline, Fexofenadine hydrochloride, Finasteride, Flecainide, Fluoracinil, Fluconazole, Flunitrazepam, Fluoxetine, Fluticasone propionate, Fluvastatin sodium, Fondaparinux sodium, Formocortal, Fosamprenavir, Fosfluconazole, Fosinopril, Fosphenytoin sodium, Galantamine, Gatifloxacin, Gemcitabine, Glimepiride, Granisetron, Hexestrol, Histrelin, Hydrochlorothiazide, Ibandronate sodium monohydrate, Ibuprofen, Ibuprofen, Icatibant, Imidafenacin, Indinavir sulfate, Irbesartan, Irinotecan, Ixabepilone, Ketorolac, Lacidipine, Lacosamide, Lacosamide, Lamotrigine, Lansoprazole, Letrozole, Levetiracetam, Levocetirizine, Levofloxacin, Levosalbutamol, Linezolid, Liraglutide, Lomifylline, Lomustine, Lopinavir, Loratadine, Losartan potassium, Mabuprofen, Meclofenoxate, Melitracen, Meloxicam, Memantine, Meptazinol, Mesalazine, Metoprolol, Miglitol, Milnacipran hydrochloride, Minaprine, Mirtazapine, Mitiglinide, Modafinil, Montelukast sodium, Moxifloxacin hydrochloride, Naproxen, Naratriptan, Nateglinide, Nebivolol, Nicardipine, Nisoldipine, Nizatidine, Nomifensine, Olanzapine, Olmesartan medoxomil, Ondansetron, Oxaprozin, Oxcarbazepine, Oxitriptan, Paliperidone, Palonosetron, Pamidronic acid, Pantoprazole sodium, Parecoxib sodium, Paroxetine, Perindopril, Phenytoin, Pimefylline, Pirofenone, Pioglitazone, Piroxone, Pivastatin, Prednisval acetate, Pregabalin, Primidone, Proxyphylline, Pyridofylline, Quetiapine fumarate, Quinapril hydrochloride, Rabepazole sodium, Raloxifene hydrochloride, Ramipril, Ranitidine, Ranolazine, Rasagiline, Repaglinide, Retapamulin, Riluzole, Risedronate sodium, Risperidone, Ritonavir, Rivastigmine, Rizatriptan benzoate, Rocuronium bromide, Ropinirole, Rosiglitazone, Rosuvastatin calcium, Rupatadine, Salmeterol, Saxagliptin, Sertaconazole, Sertraline, Sildenafil citrate, Sitagliptin phosphate, Sofifenacin, Sulindac, Sulpiride, Sumatriptan, Tacrolimus, Tadalafil, Talaporfin, Tamsulosin hydrochloride, Tapentadol, Telcagepant, Telmisartan, Tenoxicam, Terbinafine, Teriparatide, Thalidomide, Tiplanavir, Tizanidine, Tolmetin, Tolterodine, Tolvaptan, Tomoxetine hydrochloride, Topiramate, Topotecan, Trandolapril, Triclocarban, Tripelennamine, Trosipium chloride, Ulobetasol propionate, Valaciclovir, Valdecoxib, Valganciclovir hydrochloride, Valrubicin, Valsartan, Vardenafil hydrochloride, Venlafaxine, Voriconazole, Zafirlukast, Zalcitabine, Zaleplon, Ziprasidone hydrochloride, Zoledronic acid, Zolmitriptan, Zolpidem, Zonisamide

# Higher E Factor $\alpha$ Degree of Complexity ?

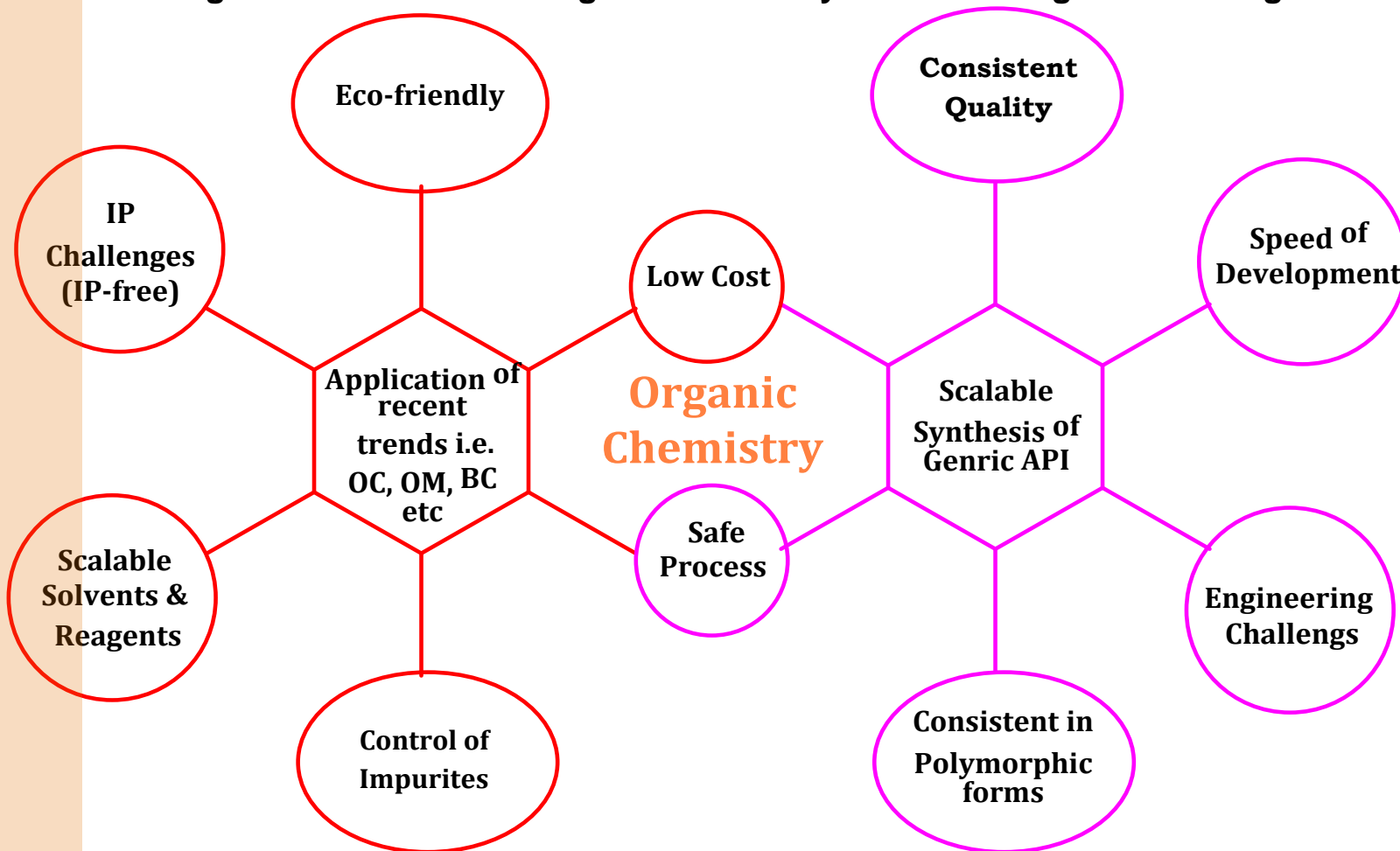


**1. Pharma; 2. Fine; 3. Bulk; 4. Oil**





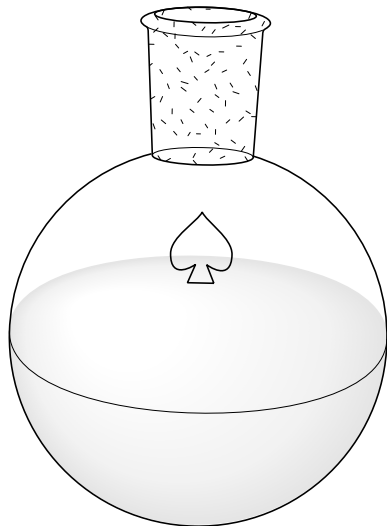
## Understanding the Pivotal Role of Organic Chemistry in Addressing the Challenges



1. Consideration of GC (TP and GM) in design phase
2. Minimize the number of steps while maintaining the desired cost component intact
3. Minimize or replace (Switch) non-green solvents
4. Work through multi-disciplinary scientific interface (Collaboration)
5. Renewable material based synthesis
6. Net output based energy efficient waste (unavoidable) management
7. Non-toxic and hazard free practices
8. Continuous mode of Chemistry/Engineering (flow technology)
9. In-expensive catalyst based transformations
10. Opt for asymmetric transformations
11. Use of immobilized recombinant enzymes for transformations with very low dilutions
12. Educate and prepare young generation considering intuitive knowledge potential to take a lead in this field

# Approaches: Flow Technology

## BATCH (space-resolved process)



- Conventional method
- Several Disadvantages
  - Time and labor intensive
  - A number of unit processes
  - Needs extensive optimization

## FLOW MICROREACTION TECHNOLOGY (time-resolved process)



- Emerging Technology
- Advantages over batch process
  - High surface area, precisely controlled conditions
  - Rapid screening of reaction conditions
  - “Scale-out” instead of “Scale-up”
  - Safety

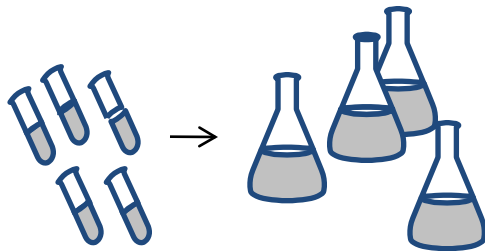
# Approaches: Biocatalysis

**Screening**

**Optimization**

**CPP**

**Production**



**Microbe, enzyme  
collection**

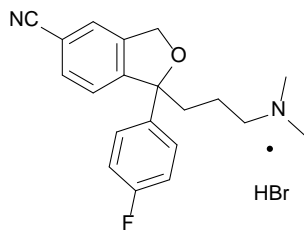
**Shake Flask**

**Lab Fermenter**

**Seed Fermenter**

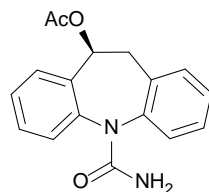
**Production  
Fermenter**

# Innovative Research Since 2007



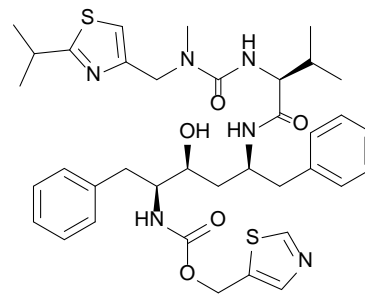
**Citalopram**

*Org. Proc. Res. & Dev.* **2013**, 17, 798-805



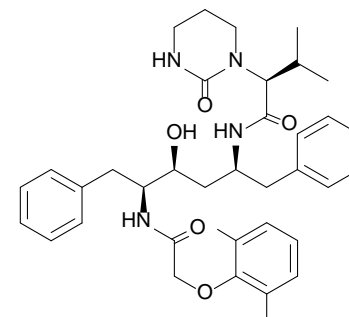
**Esclicarzepine**

*Catal. Sci. Technol.*, **2012**, 2, 1602-1605

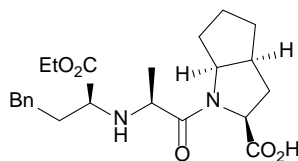


**Ritonavir**

*Tetrahedron Lett.* **2011**, 52, 6968-6970

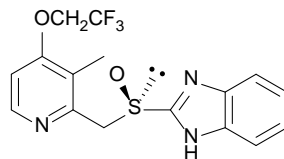


**Lopinavir**



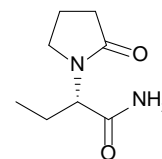
**Ramipril**

*Synthetic Commun.* **2011**, 41, 1186-1191



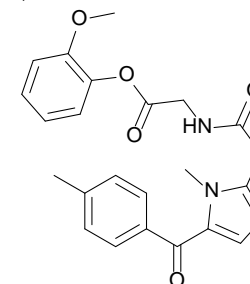
**Dexlansoprazole**

*Tetrahedron Lett.*, **2011**, 52, 5464-5468



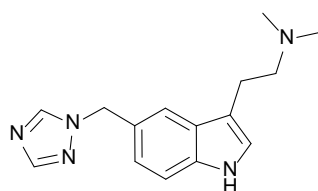
**Levetiracetam**

*Green Chem. Lett. Rev.* **2010**, 3, 225-230



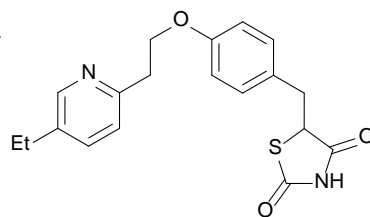
**Amtolmetin**

*Org. Proc. Res. & Dev.*, **2010**, 40, 362-368



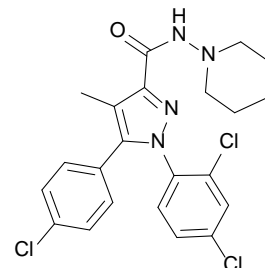
**Rizatriptan**

*Org. Proc. Res. & Dev.* **2009**, 13, 683-689



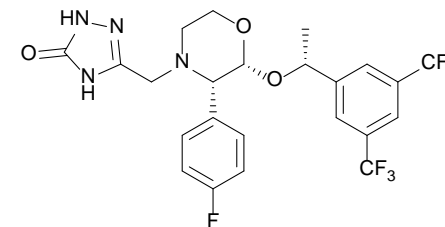
**Pioglitazone**

*Org. Proc. Res. & Dev.*, **2009**, 13, 1190-1194



**Rimonabant**

*Monatsh. Chem.*, **2008**, 139, 1091-1094



**Aprepitant**

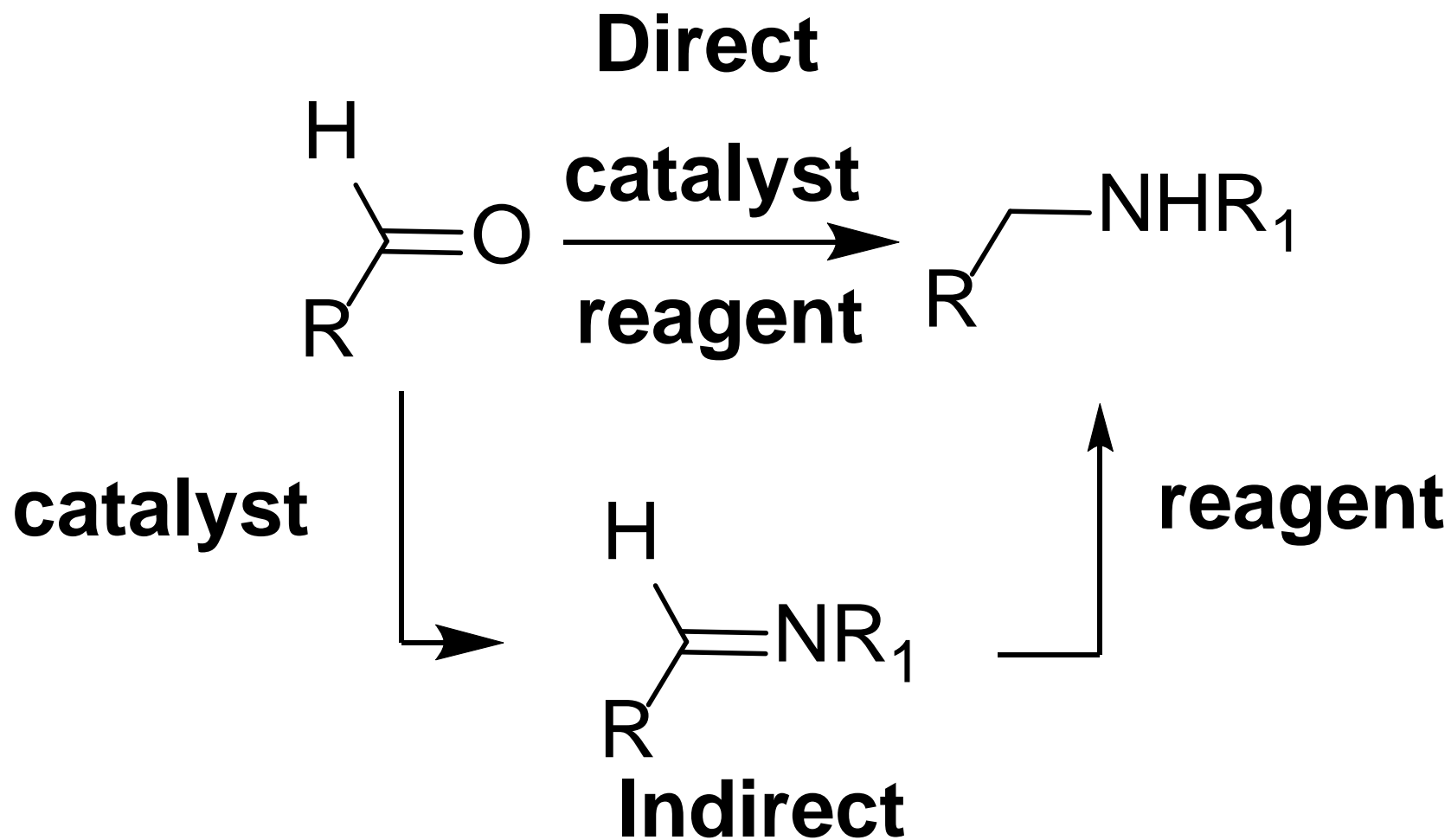
*Tetrahedron Lett.*, **2007**, 48, 8001-8004

## Types of Innovation

1. Incremental
2. Medium Size
3. Process
4. Technology based
  - a. Biocatalysis
  - b. Continuous
5. Major



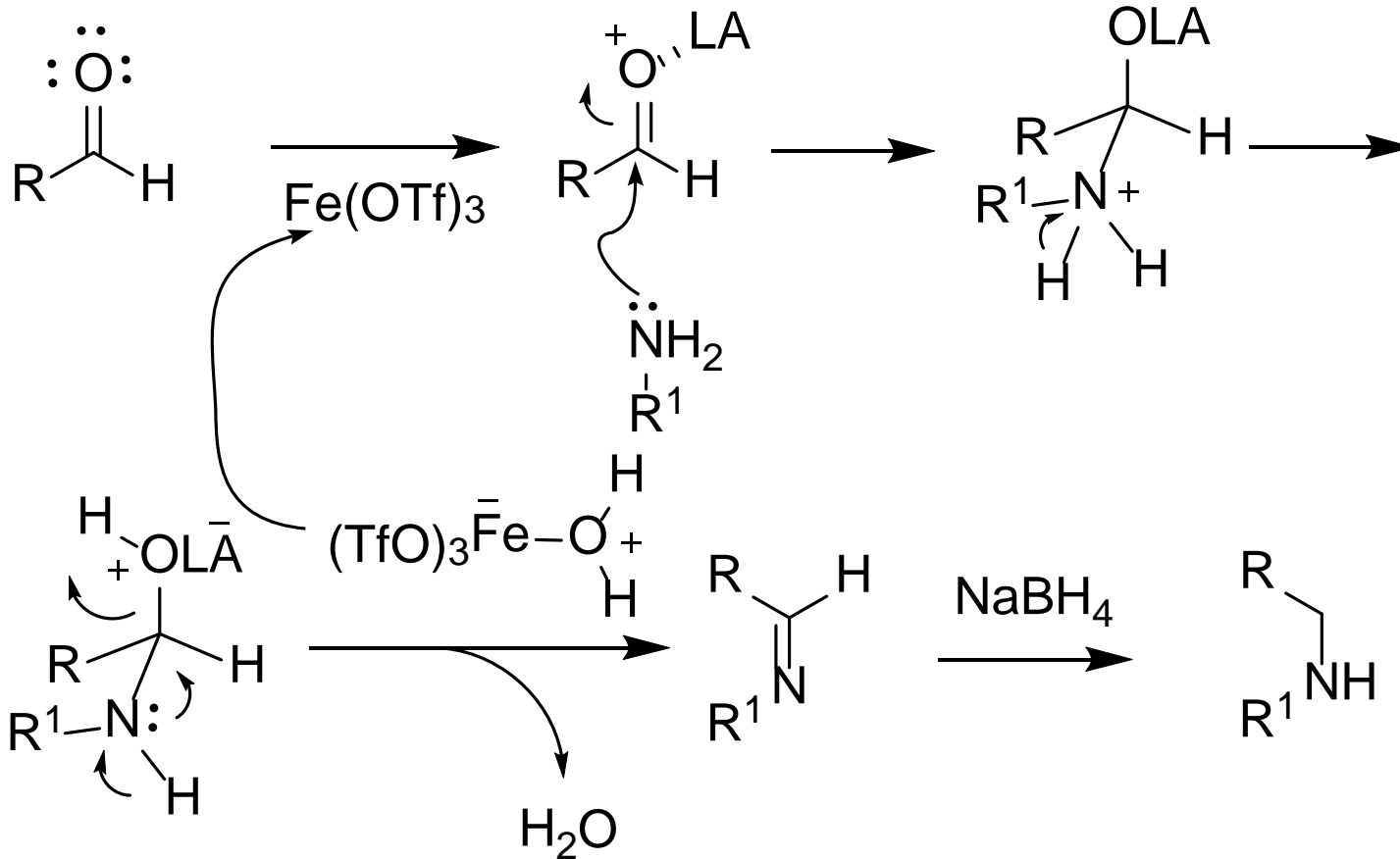
**BIG  
IMPACT**



Kumar, U. *Tetrahedron Lett.* **2012**, 53, 4354-4356

# Incremental: Reductive Amination

## Mechanistic Considerations



### Class 1 (toxic)

*Pt, Pd, Ir, Rh, Ru, Os  
Mo, Ni, Cr, V*

### Class 2 (less toxic)

*Cu, Mn, Ti, Sc*

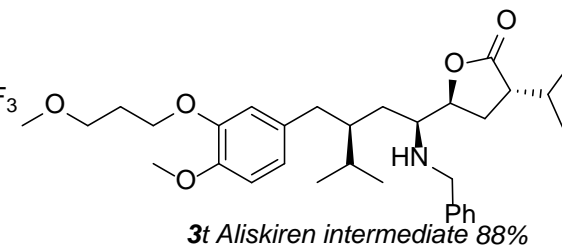
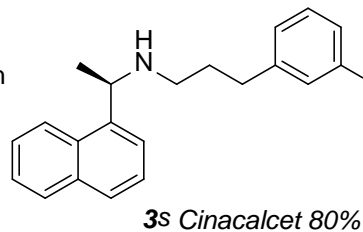
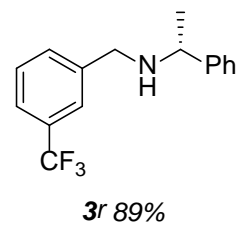
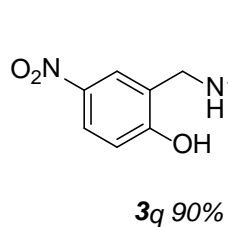
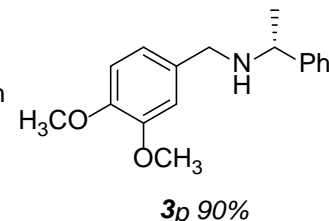
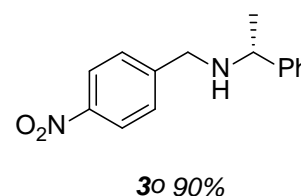
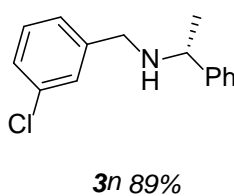
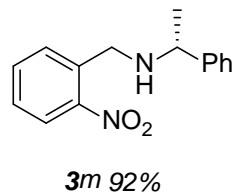
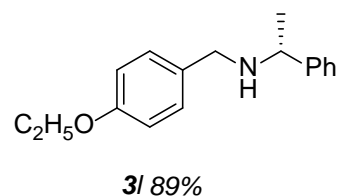
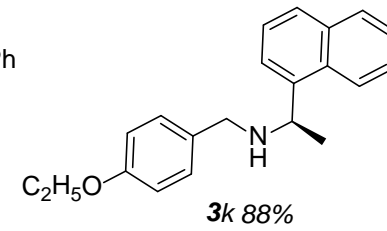
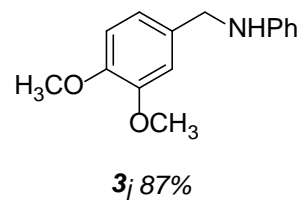
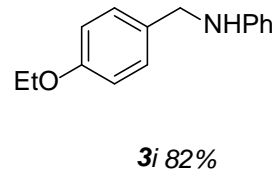
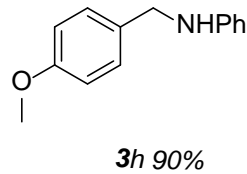
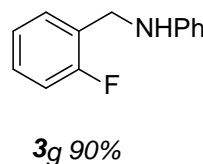
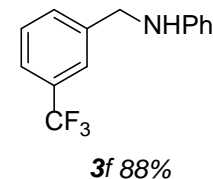
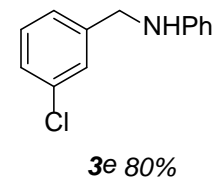
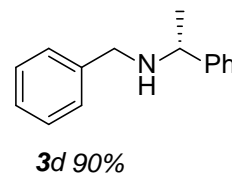
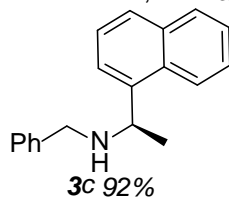
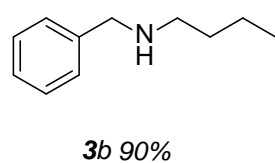
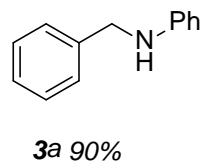
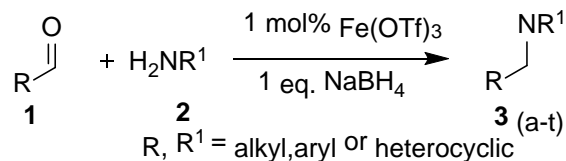
### Class 3 (non-toxic)

*Zn, Fe*

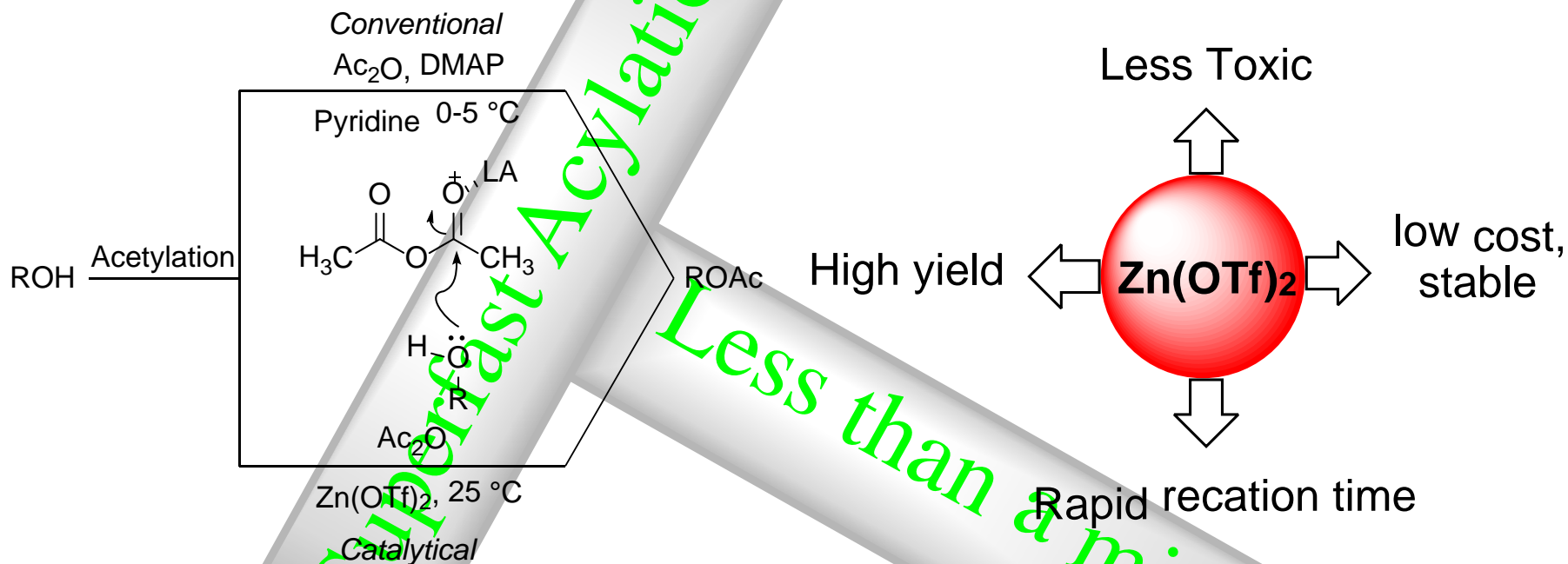


# Incremental: Reductive Amination

## Generality of the Method



# Incremental: Acylation



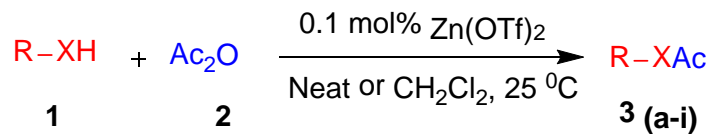
**Superfast Acylation**

**Less than a minute**

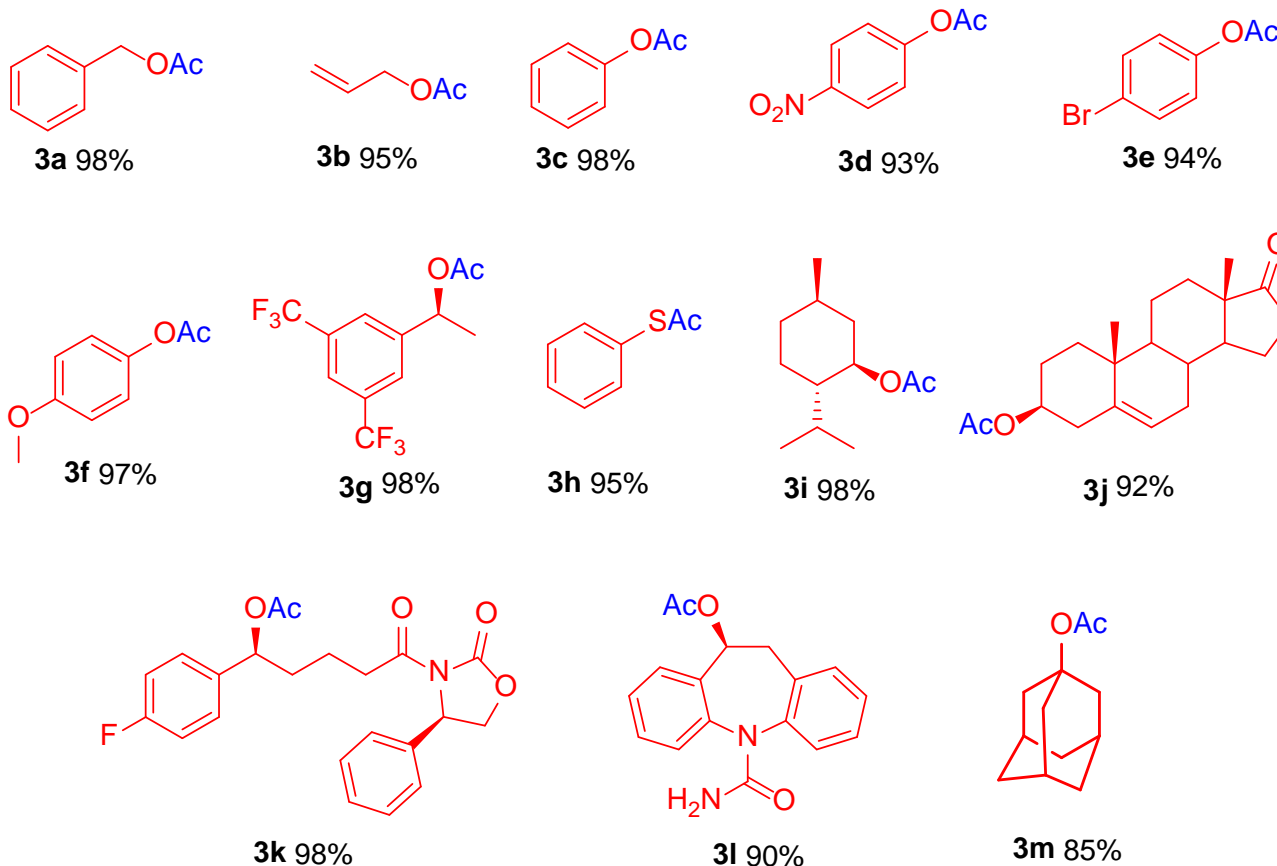
Kumar, U. *Tetrahedron Lett.* **2013**, in print

# Incremental: Acylation

## Generality of the Method



R = alkyl, benzyl, phenyl; X=O, S

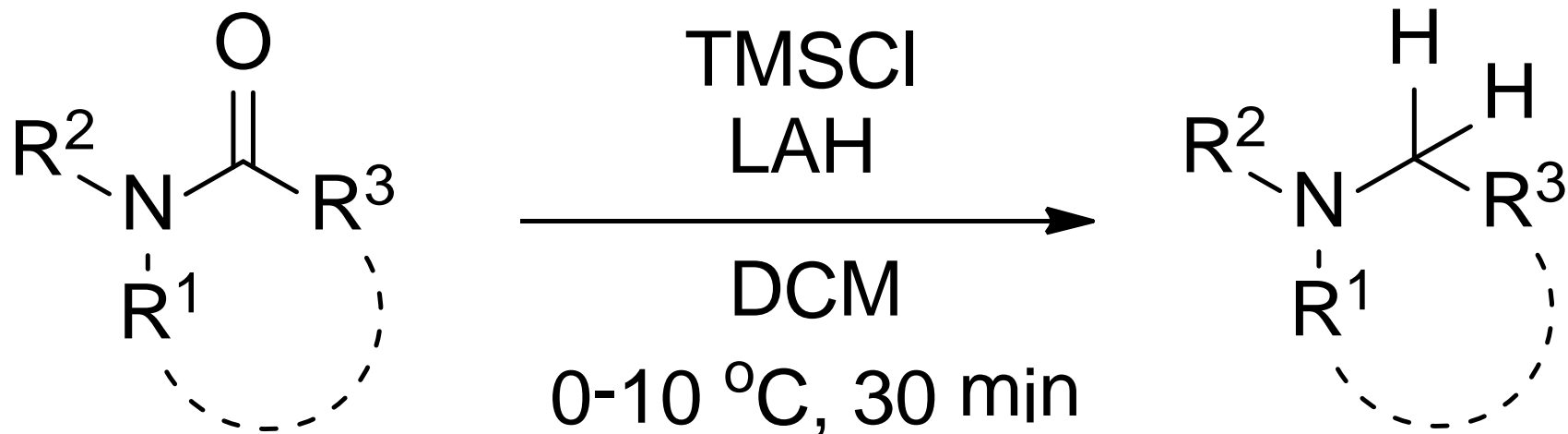


## Incremental: Acylation

### Pyridine (Py)/Ac<sub>2</sub>O Mediated vs Zn(OTf)<sub>2</sub> Catalyzed Acylation

Substrate	Reagent/catalyst	Conditions	Time	Yield (%)
<b>2l</b>	cat. DMAP, Py/Ac <sub>2</sub> O	CH <sub>2</sub> Cl <sub>2</sub> , Reflux	2 h	90
	Zn(OTf) <sub>2</sub> (0.1 mol%) /Ac <sub>2</sub> O	25-25 °C	60s	90
<b>2j</b>	Py/Ac <sub>2</sub> O	25-25 °C	3 h	87
	Zn(OTf) <sub>2</sub> (0.1 mol%) /Ac <sub>2</sub> O	25-25 °C	30s	92

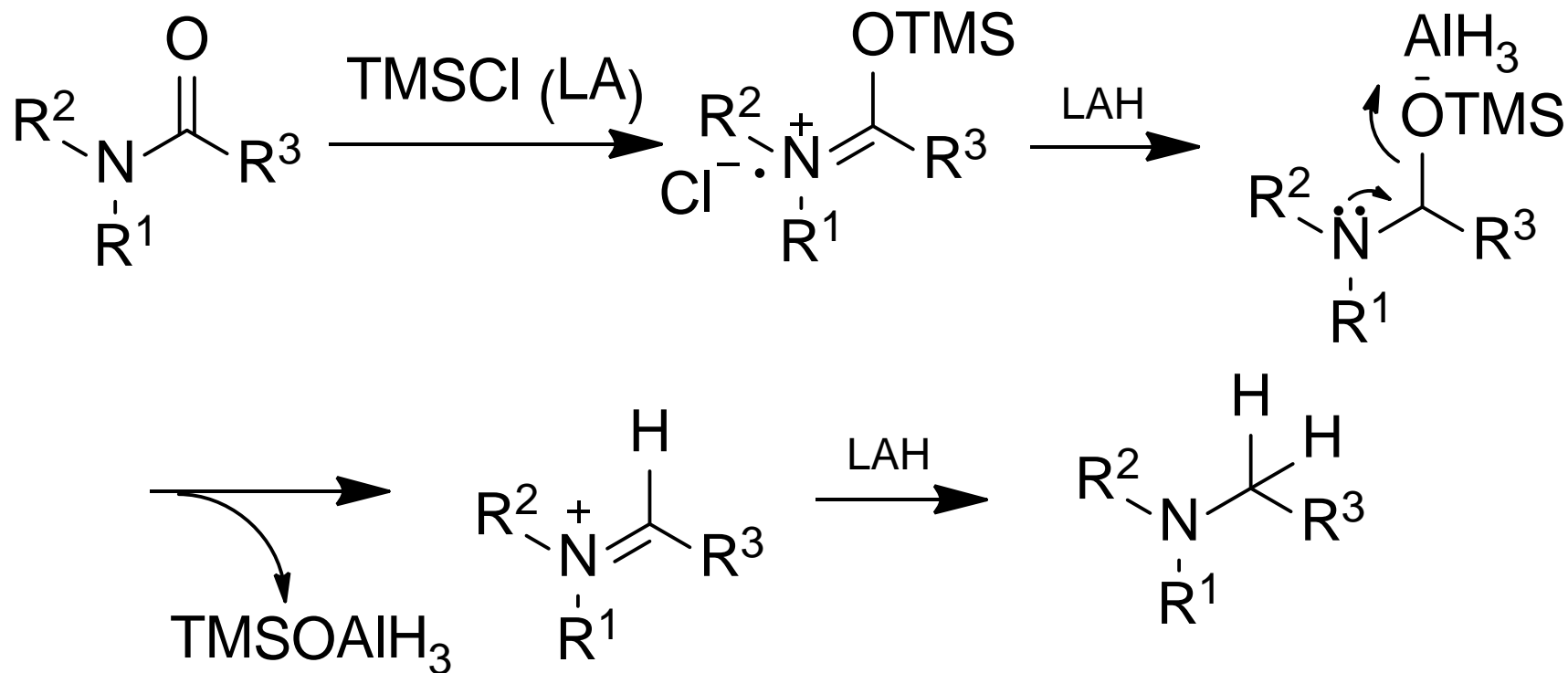
# Incremental: Amide Reduction



*Tetrahedron Lett.* **2013**, 54, 4908-4913. (Featured in R&D Highlights Published in *Org.Proc.Res.&Dev.* **2013** and being tracked in Article Usage Dashboard)

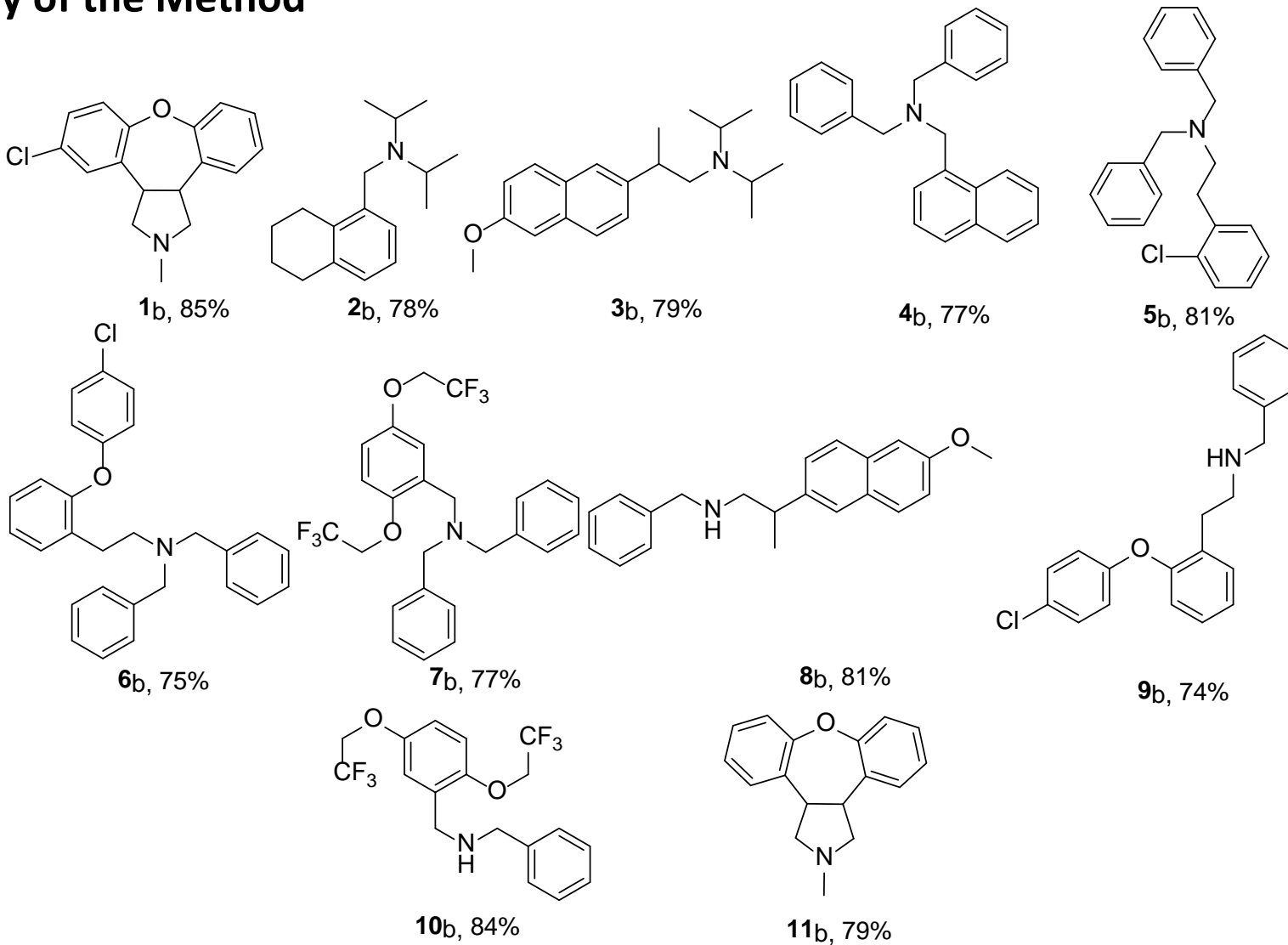
# Incremental: Amide Reduction

## Mechanistic Considerations

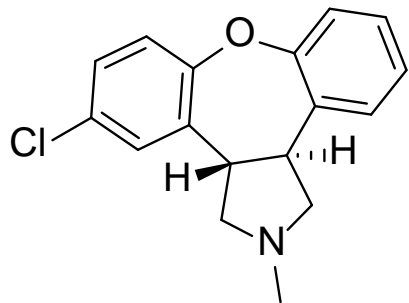


# Incremental: Amide Reduction

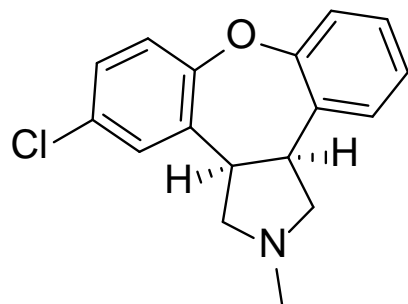
## Generality of the Method



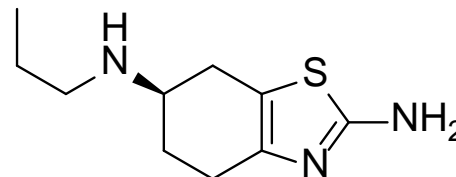
## Generality of the Method



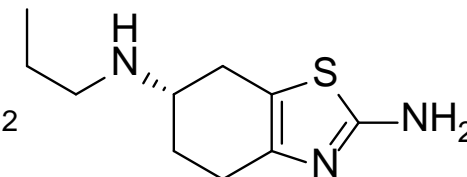
1b', 85%, 99% *de*



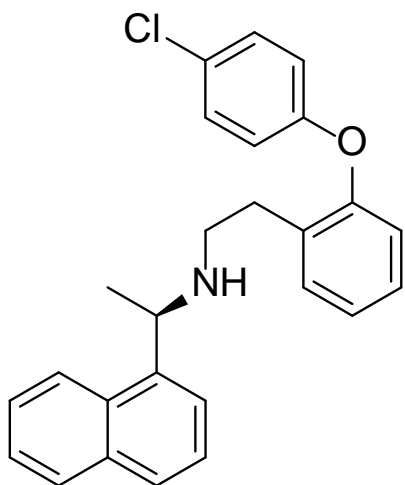
1b'', 79%, 99% *de*



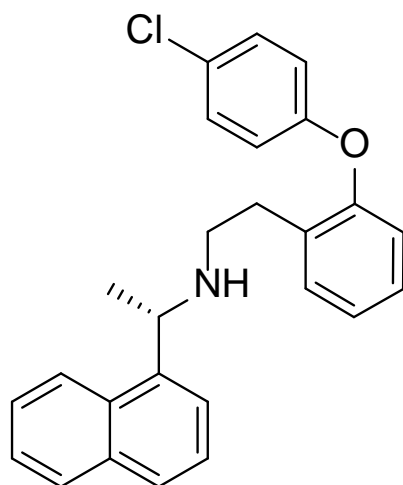
12b, 77%, 98.5% *ee*



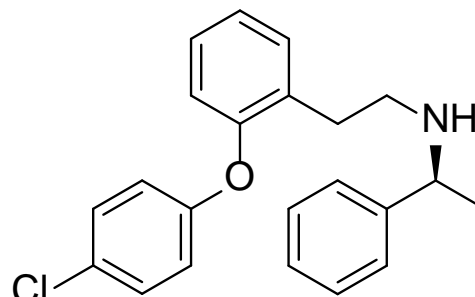
13b, 72%, 98.5% *ee*



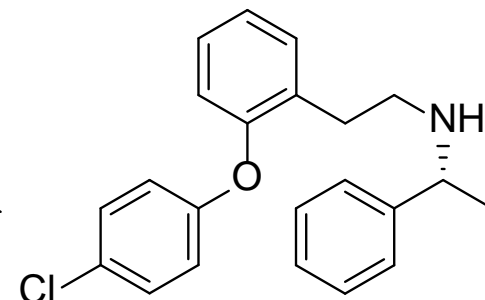
14b, 81%, 99% *ee*



15b, 79%, 99% *ee*



16b, 82%, 98.5% *ee*



17b, 79%, 98.5% *ee*



## Article Usage Dashboard



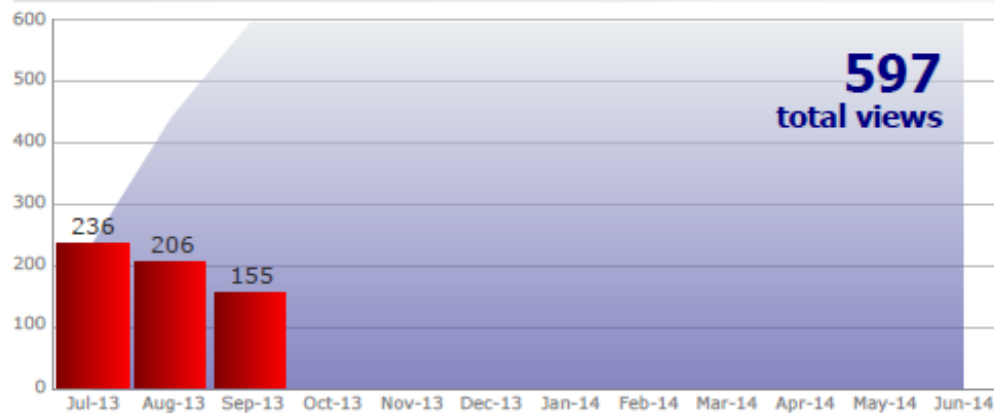
### Amide activation by TMSCl: reduction of amides to amines by LiAlH<sub>4</sub> under mild conditions

Ravinder, B.; Rajeswar Reddy, S.; Panasa Reddy, A.; Bandichhor, R.

Tetrahedron Letters, Volume(s) 54, 04-Jul-2013, Pages 4908-4913

[View Article](#)

### Trend and cumulative views



### Views by geography

Top countries	Rank	Views	Pct
United States	1	132	22%
Japan	2	77	13%
India	3	76	13%
Australia	4	48	8%
China	5	40	7%

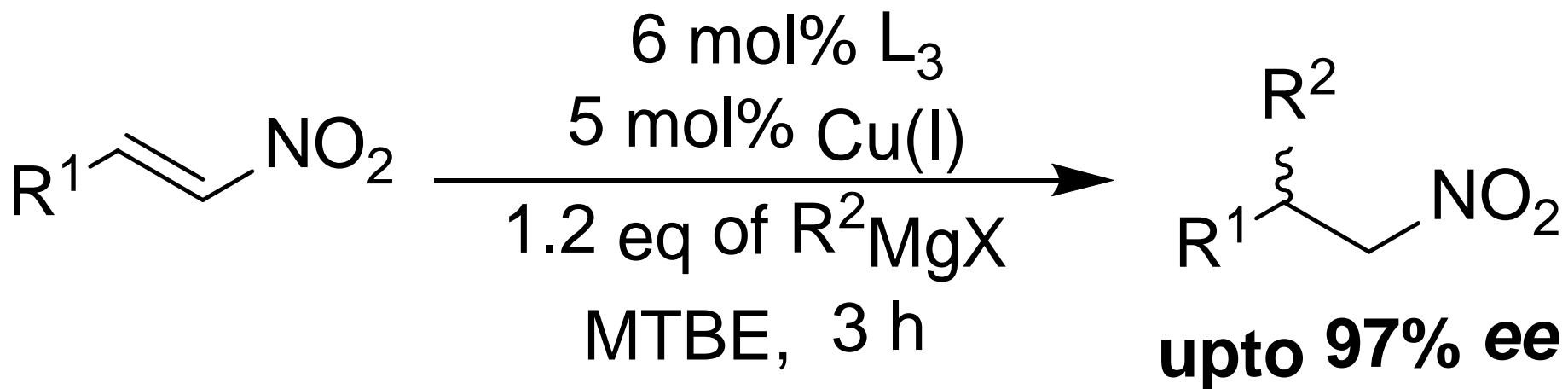
### Corporate versus Public Sector



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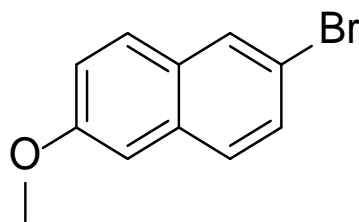
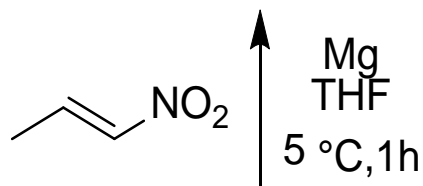
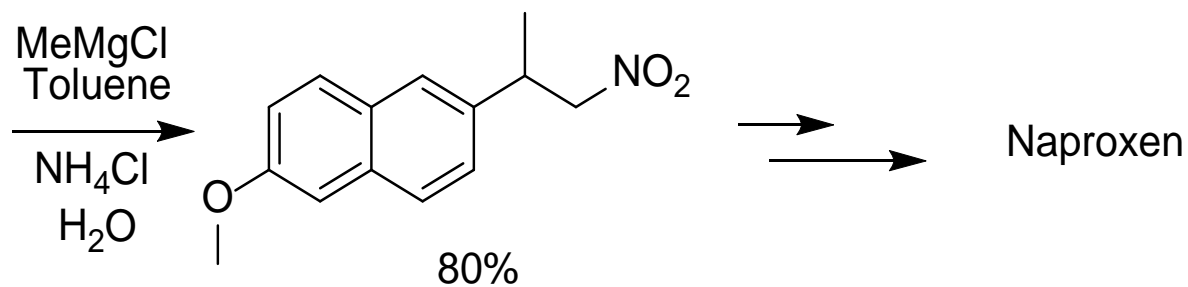
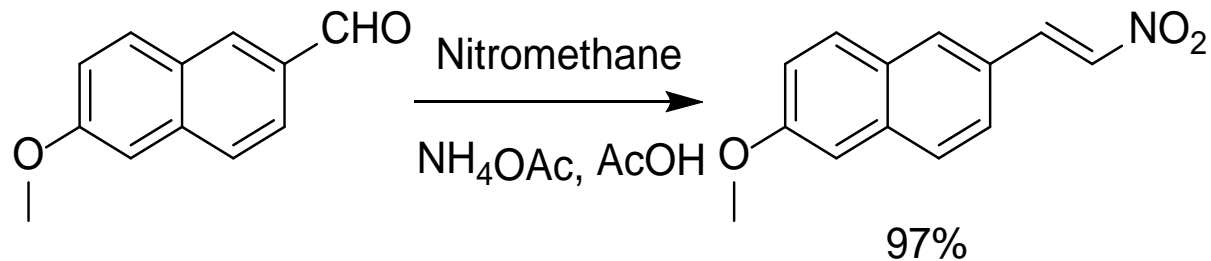
[Support](#)



Prashanth Reddy, G. *Tetrahedron Lett.*, **2013**, 54, 3911-3915

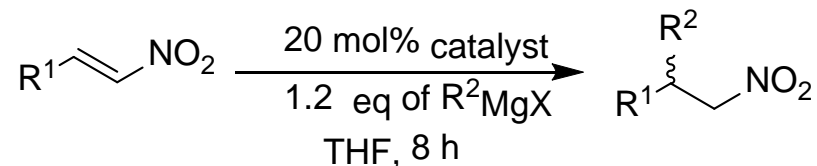
# Medium Size: Enantioselective Grignard Addition to Nitroolefin

## Potential Application

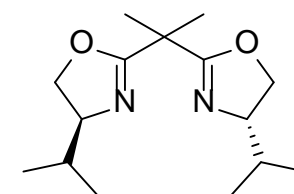


Naproxen

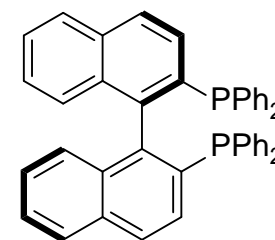
## Screening



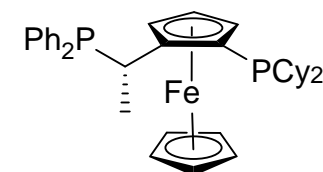
S.No.	Catalyst	R <sup>1</sup> /R <sup>2</sup> /X	°C	<i>er</i> (R/S) (HPLC)	Yield (%)
1	CuTC/L <sub>1</sub>	6Mn/Me/Br	-60	49.5/50.5	75
2	CuTC/L <sub>1</sub>	6Mn/Me/Br	-20	49/51	75
3	CuI/L <sub>1</sub>	6Mn/Me/Br	-60	45/55	70
4	CuI/L <sub>1</sub>	6Mn/Me/Br	-20	48.6/51.4	70
5	Zn(OTf) <sub>2</sub> /L <sub>1</sub>	6Mn/Me/Br	-60	49.5/50.5	70
6	Zn(OTf) <sub>2</sub> /L <sub>1</sub>	6Mn/Me/Br	-20	49.5/50.5	70
7	CuTC/L <sub>2</sub>	6Mn/Me/Br	-60	49/51	75
8	CuTC/L <sub>2</sub>	6Mn/Me/Br	-20	49.5/50.5	75
9	CuI/L <sub>2</sub>	6Mn/Me/Br	-60	49.3/50.7	70
10	CuI/L <sub>2</sub>	6Mn/Me/Br	-20	49.7/50.3	70
11	Zn(OTf) <sub>2</sub> /L <sub>2</sub>	6Mn/Me/Br	-60	49.2/50.8	70
12	Zn(OTf) <sub>2</sub> /L <sub>2</sub>	6Mn/Me/Br	-20	49.6/50.4	70
13	CuTC/L <sub>1</sub>	6Mn/Me/Cl	-70	49.2/50.8	70
14	CuTC/L <sub>1</sub>	6Mn/Me/Cl	-20	49.3/50.7	70
15	CuTC/L <sub>1</sub>	6Mn/Me/Cl	-35	48/52	72



L<sub>1</sub>: (SS)-isopropyl bisoxazoline

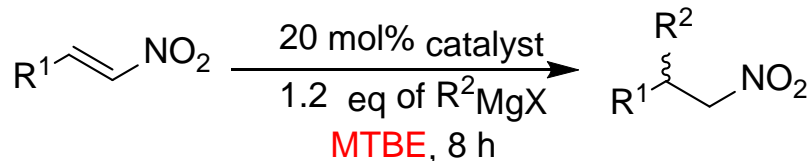


L<sub>2</sub>: (R)-BINAP

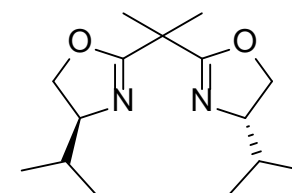


L<sub>3</sub>: [(S)-1-[(Rp)-2-(Dicyclohexylphosphino)-ferrocenylethyl]diphenylphosphine]

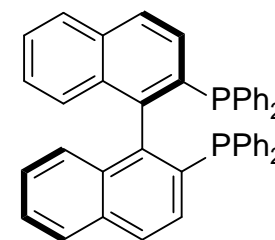
## Screening



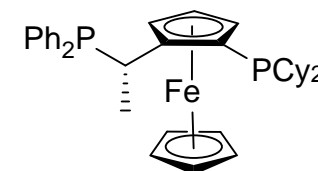
S.No.	Catalyst	R <sup>1</sup> /R <sup>2</sup> /X	°C	<i>er</i> (R/S) (HPLC)	Yield %
1	CuTC/L <sub>1</sub>	6Mn/Me/Cl	-40	48.7/51.3	70
2	CuTC/L <sub>1</sub>	6Mn/Et/Cl	-40	49/51	69
3	CuTC/L <sub>1</sub>	6Mn/ <i>i</i> Pr/Cl	-40	51/49	67
4	<b>CuTC/L<sub>1</sub></b>	<b>6Mn/<i>t</i>Bu/Cl</b>	<b>-40</b>	<b>72/28</b>	<b>68</b>
5	<b>CuTC/L<sub>1</sub></b>	<b>6Mn/<i>t</i>Bu/Cl</b>	<b>-70</b>	<b>77/23</b>	<b>69</b>



L<sub>1</sub>: (SS)-isopropyl bisoxazoline

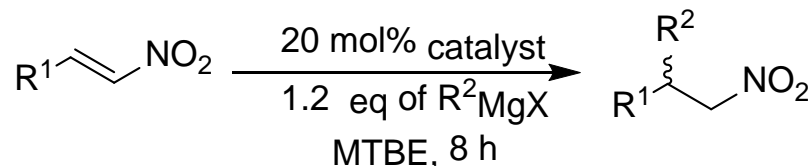


L<sub>2</sub>: (R)-BINAP

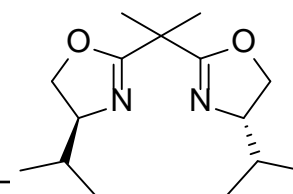


L<sub>3</sub>: [(S)-1-[(Rp)-2-(Dicyclohexylphosphino)-ferrocenylethyl]diphenylphosphine]

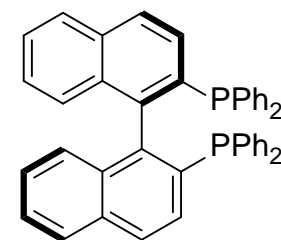
## Screening



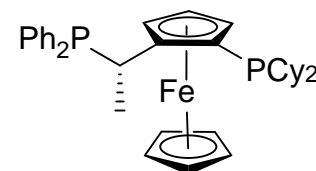
S.No.	Catalyst	R <sup>1</sup> /R <sup>2</sup> /X	°C	<i>er</i> (R/S) (HPLC)	Yield %
1	CuI/L <sub>3</sub>	6Mn/Me/Cl	-70	49.2/50.8	68
2	CuI/L <sub>3</sub>	6Mn/Et/Cl	-70	49.5/50.5	64
3	CuI/L <sub>3</sub>	6Mn/ <i>i</i> Pr/Cl	-70	51.97/48.03	64
4	CuTC/L <sub>3</sub>	6Mn/ <sup>t</sup> Bu/Cl	-70	61/39	62
5	<b>CuI/L<sub>3</sub></b>	<b>6Mn/<sup>t</sup>Bu/Cl</b>	<b>-70</b>	<b>98.5/1.5</b>	<b>62</b>
6	<b>CuI/L<sub>3</sub></b>	<b>6Mn/<sup>t</sup>Bu/Cl</b>	<b>-70</b>	<b>97.9/2.1</b>	<b>64</b>
7	<b>CuI/L<sub>3</sub></b>	<b>6Mn/<sup>t</sup>Bu/Cl</b>	<b>-70</b>	<b>97.9/2.1</b>	<b>65</b>
8	<b>CuI/L<sub>3</sub></b>	<b>6Mn/<sup>t</sup>Bu/Cl</b>	<b>-70</b>	<b>97.8/2.2</b>	<b>63</b>
9	CuI/L <sub>3</sub>	<i>i</i> Bu/6Mn/Br	-70	25/75	75
10	CuTC/L <sub>3</sub>	6Mn/Ph/Cl	-70	50.56/49.44	62
11	CuTC/L <sub>3</sub>	6Mn/Benzyl/Cl	-70	49.49/50.51	66
12	CuI/L <sub>3</sub>	Ph/ <sup>t</sup> Bu/Cl	-70	39.5/60.5	60
13	CuI/L <sub>3</sub>	<i>p</i> -EtO-Ph/ <sup>t</sup> Bu/Cl	-70	23/77	65
14	<b>CuI/L<sub>3</sub></b>	<b><i>p</i>-F-Ph/<sup>t</sup>Bu/Cl</b>	<b>-70</b>	<b>4/96</b>	<b>60</b>



L<sub>1</sub>: (SS)-isopropyl bisoxazoline



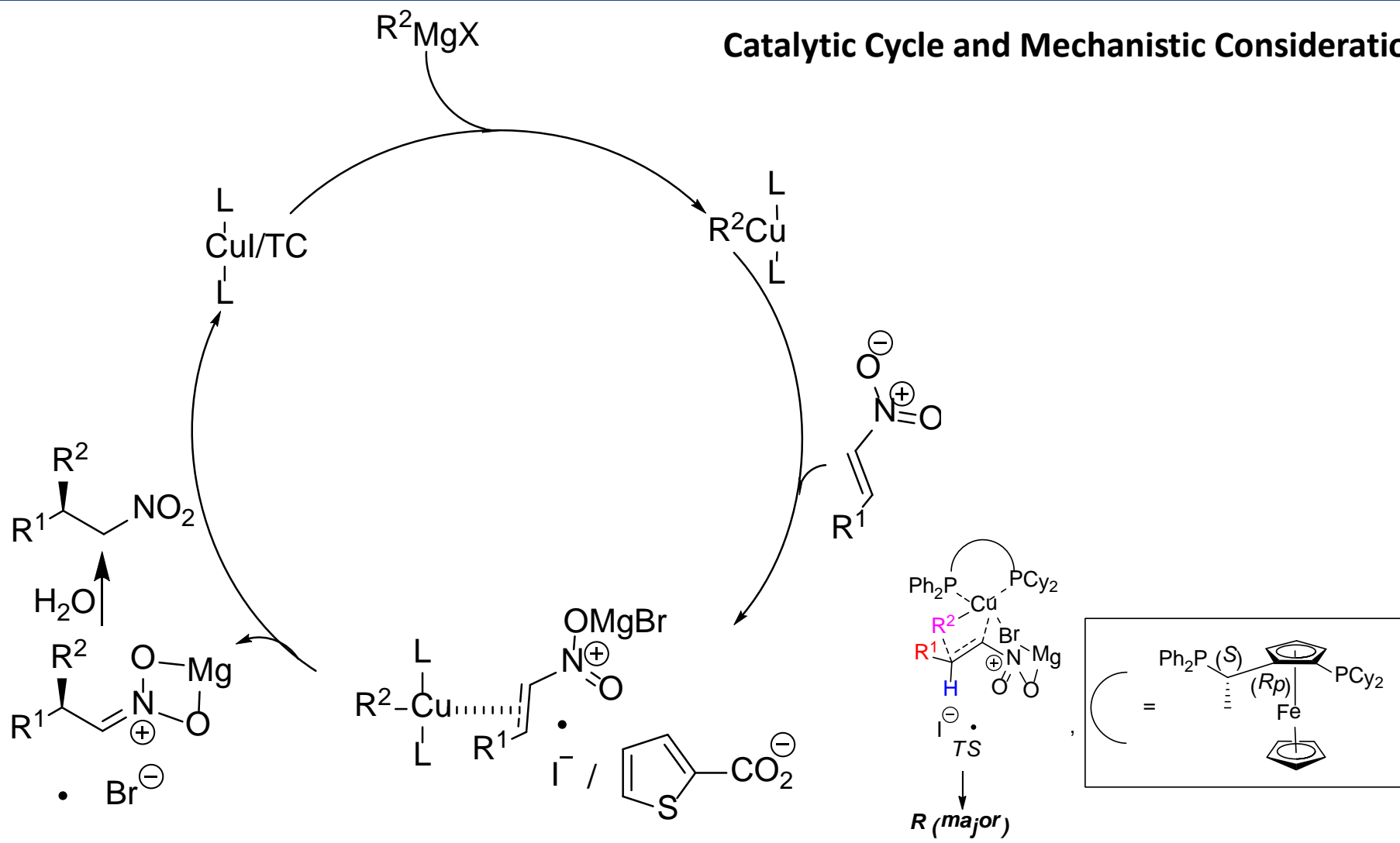
L<sub>2</sub>: (R)-BINAP



L<sub>3</sub>: [(S)-1-[(Rp)-2-(Dicyclohexylphosphino)ferrocenylethyl]diphenylphosphine]

# Medium Size: Enantioselective Grignard Addition to Nitroolefin

## Catalytic Cycle and Mechanistic Consideration



## Article Usage Dashboard



### Enantioselective Grignard addition to nitroolefin

Reddy, P.; Bandichhor, R.

Tetrahedron Letters, Volume(s) 54, 10-May-2013, Pages 3911-3915

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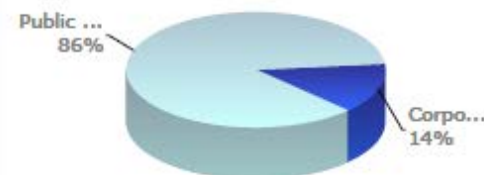
### Views by geography

Top countries	Rank	Views	Pct
India	1	148	21%
China	2	126	18%
United States	3	109	16%
Japan	4	59	8%
Spain	5	20	3%

### Trend and cumulative views



### Corporate versus Public Sector

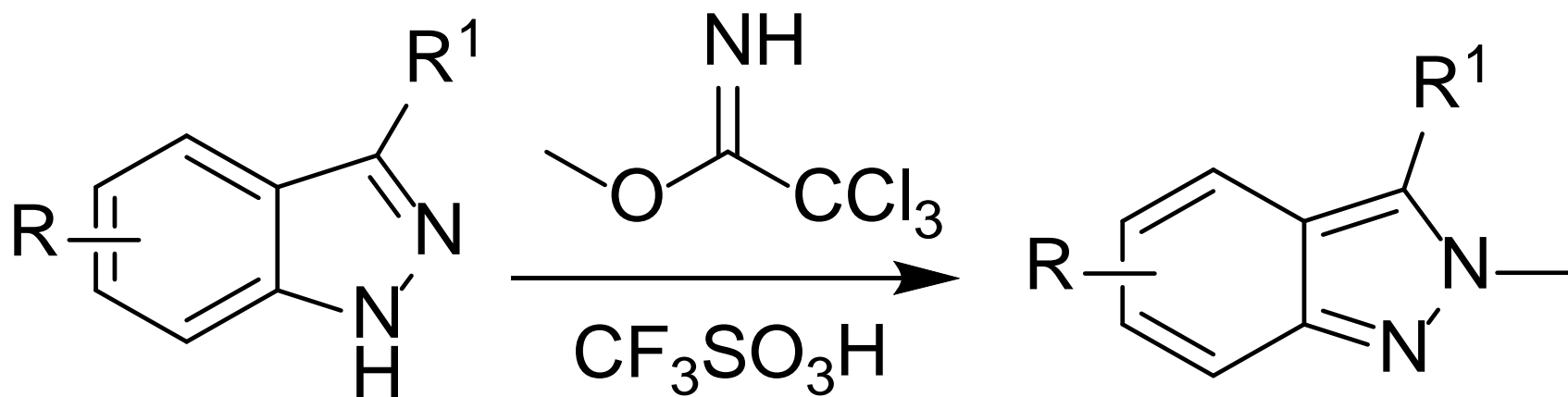


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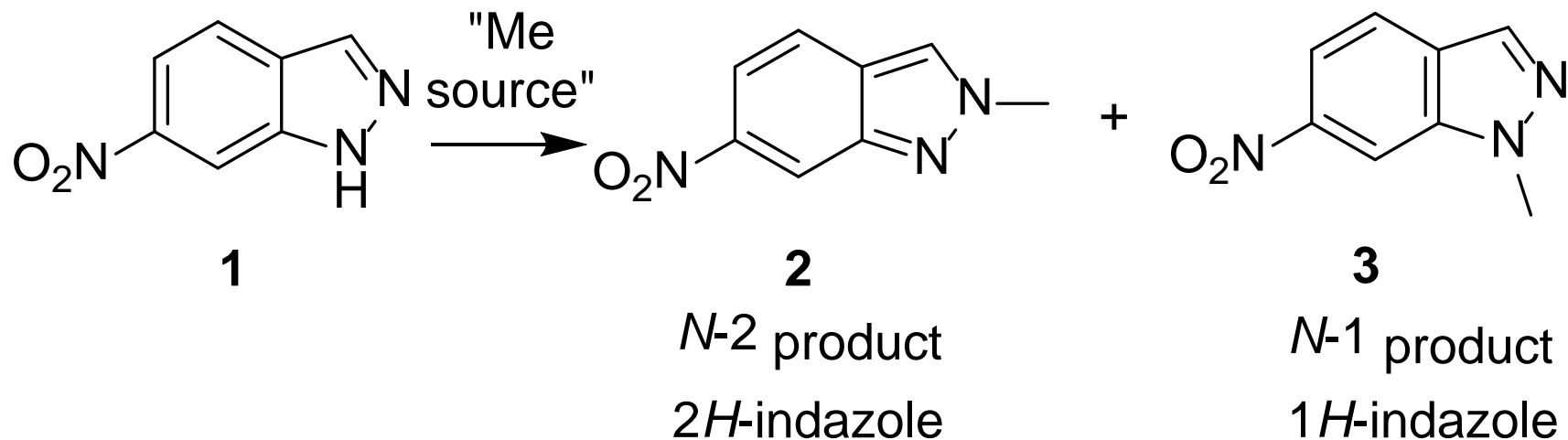
[Support](#)





Sudhakar, *Tetrahedron Lett.*, **2013**, 54, 1661-1663

## Different Methods

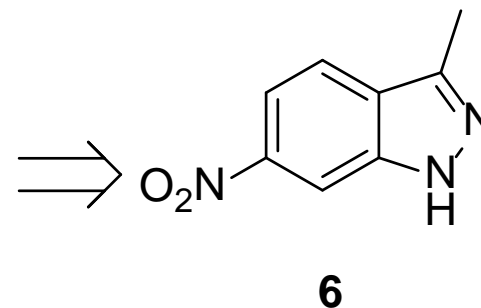
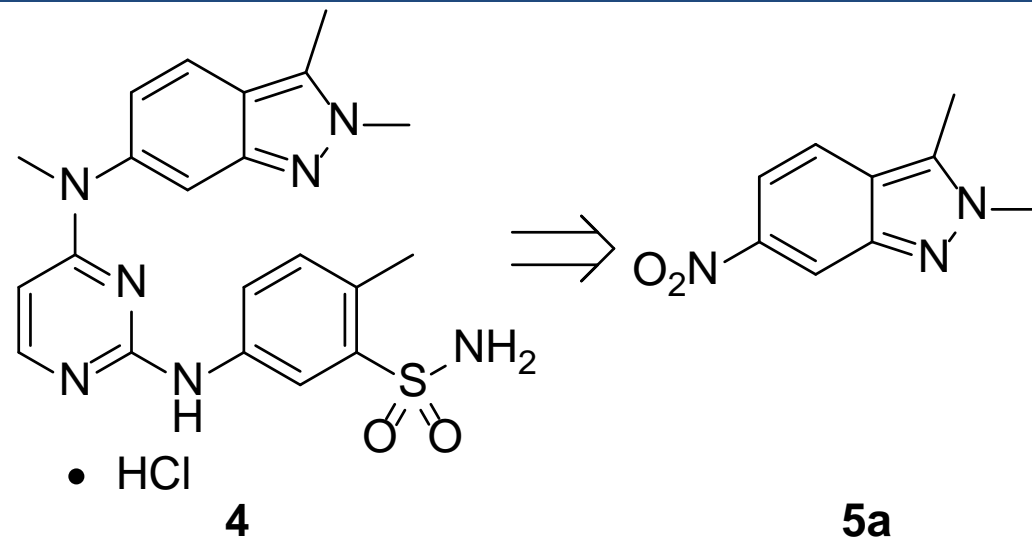
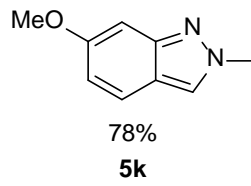
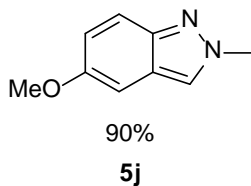
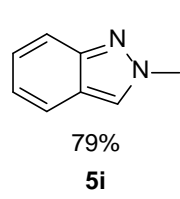
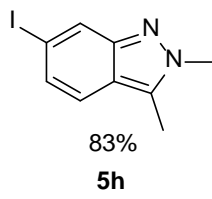
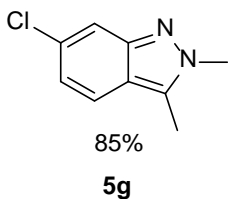
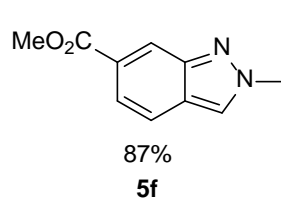
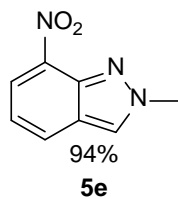
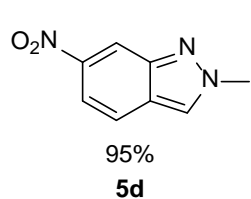
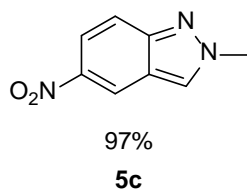
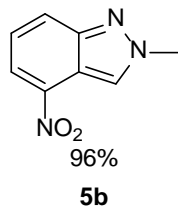
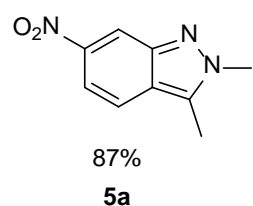
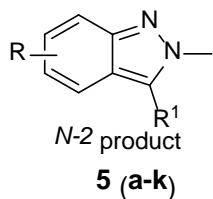
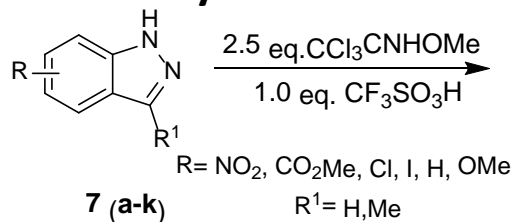


*Basic condition: No selectivity*

*Acidic condition: N-1 (thermodynamic); N-2 (kinetic)*

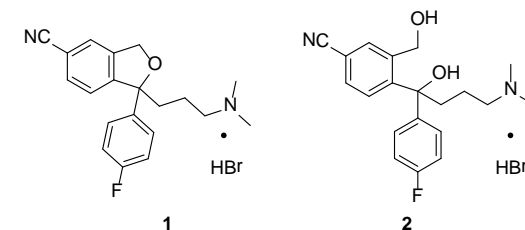
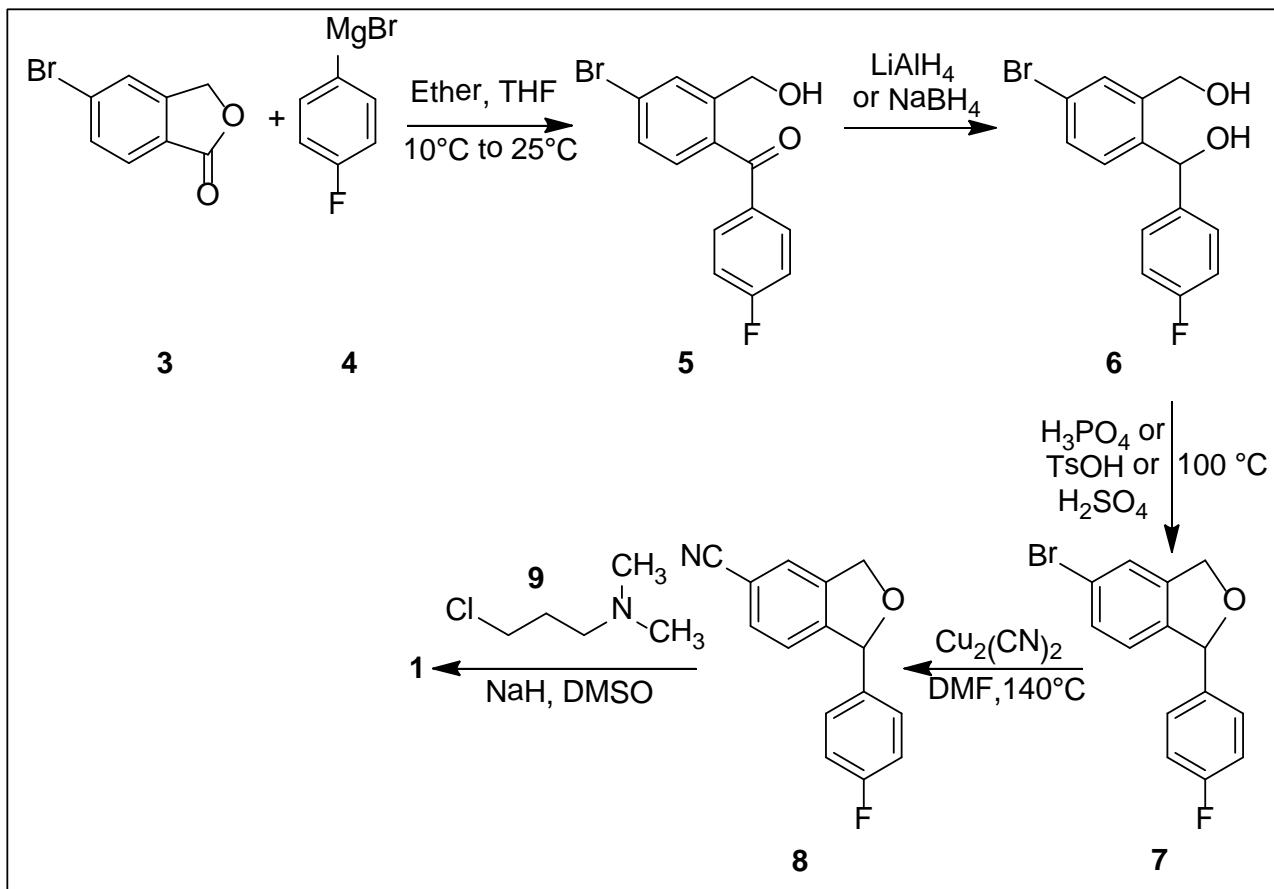
*Trimethyloxonium tetrafluoroborate (Meerwein's reagent) is considered to be the best reagent for regioselective methylation*

## Generality of the Methods



*This reagent is suitable to both EWG and EDG containing substrates*

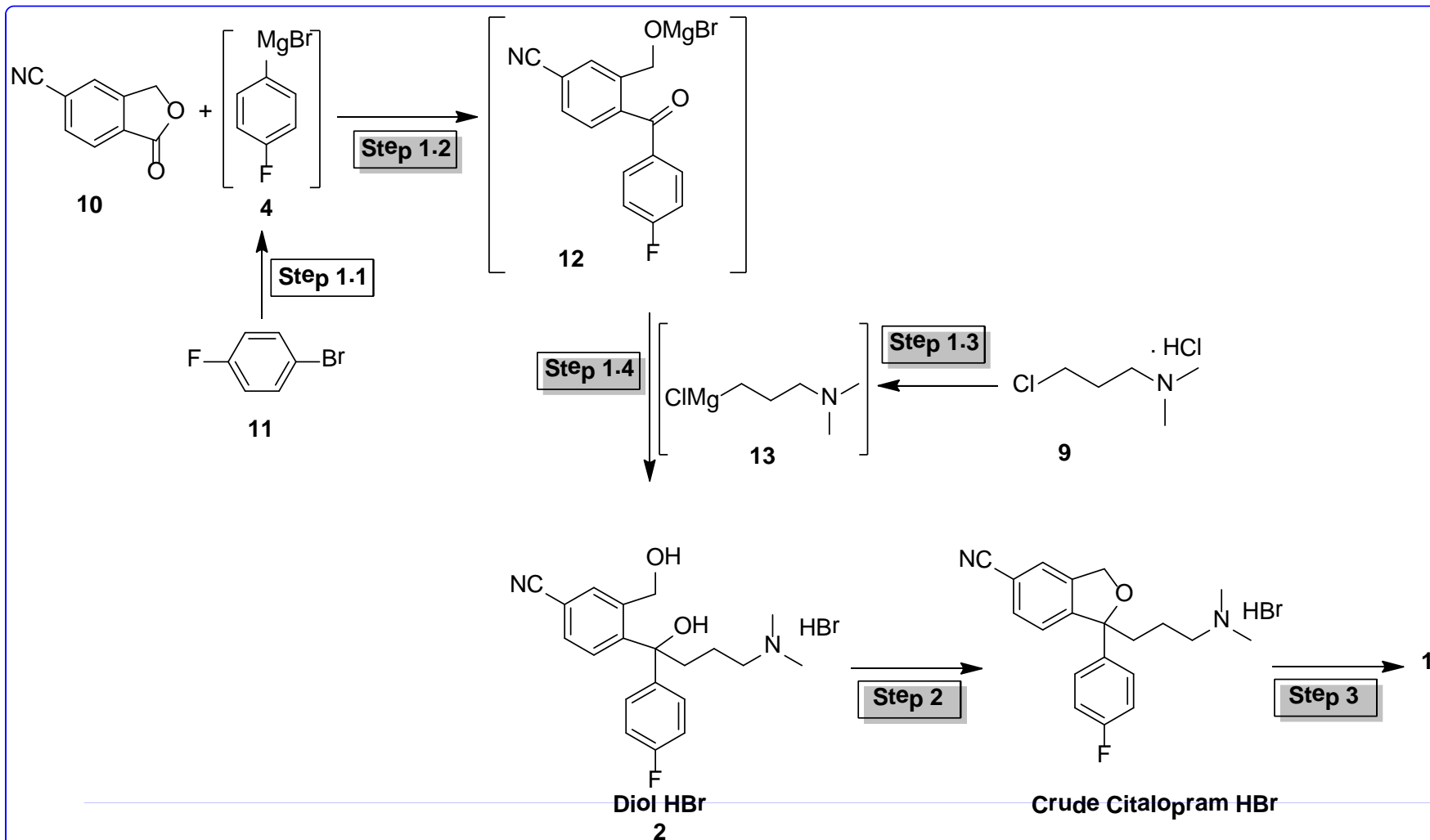
## First reported synthesis of Citalopram 1



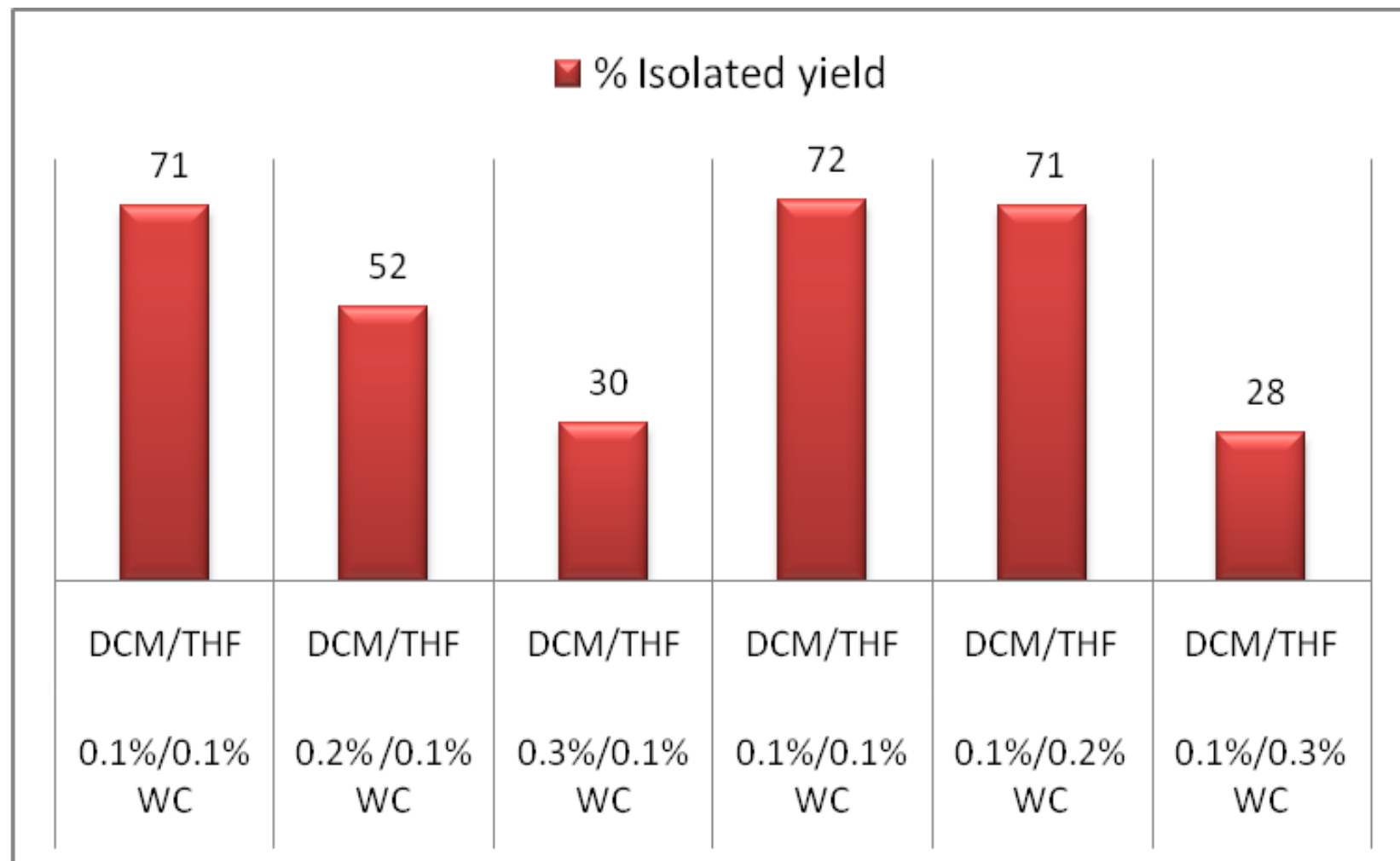
### Strategies for Process Improvements

1. Can we freshly prepare GR and use it *in situ*?
2. Can we avoid the use of LAH or Sodium borohydride reagent?
3. Can we avoid Copper cyanide?
4. Can we avoid NaH during alkylation?
5. Can we do most of the transformations at room temperature or at least can we avoid higher temperature ( $>100\text{ }^{\circ}\text{C}$ ) ?
6. Is it possible to telescope this process to all possible extent?
7. ....etc

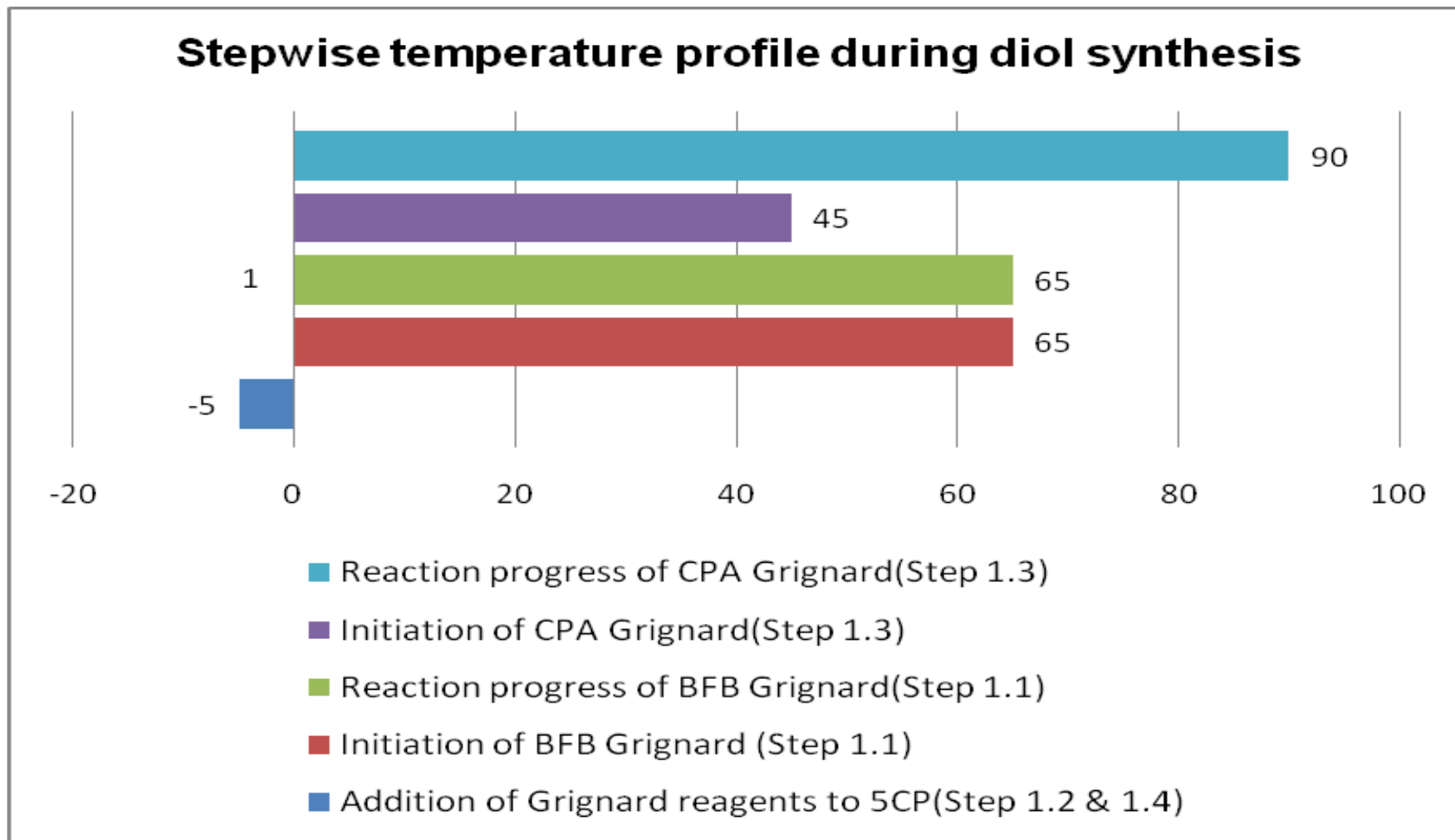
## Outline Towards Realizing the Strategies : One Pot Synthesis of Diol HBr



## Role of Solvents and Inherent KF



## Temperature Profile





### DoE: Full Factorial

1. Pre DoE experiments
2. Based on domain knowledge, deciding on the variable and response factors
3. Use of software e.g. Design Expert
4. Augmentation of initial results with the help of Response Surface Model to arrive on the optimal conditions
5. Analysis of results by considering ANOVA variance method to derive significant model

## Operable ranges for DoE based on pre-DoE experiments

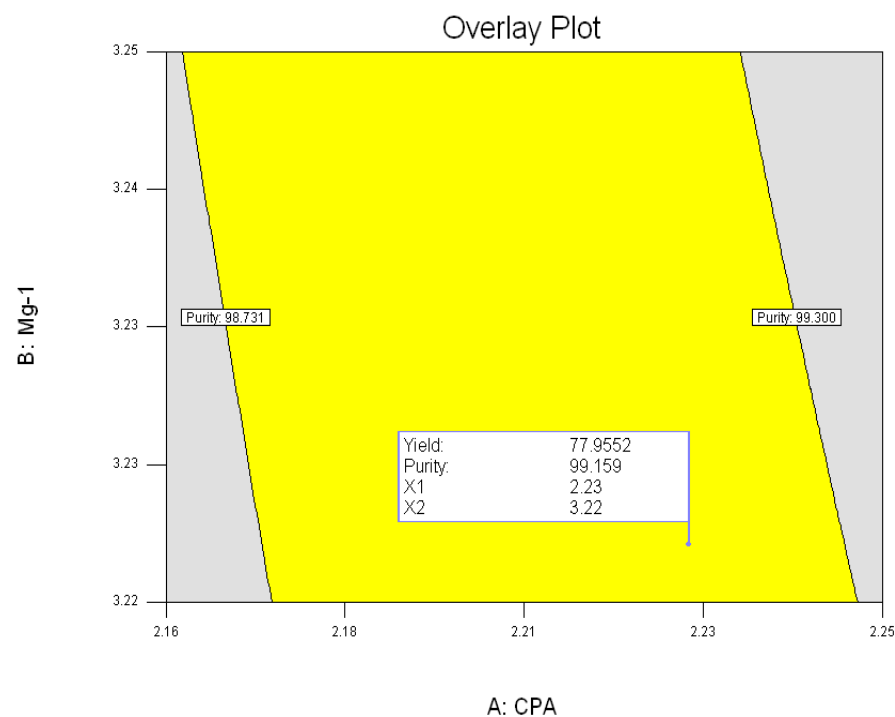
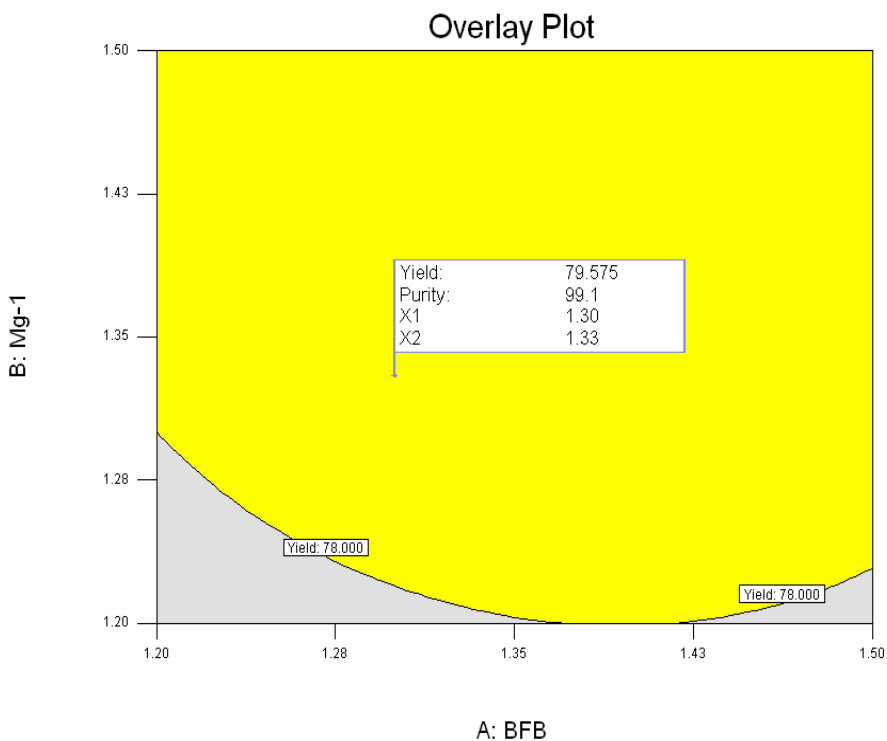
process variables	low	high
BFB (mol equiv.)	1.2	1.5
Mg (mol equiv.)	1.2	1.5
Iodine (% w/w)	1	10

process variables	low	high
CPA (mol equiv.)	1.8	3.0
Mg (mol equiv.)	2.0	4.0
Iodine (% w/w)	1	10

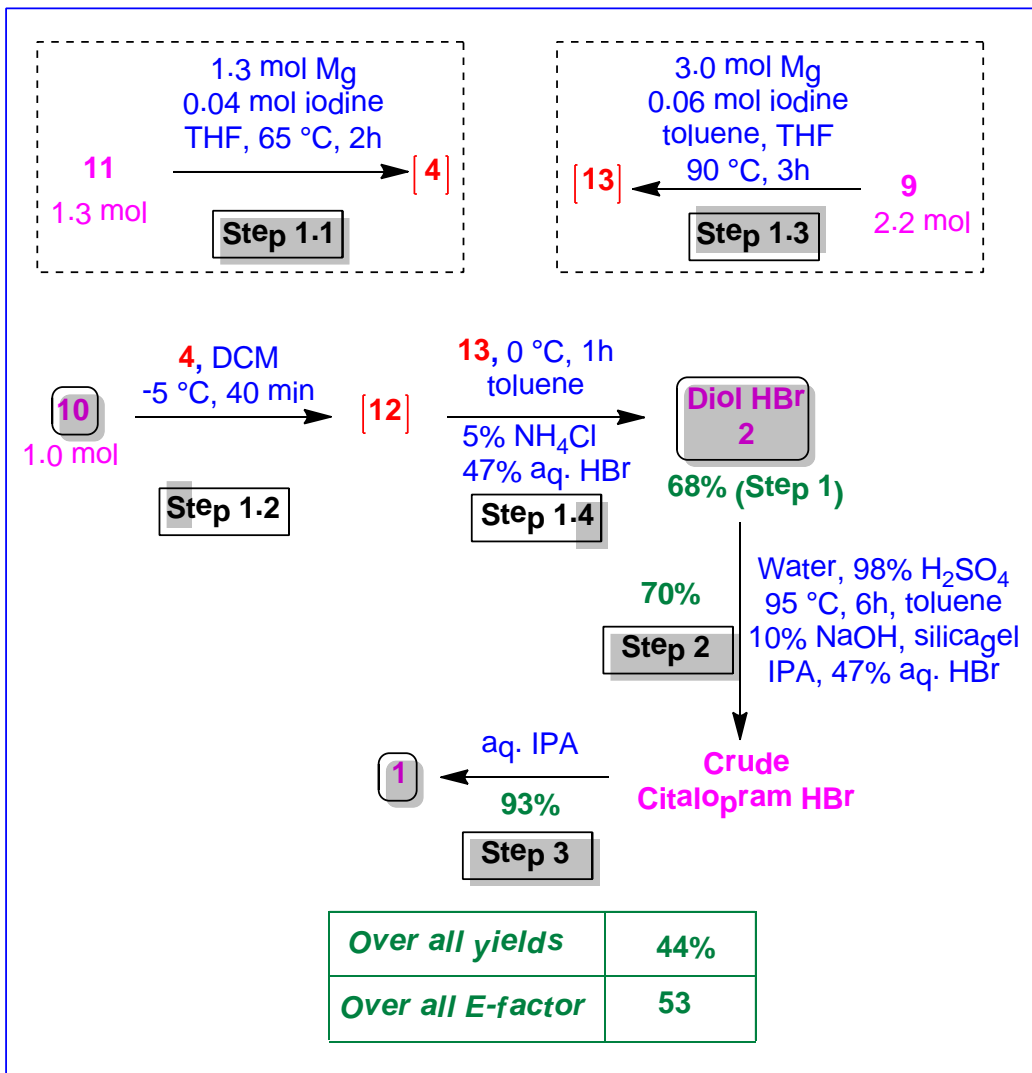
### Response Factors

>75% yield and >98% purity

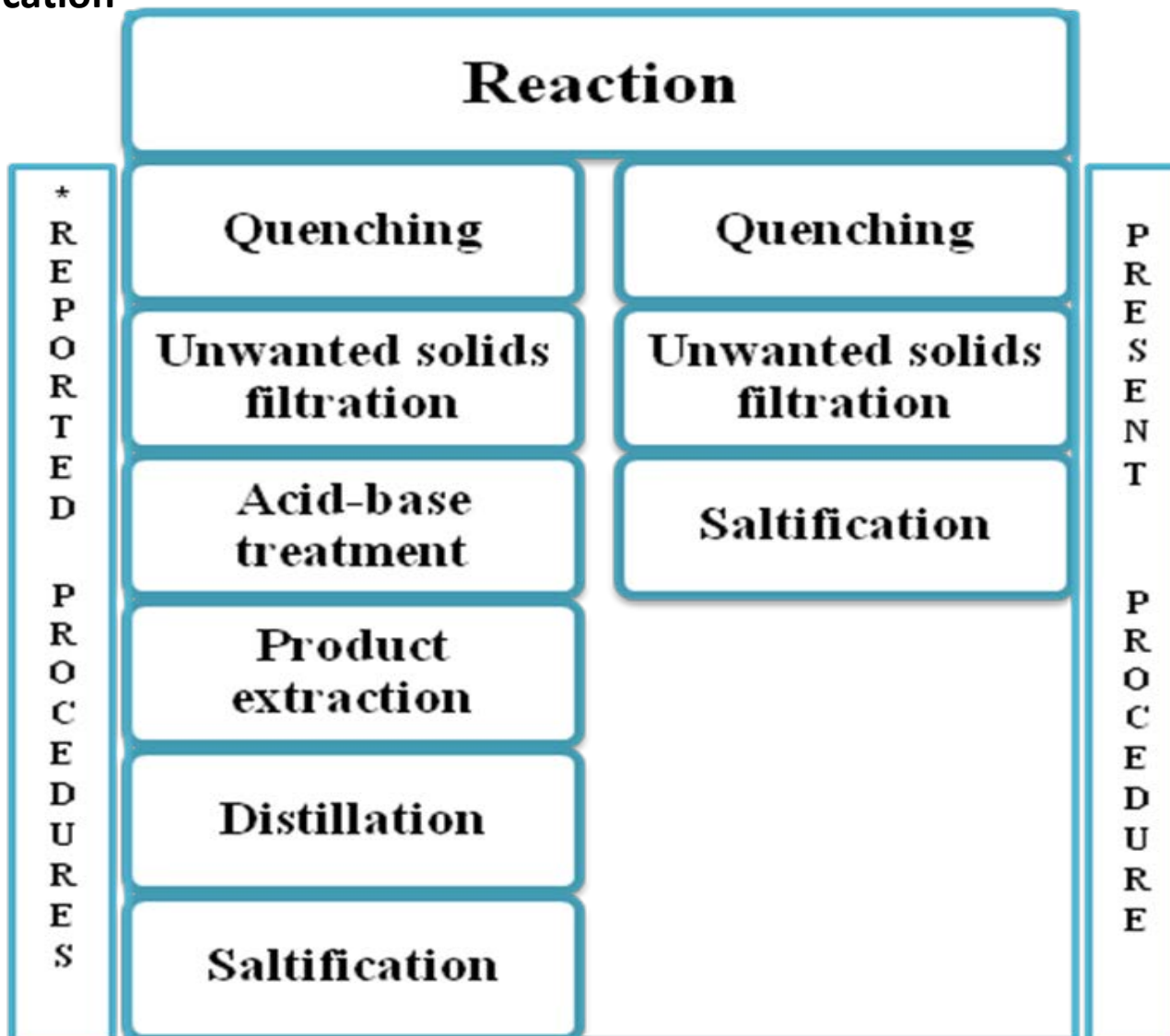
## Design space obtained for BFB and CPA Grignard reactions



## Optimized Conditions



Work up Simplification



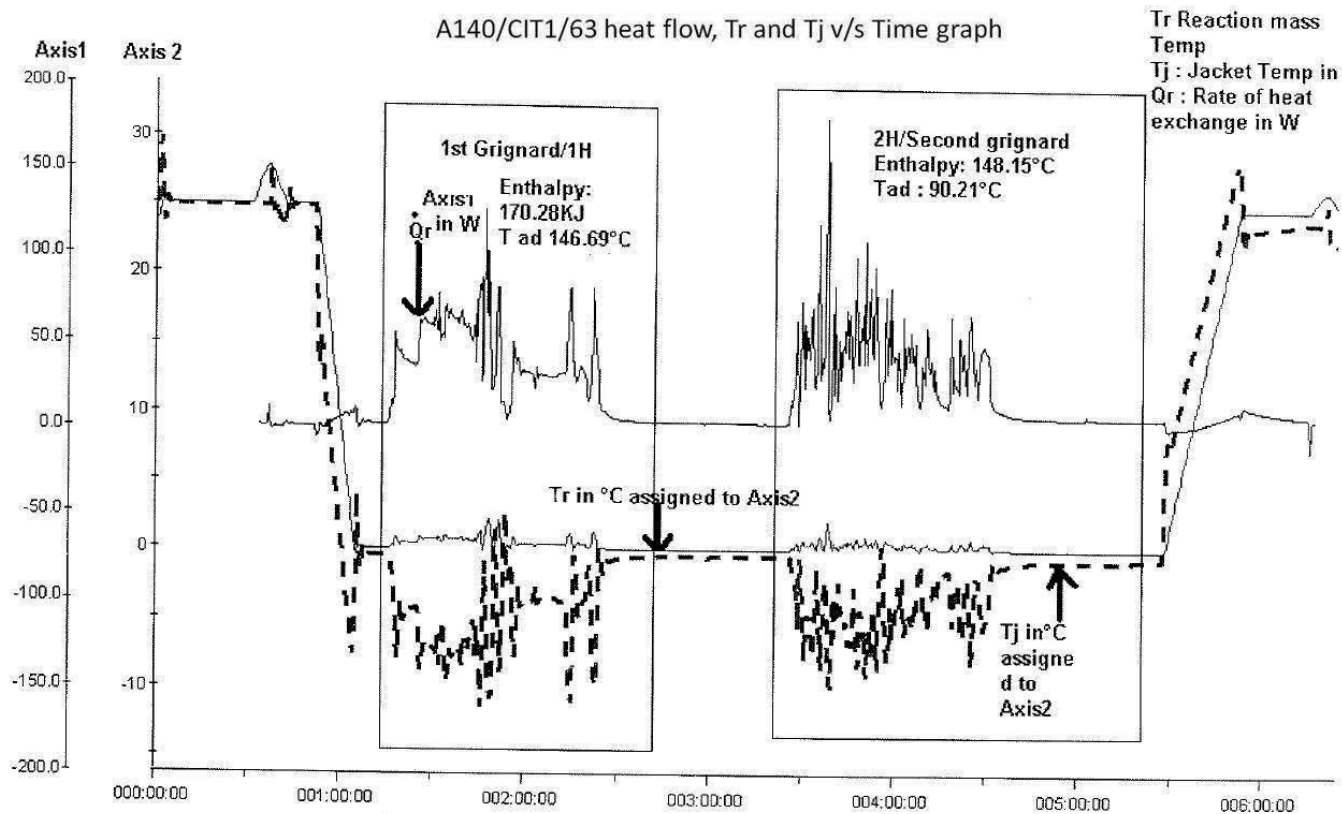
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### Safety Consideration

1. Understanding the heat of reaction and adiabatic temperature rise
2. Comparison of the batch temperature profile inside the reactor ( $T_r$ ) with the reactor jacket temperature profile ( $T_j$ ) in isothermal mode reveals whether the reaction is instantaneous or not. (Exo or Endothermic)
3. Enthalpy can be calculated which indicates temperature rise in a given batch size
4. This helps to avoid accidents at a scale by keeping control system in place without compromising on process variables and responses

## Safety Consideration



For 100g input of **10**; Temperature rise for BFB Grignard is from 0 °C to 3 °C, E=170.2 KJ and Tad=146.69 °C; Temperature rise for CPA Grignard is from 0 °C to 1.5 °C, E=148.15 KJ and Tad=91.2 °C. Recommendation: Rate of heat exchange must be controlled by keeping efficient cooling in Jacket

### Appreciation

Michael E Kopach <kopach\_michael@lilly.com>

05/10/2013 04:45 AM

To

"rakeshwarb@drreddys.com" <rakeshwarb@drreddys.com>

Cc

Subject

Citalopram Paper

Dear Rakesh,

From on Grignard practioner to another – this is an outstanding paper:

<http://pubs.acs.org/doi/abs/10.1021/op3002596>

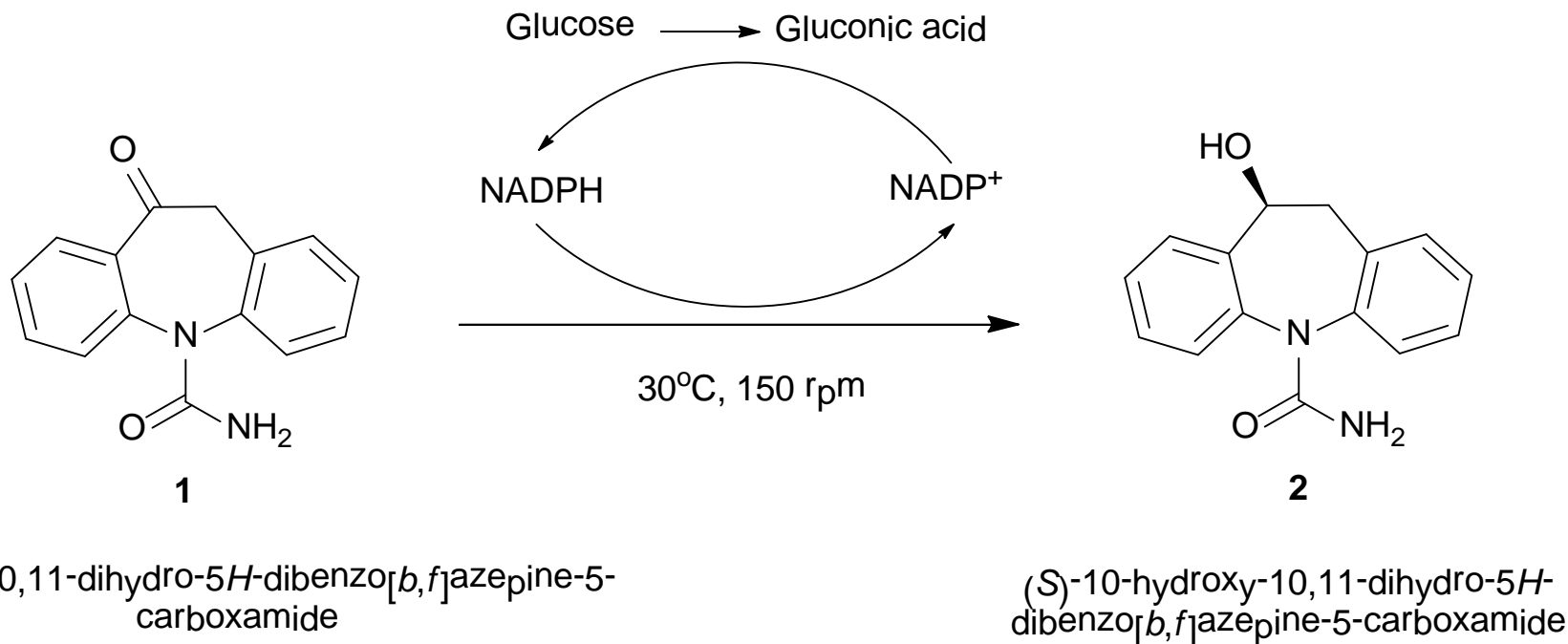
Best Regards,

Mike

*Org.Proc.Res.&Dev.* **2013**, 17, 798-805.

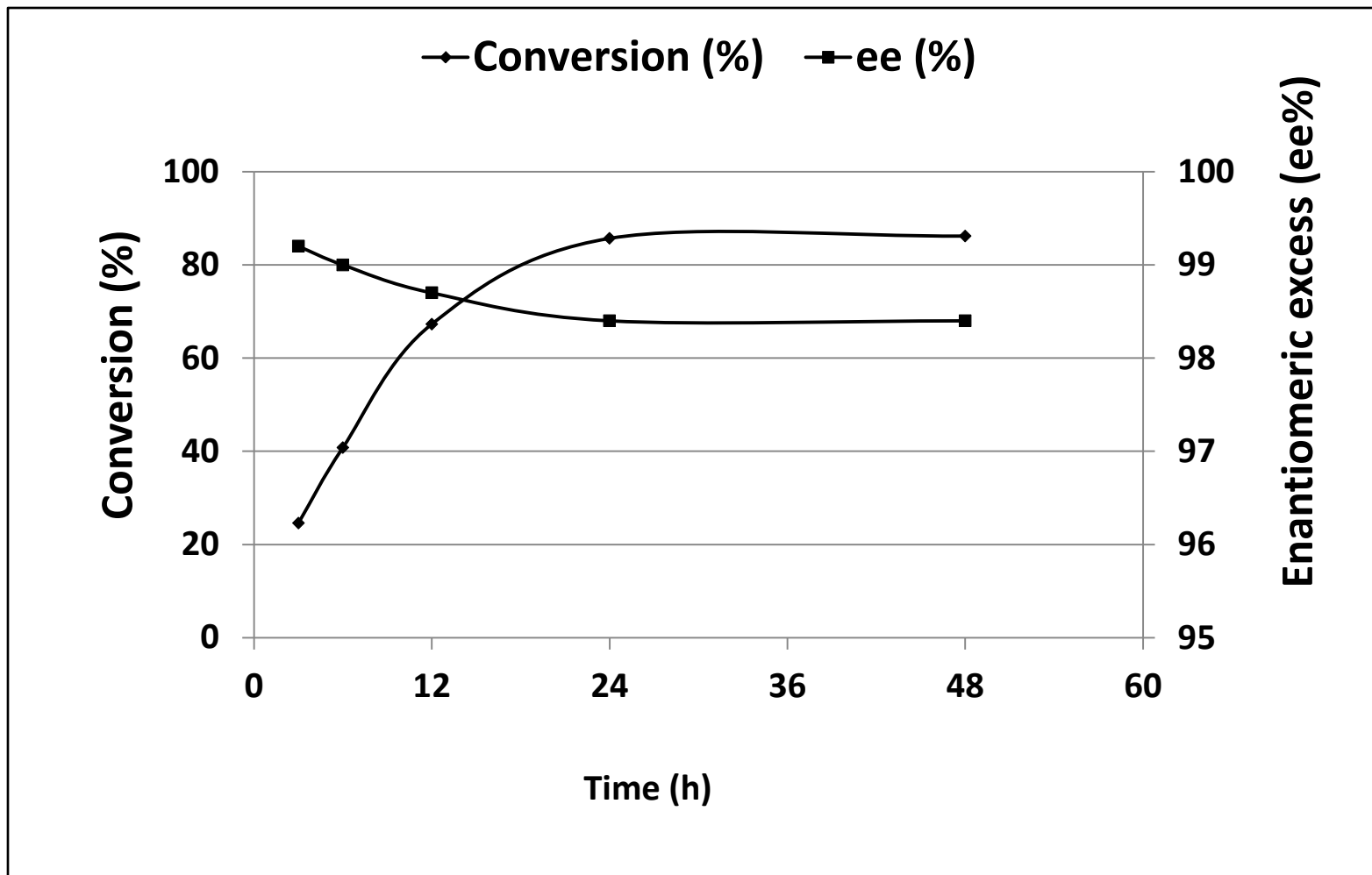


## Asymmetric Reduction of a Key Intermediate of Esclicarbazepine Acetate Using Whole Cell

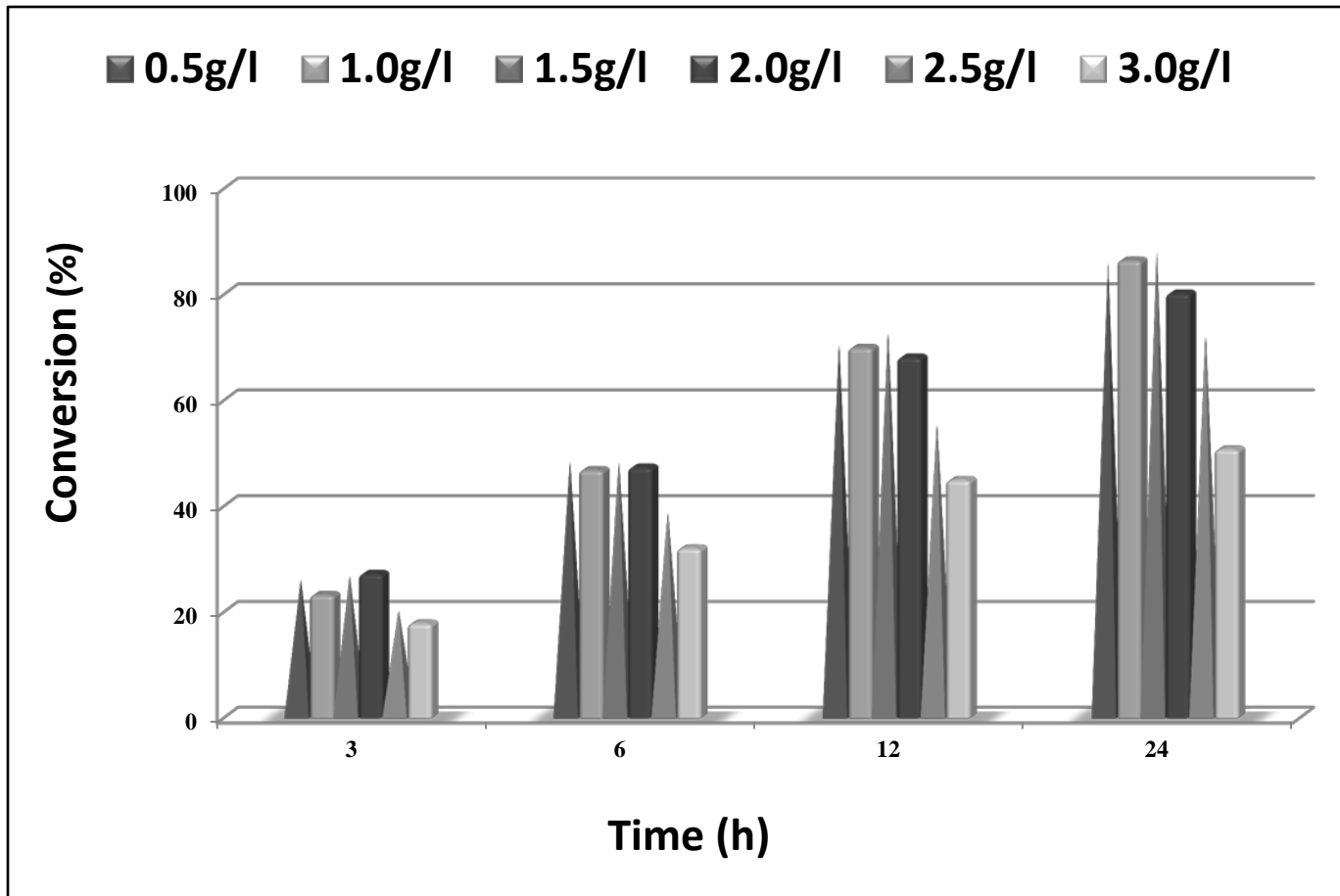


*Catalysis Science & Technology* **2012**, 2, 1602-1605. (**One of the Hot Articles**).

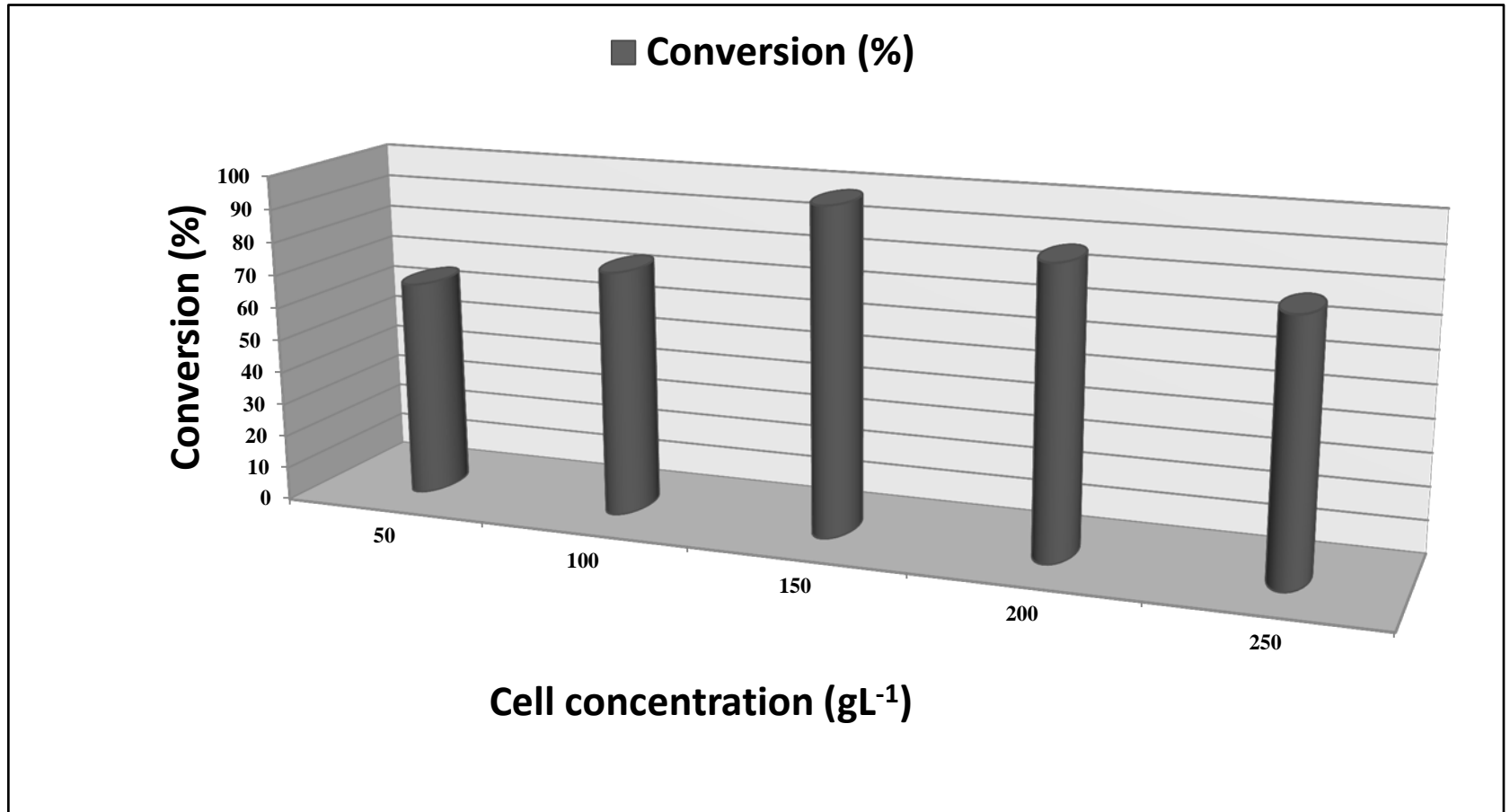
Bioreduction of Ketone 1 by *P. methanolica* whole-cells in biphasic system at 30°C



## Effect of Substrate Concentration



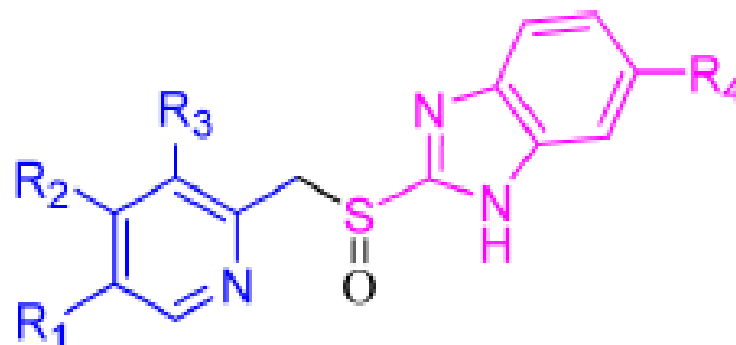
### Effect of Cell Concentration



## Output at a Scale with Optimized Conditions

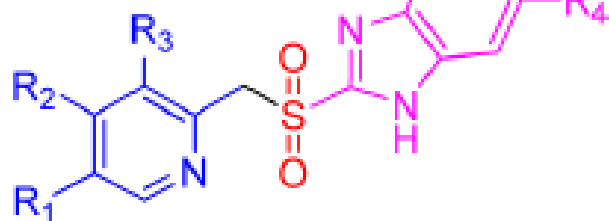
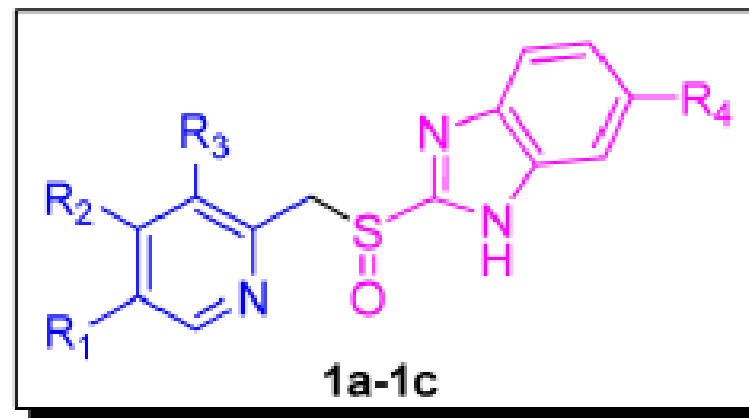
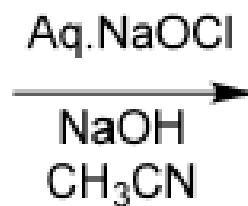
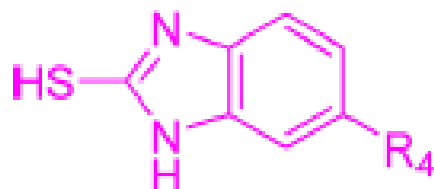
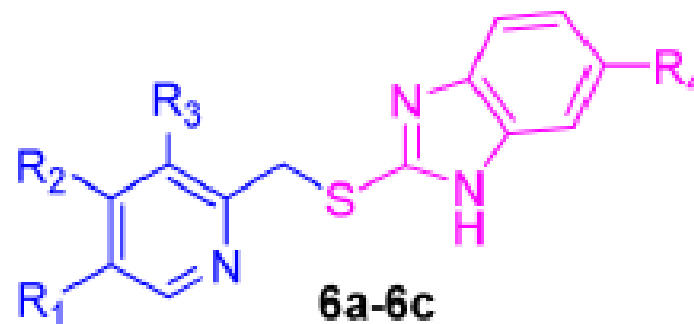
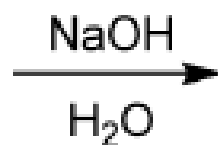
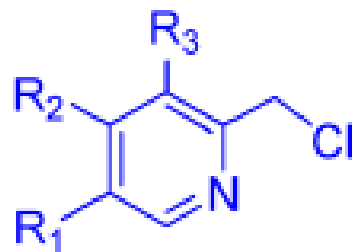
1. Resting cells of *P. methanolica* ( $150.0 \text{ g l}^{-1}$ ),  
H<sub>2</sub>O: hexane (2 L), glucose (0.5%)
  2. Reaction at 30 °C for 48 h
  3. 85% isolated with >98% *ee*

## Application of Flow Technology in the Process Development of Prazoles

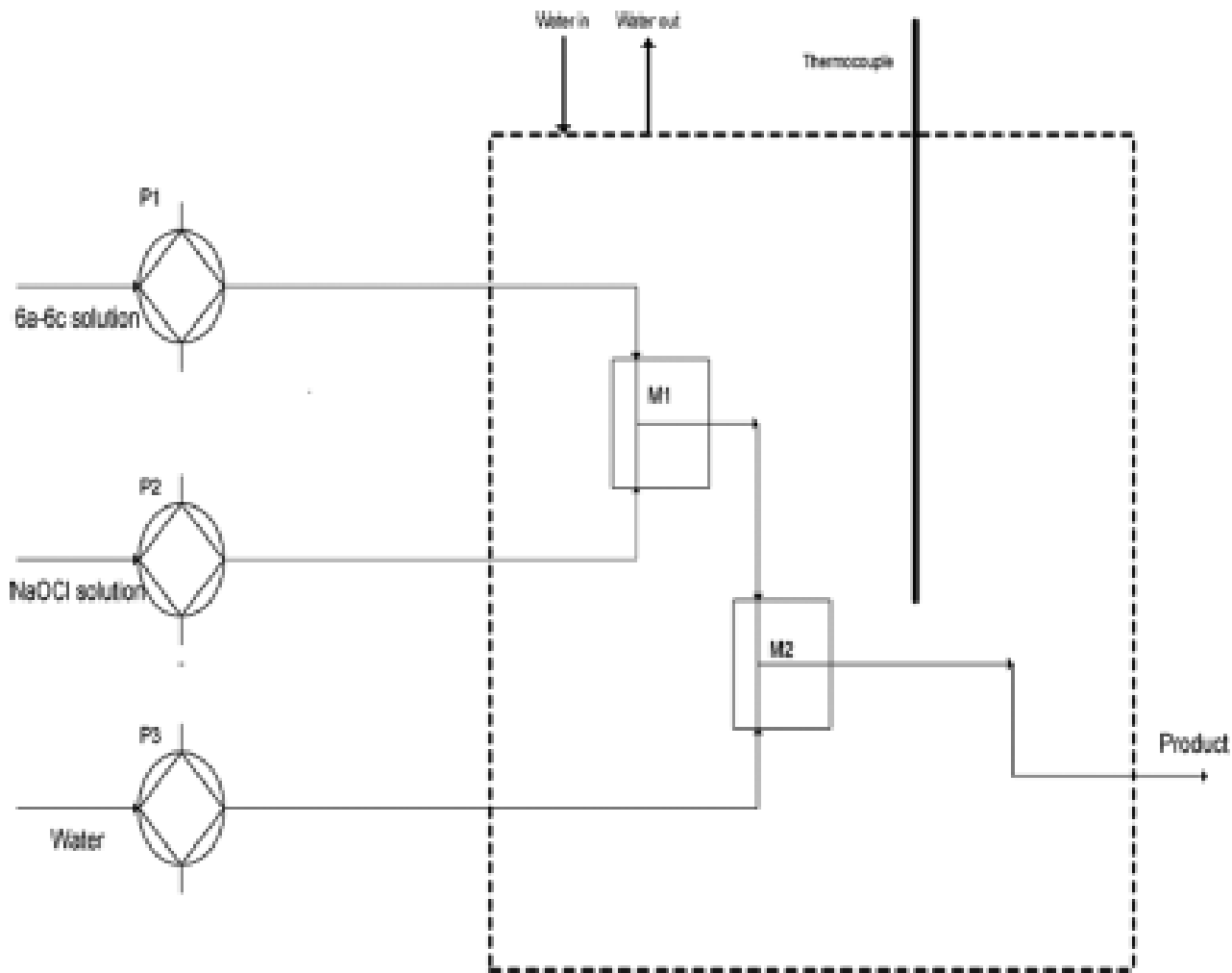


	Drug name	R1	R2	R3	R4
<b>1a</b>	Lansoprazole	-H	F <sub>3</sub> CCH <sub>2</sub> O-	Me	-H
<b>1b</b>	Pantaprazole	-H	MeO	MeO	F <sub>2</sub> CHO-
<b>1c</b>	Raberprazole	Me	MeO (CH <sub>2</sub> ) <sub>3</sub> O	Me	-H

### Synthesis of Prazoles



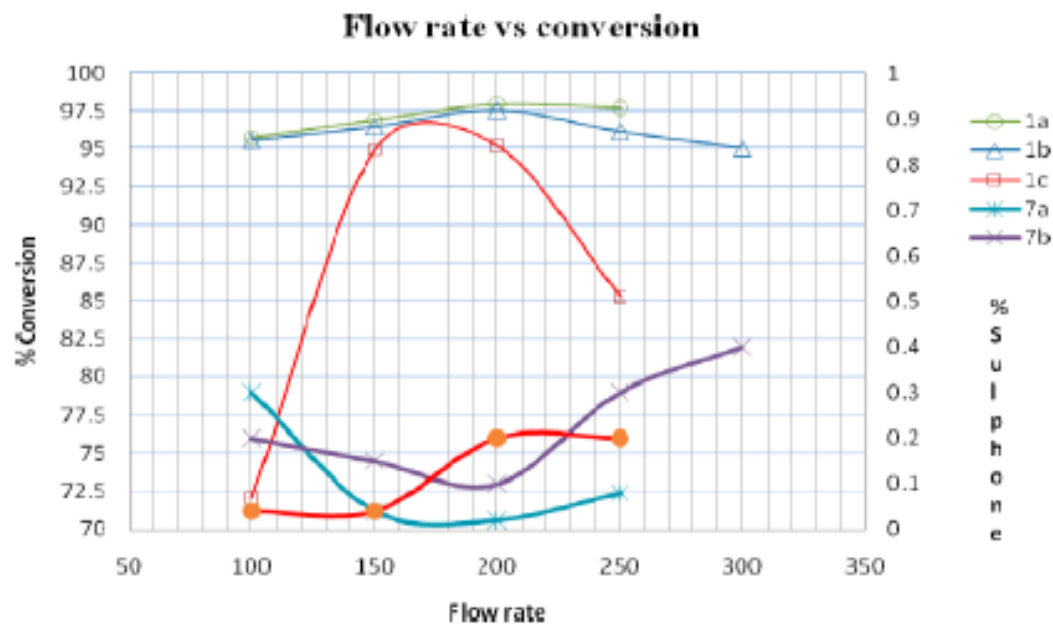
## Continuous Flow Micromixing Reactor Set up





## Effect of Flow Rate on Conversion

entry	flow rate (mL/min)	reaction mass HPLC purity (%)		
		6a	1a	7a
1	100	0.95	95.72	0.30
2	150	0.75	96.92	0.04
3	200	0.30	97.94	0.02
4	250	0.70	97.70	0.08



## Batch vs Flow

entry	oxid'n of 6	synthesis method	residence time
1	6a	batch process	2.5 h
2	6b		2.5 h
3	6c		2.5 h
4	6a	CFMMR <sup>ca</sup>	~1 s
5	6b		~1 s
6	6c		~1 s

## Batch vs Flow

before purification			after purification			
yield (%)	purity (%)		purity (%)		yield (%)	overall yield (%)
	1a-1c	7a-7c	1a-1c	7a-7c		
85.0	97.5	0.13	99.6	0.18	74.0	62.9
92.9	97.2	0.05	99.7	0.08	86.0	79.8
85.0	98.2	0.08	99.5	0.05	78.2	66.5
89.0	96.5	—	99.9	0.06	79.4	71.0
94.8	97.9	—	99.5	0.07	89.8	85.1
88.2	98.7	0.10	99.5	0.03	85.1	75.0

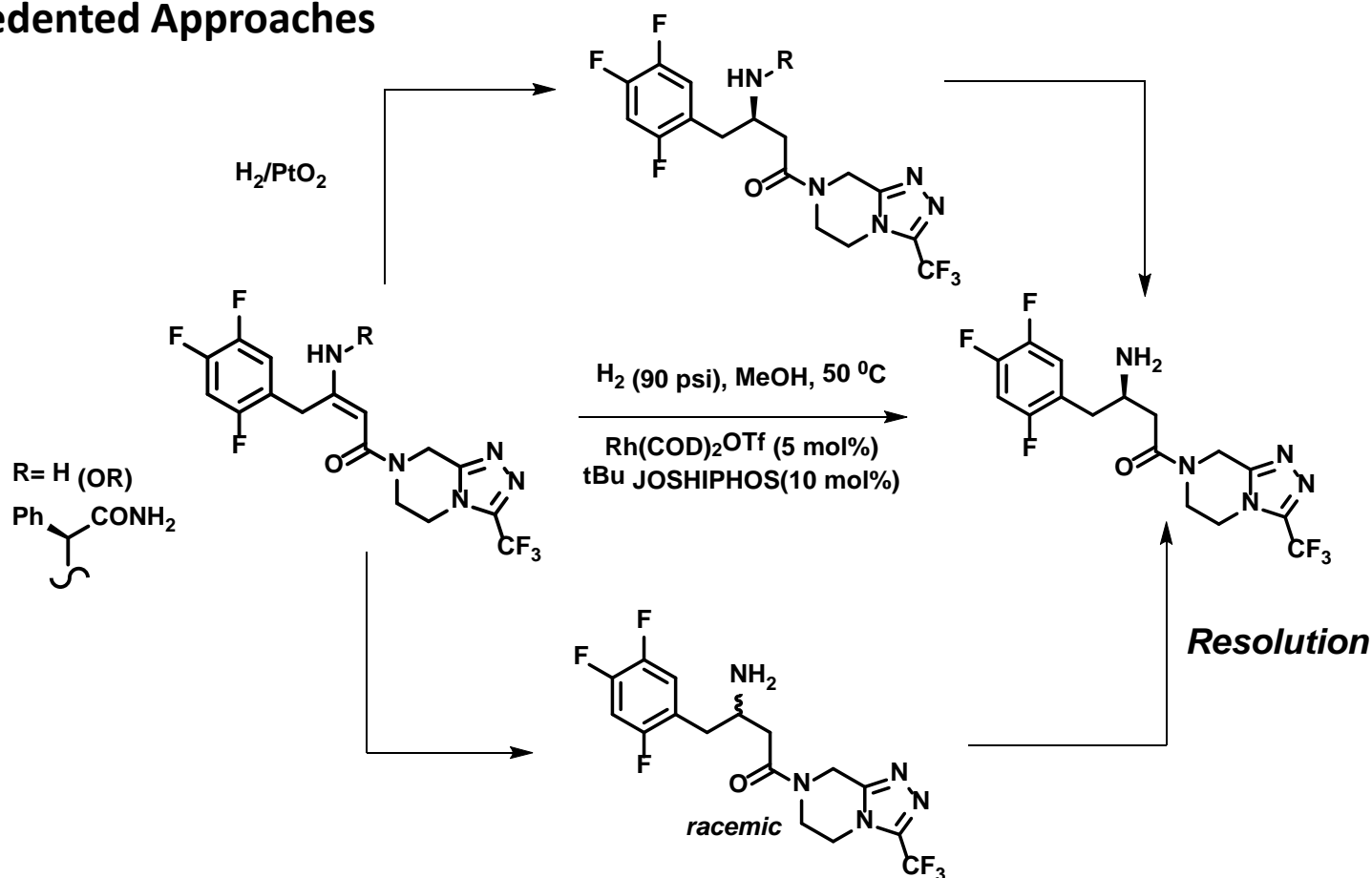
**First three entries are from Batch and last three belong to Flow**

*Org. Process Res. Dev.*, **2010**, *14*, 229–233 (DRL, India)

*Org. Process Res. Dev.*, **2013**, *17*, 1272–1276 (DRL, India)

# Major Innovation: Diastereoselective Reduction

## Precedented Approaches



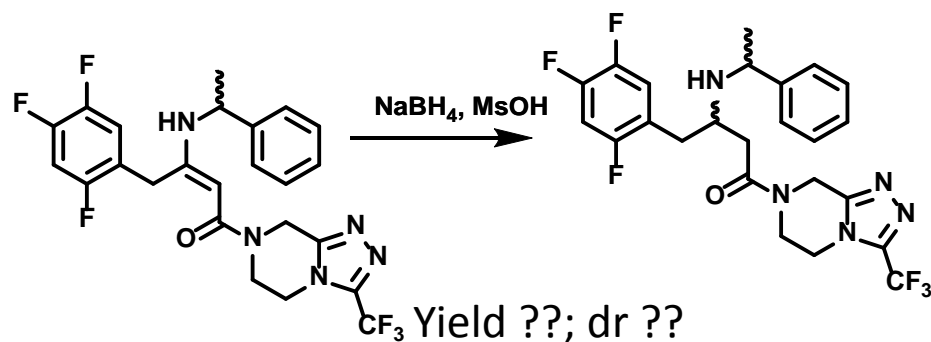
**Recently Developed Asymmetric Reduction Involving Biocatalysis is More Preferred**

Hansen, K.B.; Hsiao, Y.; Xu, F.; Rivera, N.; Clausen, A.; Kubryk, M.; Krska, S.; Rosner, T.; Simmons, B.; Balsells, J.; Ikemoto, N.; Sun, Y.; Spindler, F.; Malan, C.; Grabowski, E.J.J.; Armstrong III, J.D. *J. Am. Chem. Soc.* 2009, *131*, 8798-8804

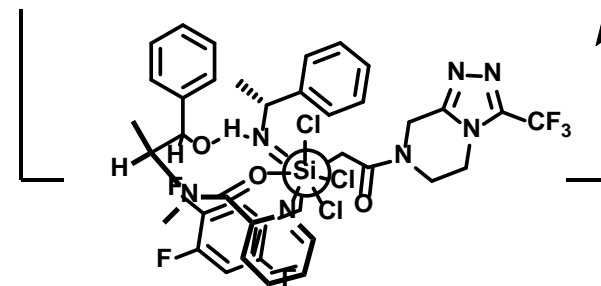
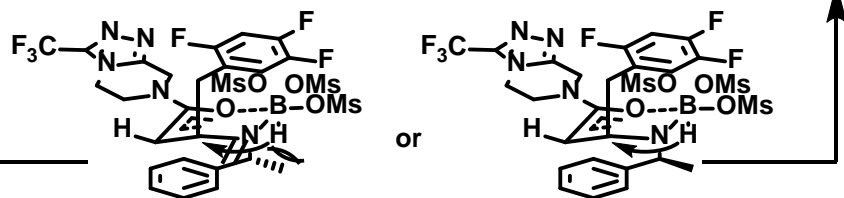
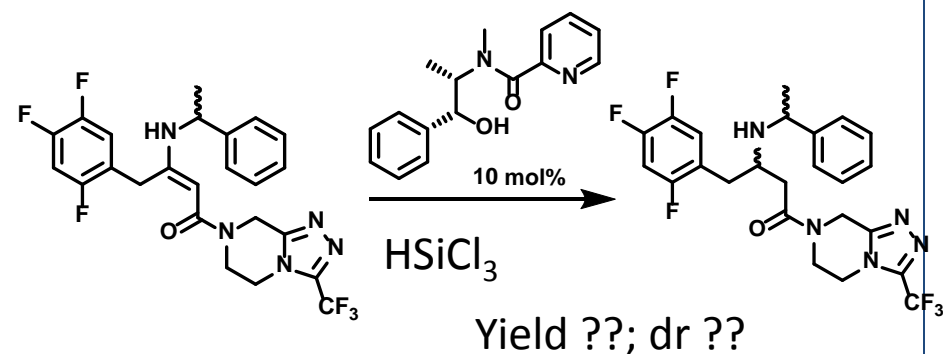
# Major Innovation: Diastereoselective Reduction

## Innovative Approaches

### Strategy I

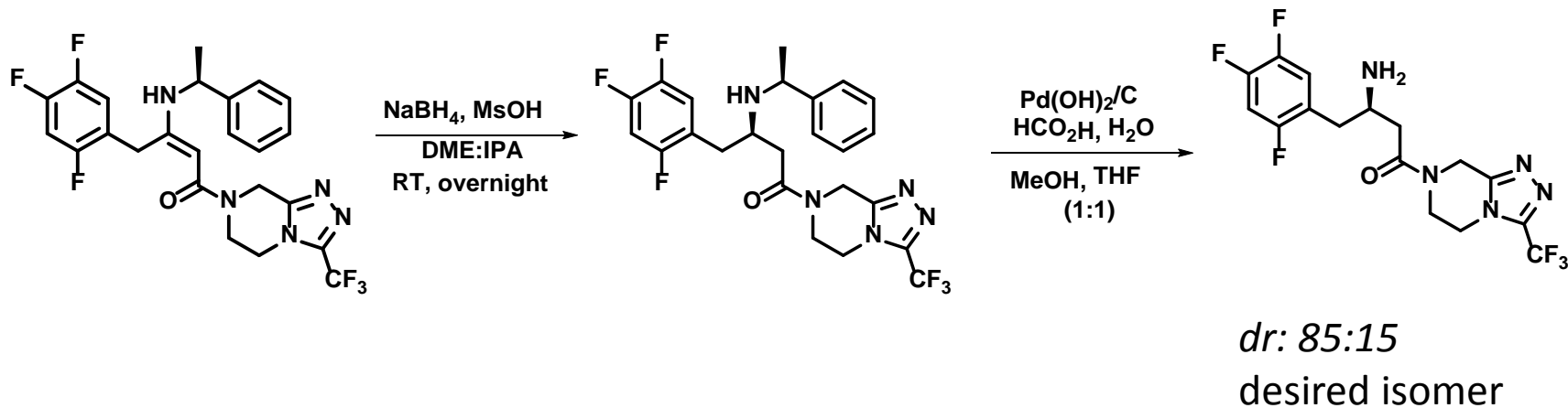
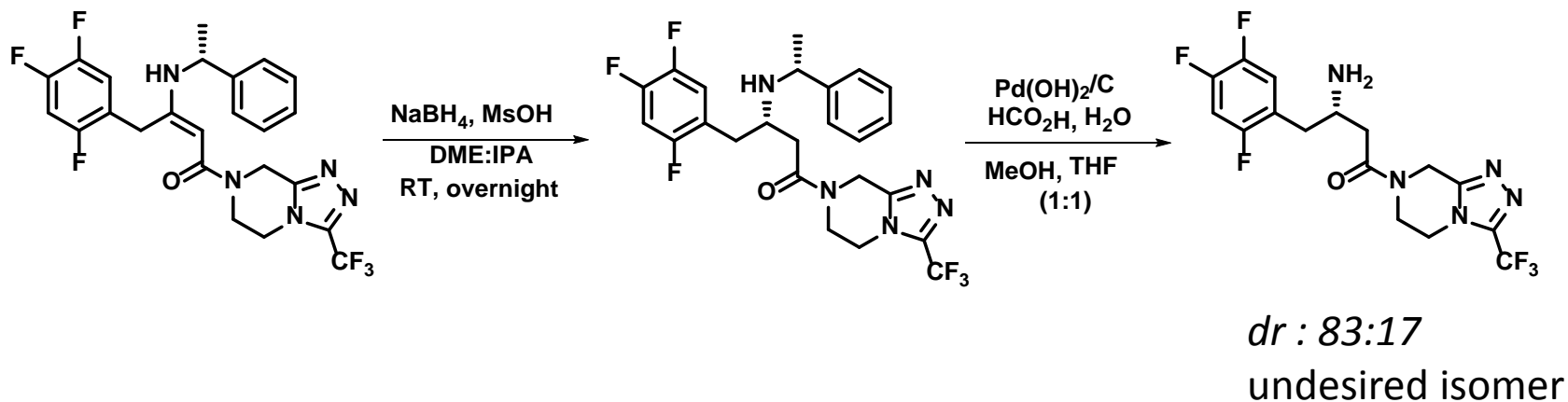


### Strategy II

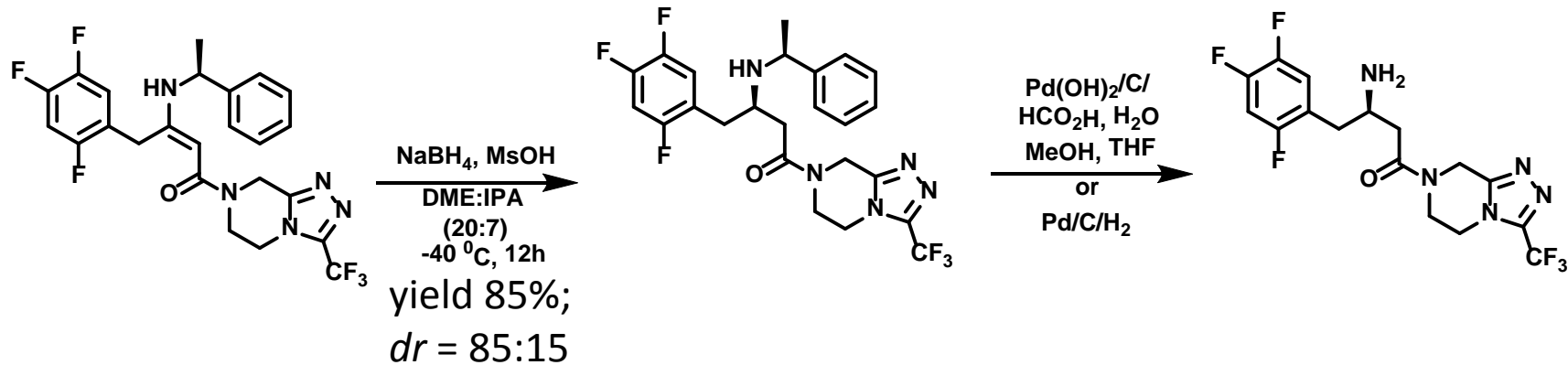


# Major Innovation: Diastereoselective Reduction

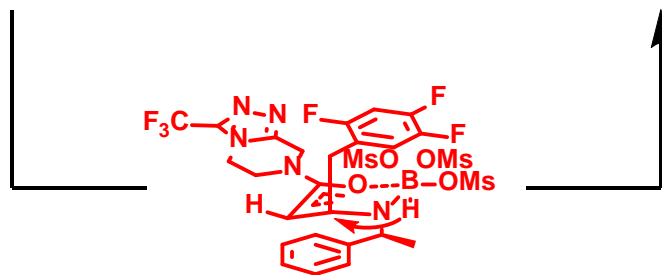
## Evolution of Concept



# Major Innovation: Diastereoselective Reduction



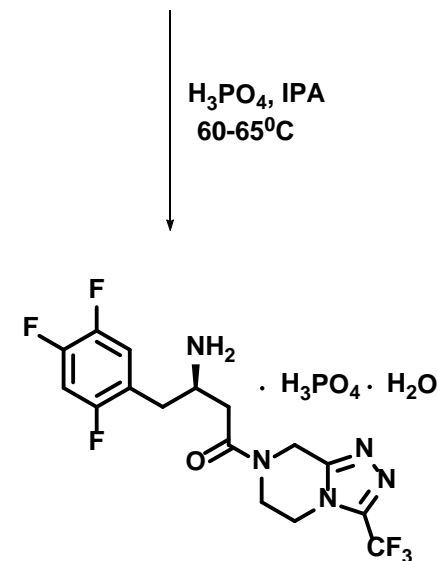
Required diastereomer crystallized with  
 $dr > 99\%$



$-90^\circ\text{C}$ :  $dr = 93:07$ , Yield =  $> 90\%$

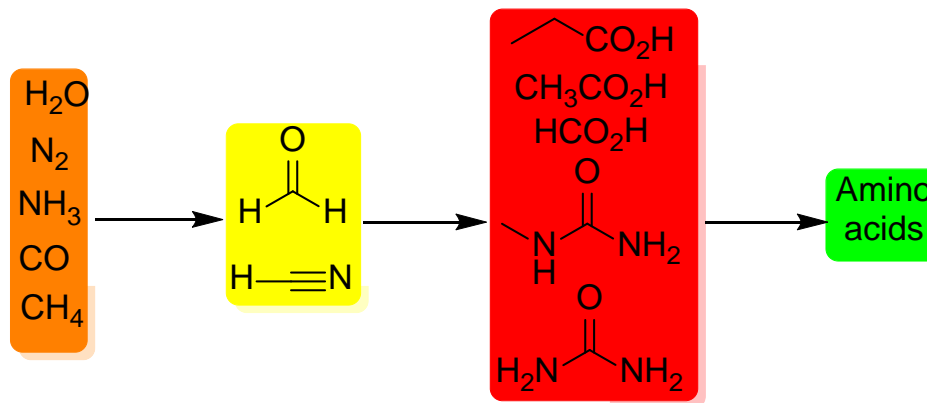
**7000 vs 700 USD**

Bandichhor, R.; et al. WO 2011025932 A2 20110303

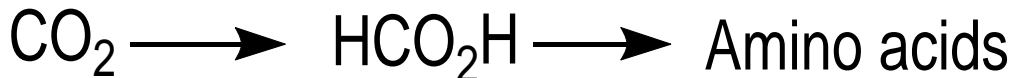


## Discovery of Redox System Enabling C-N-C Bonds Formation: Indicator of Prebiotic Synthesis of Amino Acid

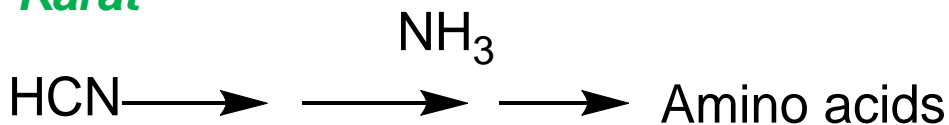
### Miller-Urey



### Calvin

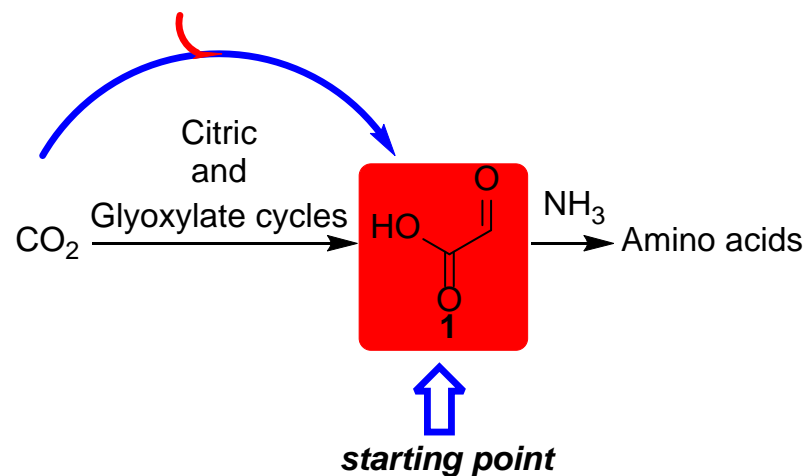


### Karat



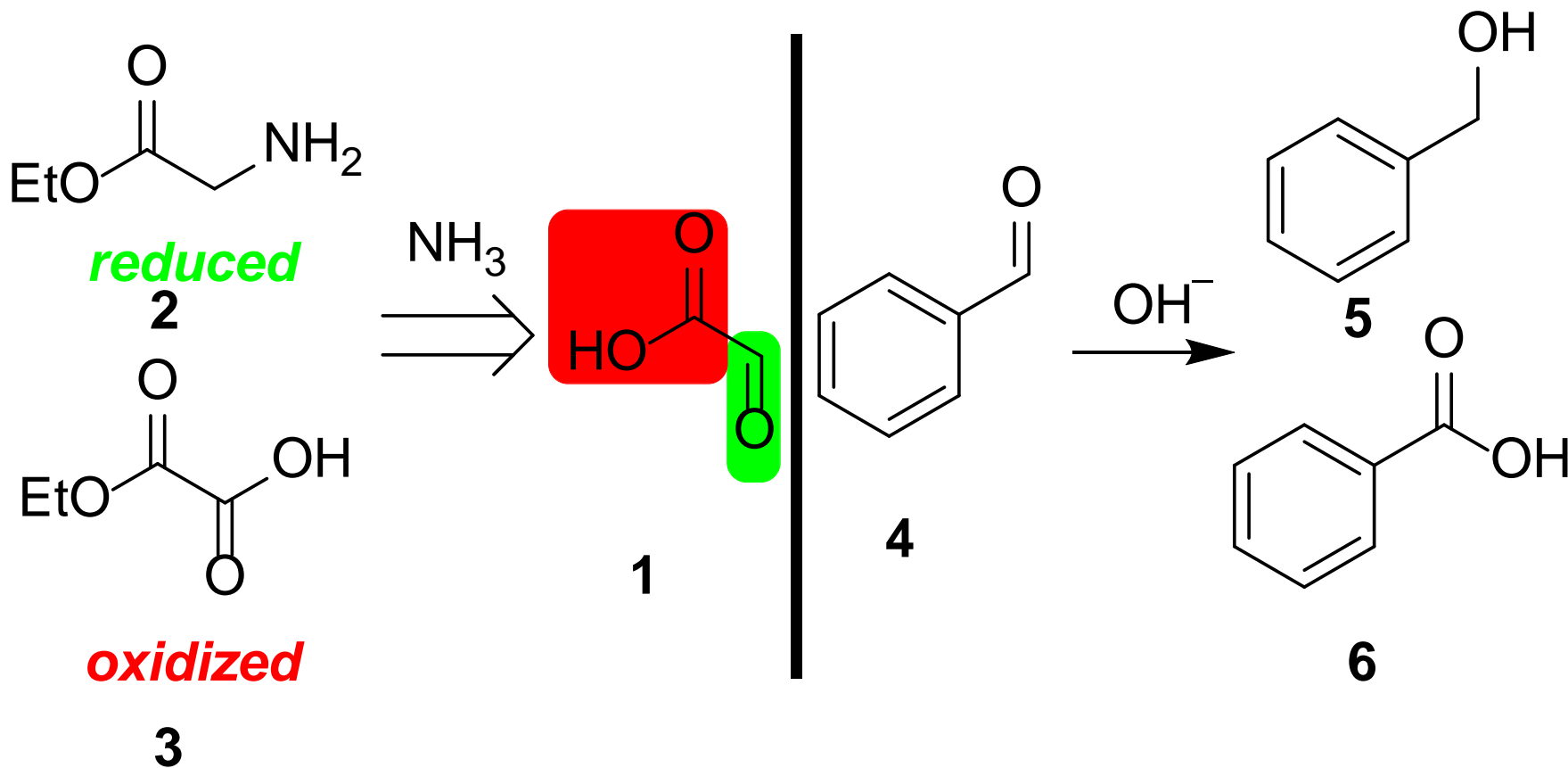
### Our work

*prebiotic high energy induced*

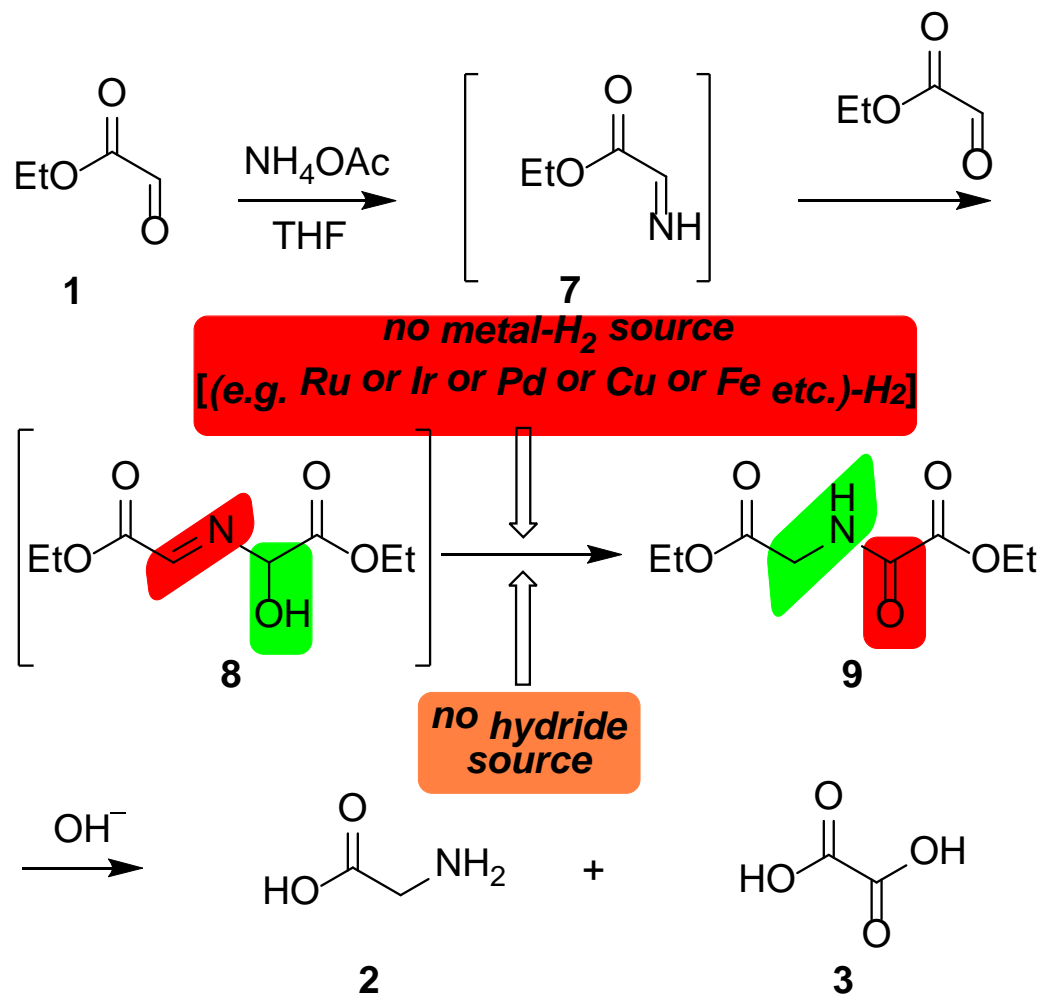


Different conceived approaches towards the synthesis of amino acids

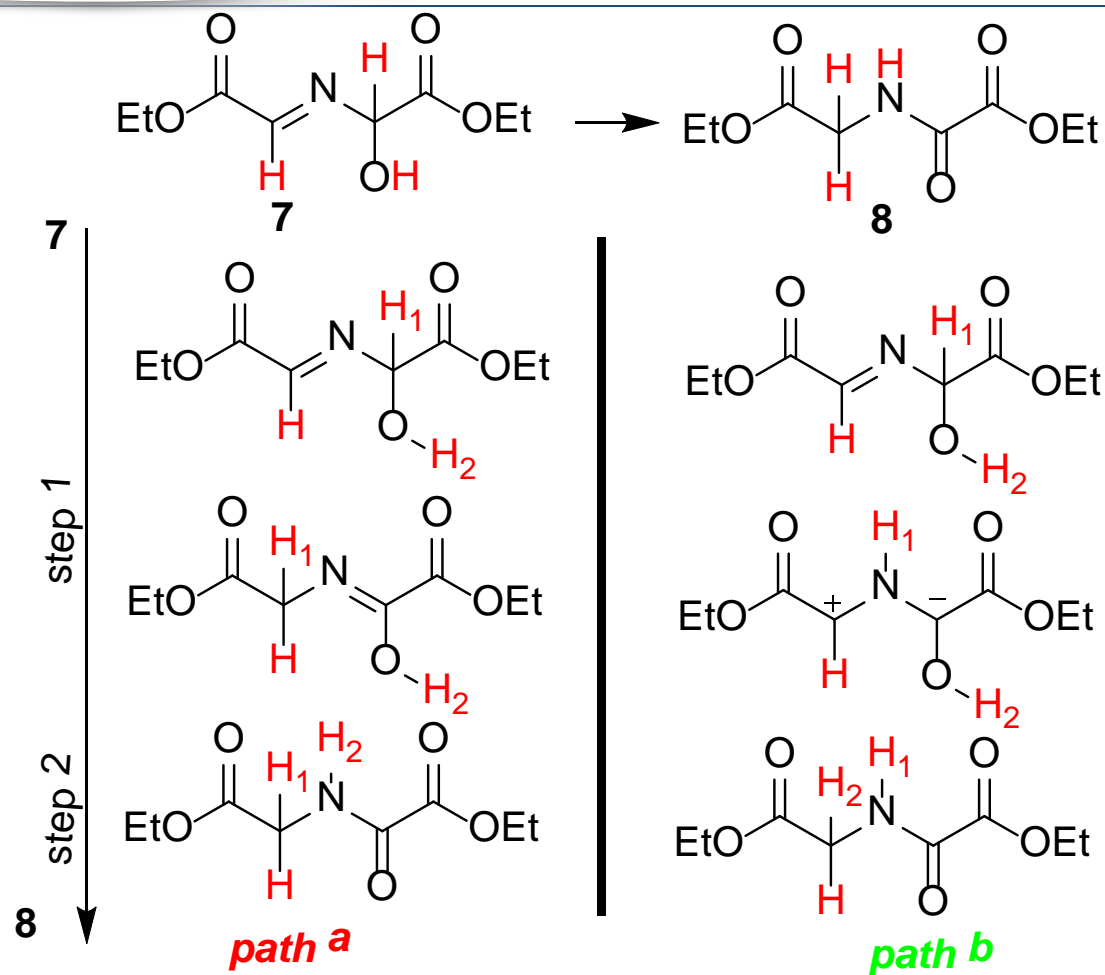




Redox chemistry on ethyl glyoxylate and Cannizzaro reaction

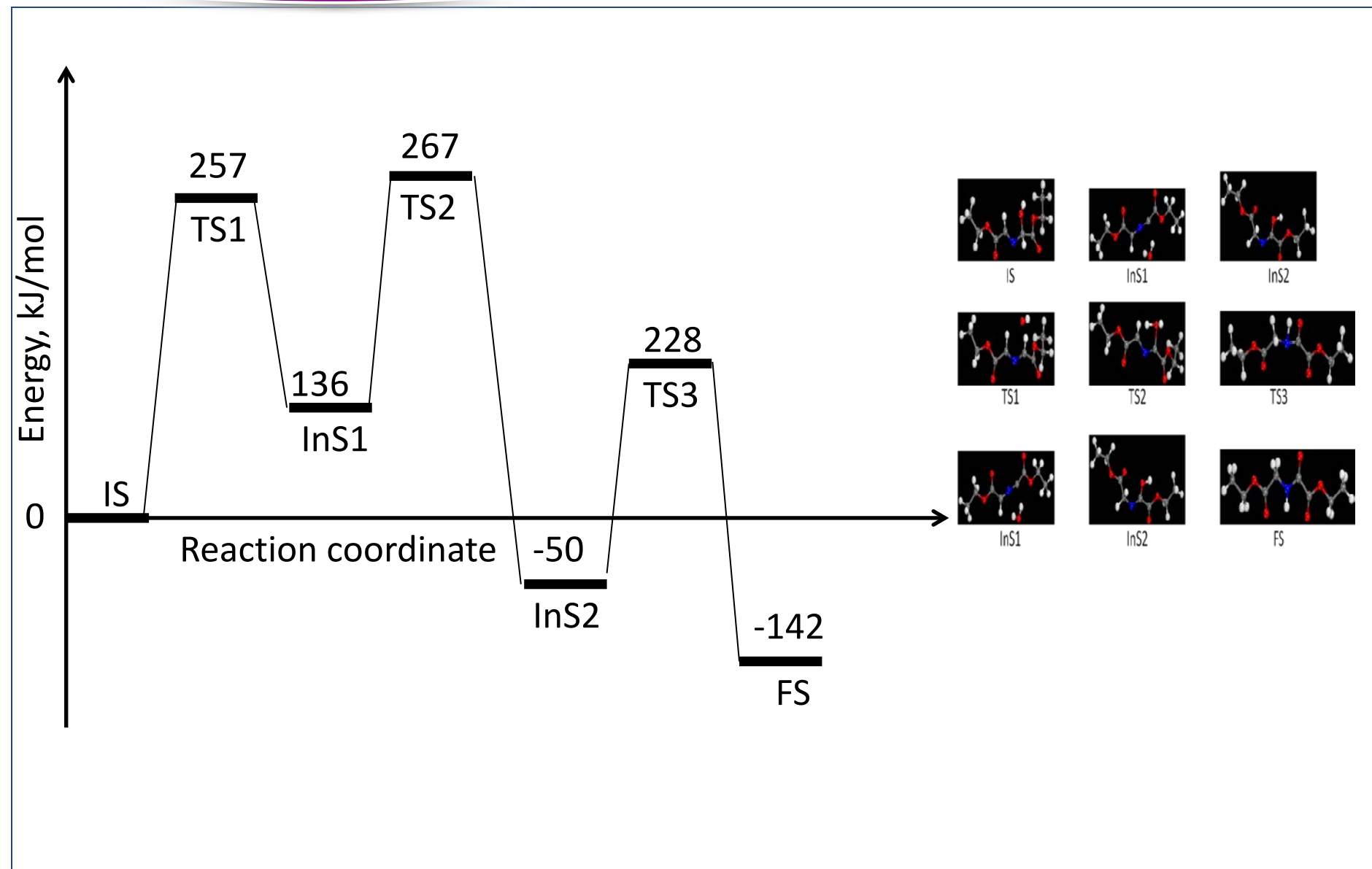


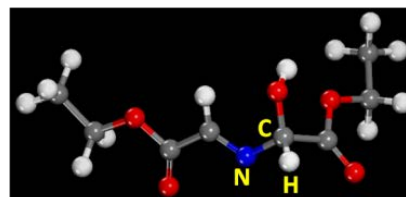
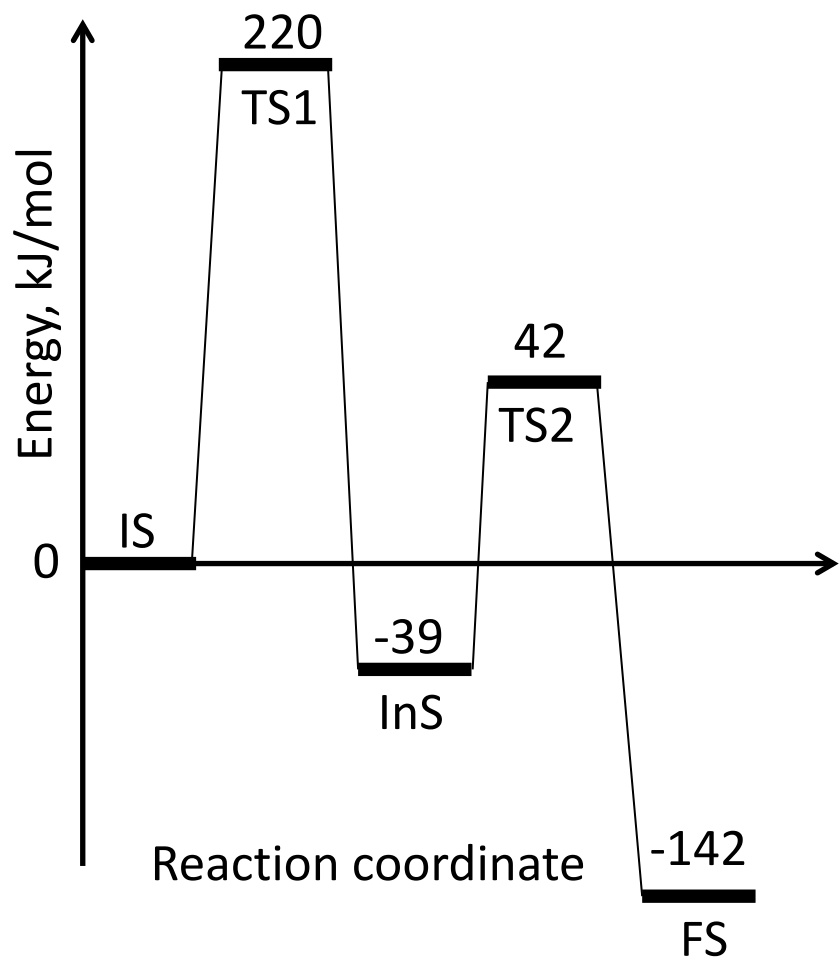
Synthesis of glycine 2



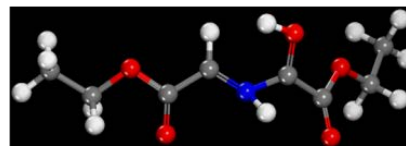
Two different reaction pathways (a and b)

# Major Innovation: Amino Acid

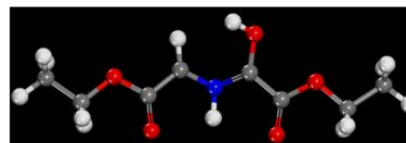




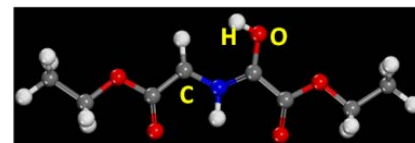
IS



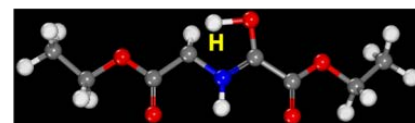
TS1



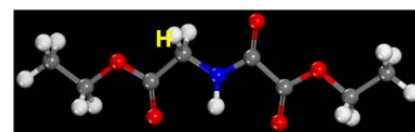
InS



InS



TS2



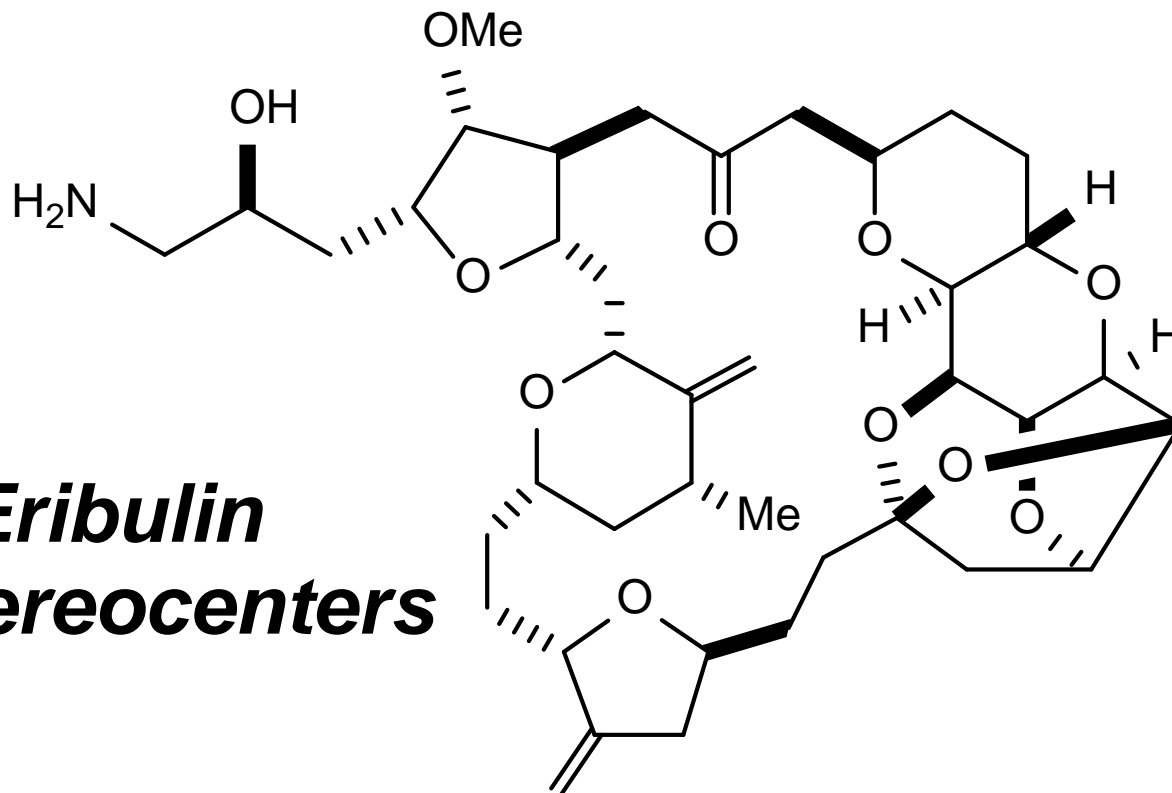
FS

Bandichhor et al. *Chem. Com.* **2014** under revision

**Major Innovation: Synthesis of Eribulin**

**Target 64 to 20 Steps**

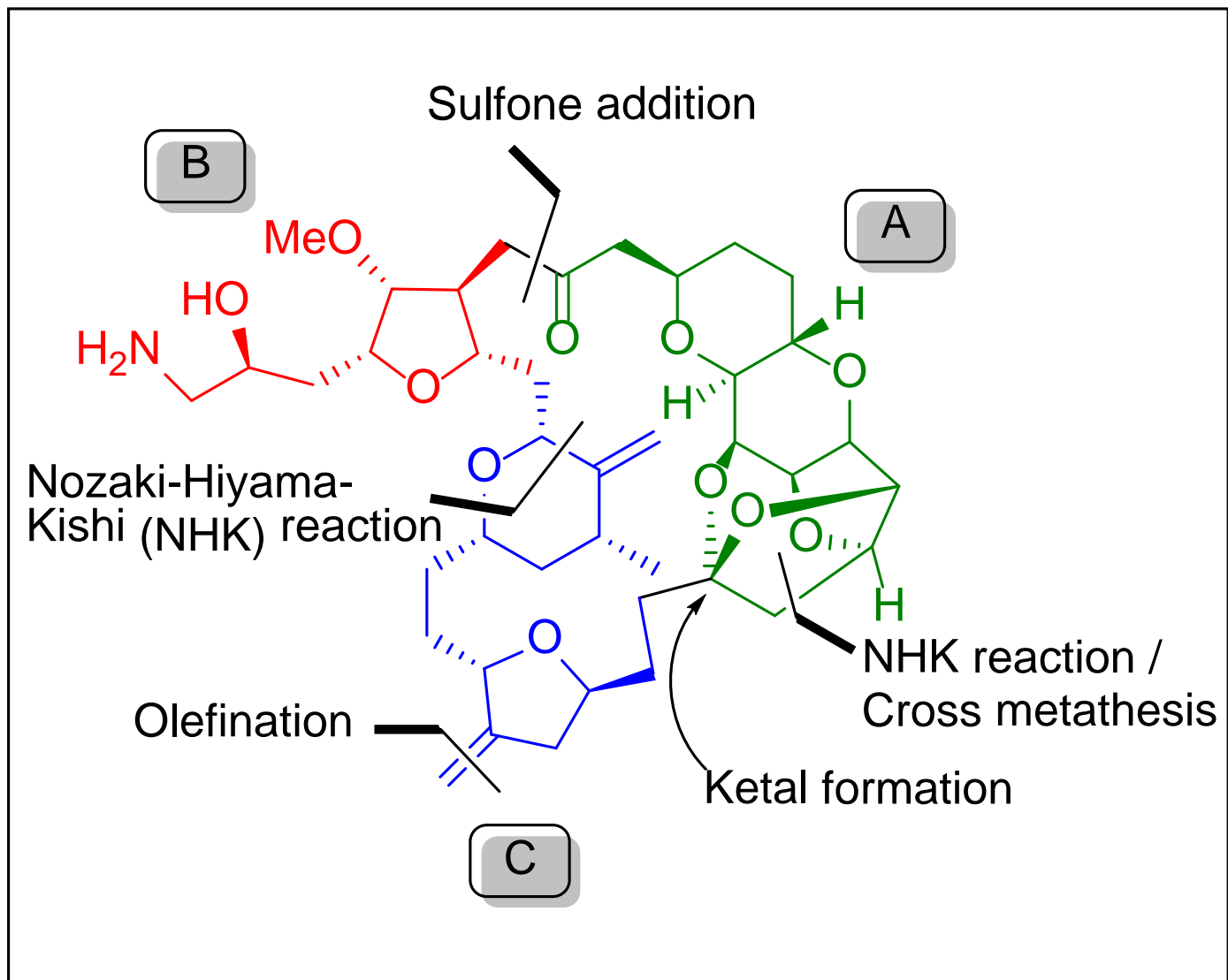
HALAVEN is a clear, colorless, sterile solution for intravenous administration. Each vial contains 1 mg of eribulin mesylate as a 0.5 mg/mL solution in ethanol: water (5:95).



**Eribulin**  
**19 stereocenters**

**Late Stage Breast Cancer (Two Chemotherapies and Treated With Anthramycin and Taxane Class of Medicine)**

# Major Innovation: Synthesis of Eribulin



***“A chain is as strong  
as its weakest link”***

Reading Materials:

1. Scalable Green Chemistry: Case Studies from the Pharmaceutical Industry
2. Green Chemistry in the Pharmaceutical Industry

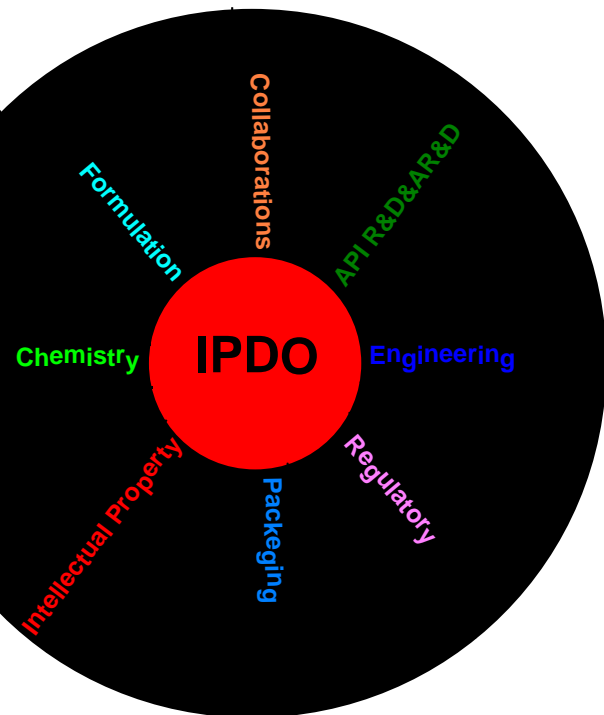
**Apart from Leading Journals**

1. OPRD
2. Journal of Chemical Education





Green Innovation Award-2013 in the Large MNC Category



*Amit Biswas*  
*Vilas Dhahanukar*  
*Apurba Bhattacharya*  
*Dhileep K. Krishnamurthy*  
*P. Pratap Reddy*  
*Arnab*  
*Amar*  
*Namrata*  
*Dinesh*  
*Gade Srinivas*  
*David (Clinton Foundation)*  
*Nagaraju*  
*Udai*  
*Sudhakar*  
*Kiran*  
*Prasanth*  
*Pallavi*  
*ERR*  
*Gangula*  
*Chirotech team*  
*Srinivas (DRILS)*  
*All the Chemistry, PE and Formulation scientists at IPDO*

Oliver (Regensburg)  
Marisa (Upenn)  
Burgess (College Station)  
Greg Fu (MIT)  
Mike Kirsche (UT Austin)  
G. Mehta (DRILS)  
J. Iqbal (DRILS)  
V.K.Tandon (LU)  
Hajela (CDRI)  
S.N.Rastogi (CDRI)



Department of Science and Technology



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**Thank You**