

# About OMICS Group

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OMICS Group is an amalgamation of [Open Access Publications](#) and worldwide international science conferences and events. Established in the year 2007 with the sole aim of making the information on Sciences and technology 'Open Access', OMICS Group publishes 500 online open access [scholarly journals](#) in all aspects of Science, Engineering, Management and Technology journals. OMICS Group has been instrumental in taking the knowledge on Science & technology to the doorsteps of ordinary men and women. Research Scholars, Students, Libraries, Educational Institutions, Research centers and the industry are main stakeholders that benefitted greatly from this knowledge dissemination. OMICS Group also organizes 500 [International conferences](#) annually across the globe, where knowledge transfer takes place through debates, round table discussions, poster presentations, workshops, symposia and exhibitions.

# OMICS International Conferences

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OMICS International is a pioneer and leading science event organizer, which publishes around 500 open access journals and conducts over 500 Medical, Clinical, Engineering, Life Sciences, Pharma scientific conferences all over the globe annually with the support of more than 1000 scientific associations and 30,000 editorial board members and 3.5 million followers to its credit.

OMICS Group has organized 500 conferences, workshops and national symposiums across the major cities including San Francisco, Las Vegas, San Antonio, Omaha, Orlando, Raleigh, Santa Clara, Chicago, Philadelphia, Baltimore, United Kingdom, Valencia, Dubai, Beijing, Hyderabad, Bengaluru and Mumbai.

# Identification of Aconitine Artefact in Alcoholic Extracts

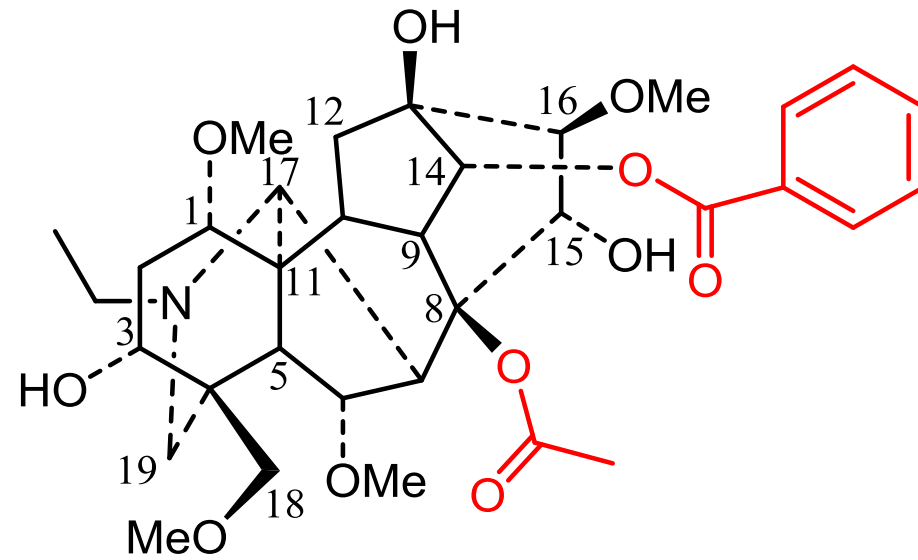
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DRUG DISCOVERY, FRANKFURT, GERMANY, AUGUST 2015

Mai Ahmed

# Aconitine

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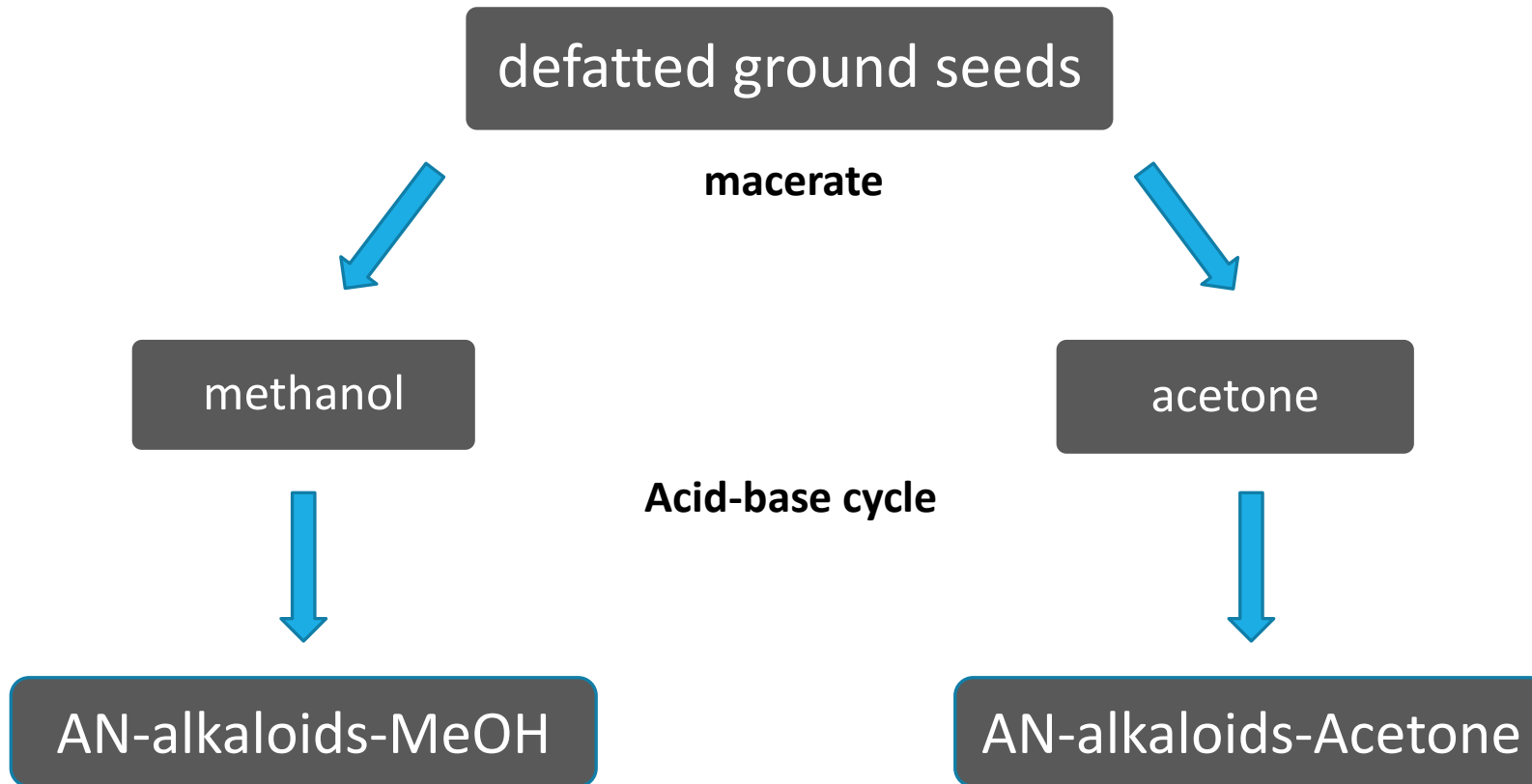
# *Aconitum napellus*

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# Extraction of diterpenoid alkaloids from *A. napellus* seeds

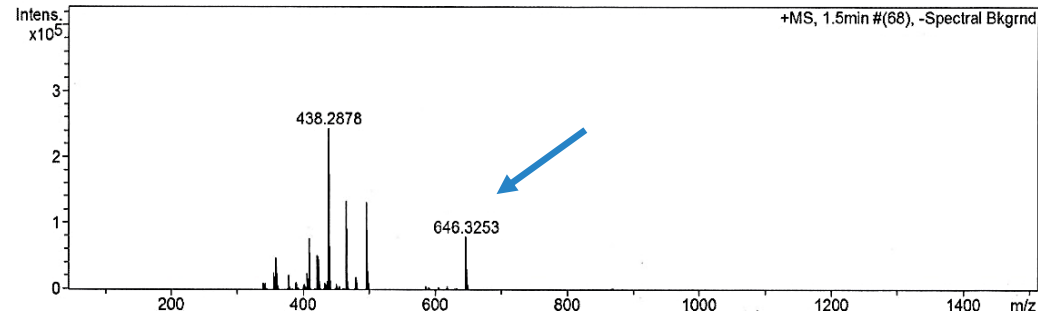
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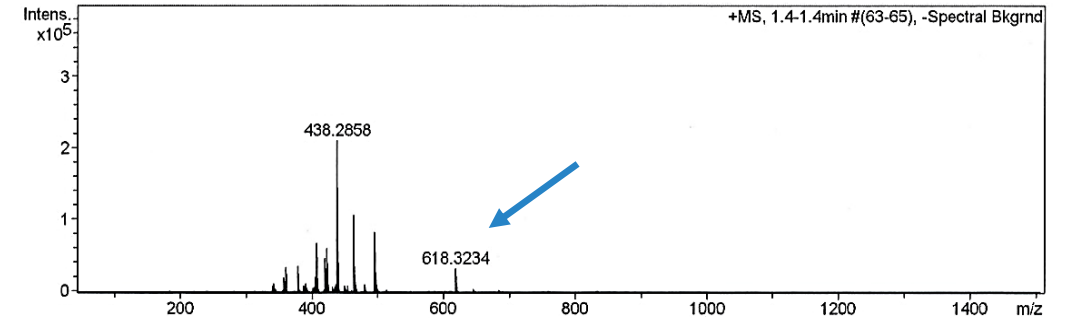
# HRMS data of *A. napellus* extracts

Both extracts showed the same alkaloidal profile except for an extra peak at  $MH^+$  618 in the methanolic extract.

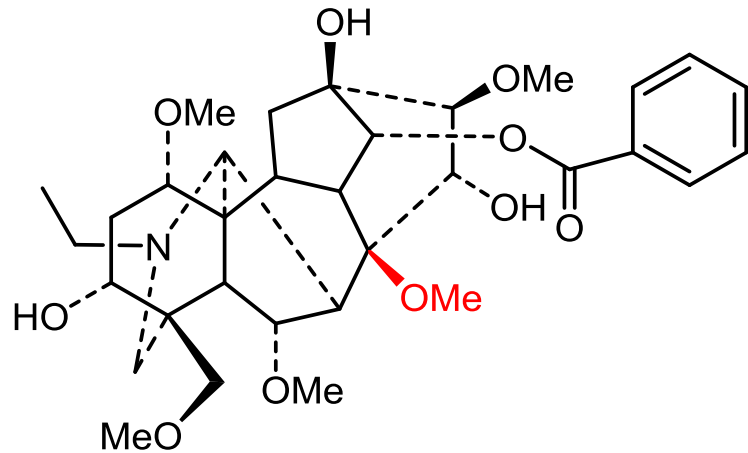
AN-alkaloids-MeOH



AN-alkaloids-Acetone



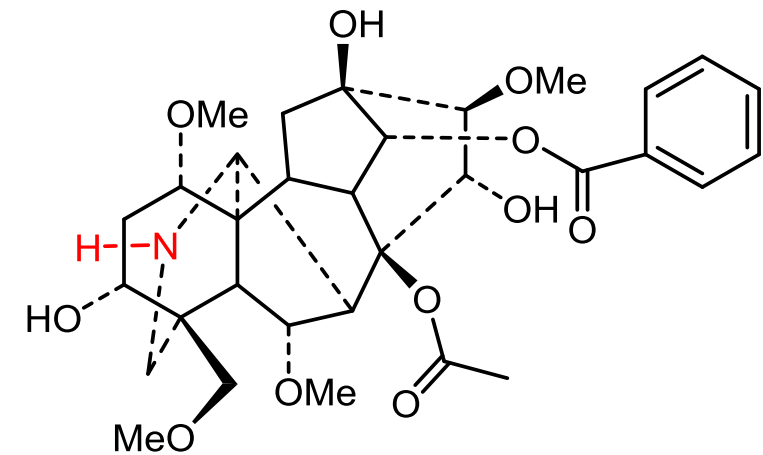
# HRMS data of the aconitine artefact



**14-O-benzoyl-8-O-methylnonaconine MH<sup>+</sup> 618.3278**

**C<sub>33</sub> H<sub>48</sub> N O<sub>10</sub>**

**-(CO)**



**N-deethylnonaconine MH<sup>+</sup> 618.68**

**C<sub>32</sub> H<sub>43</sub> N O<sub>11</sub>**

**-(C<sub>2</sub>H<sub>4</sub>)**



# Semi-synthesis of aconitine artefact

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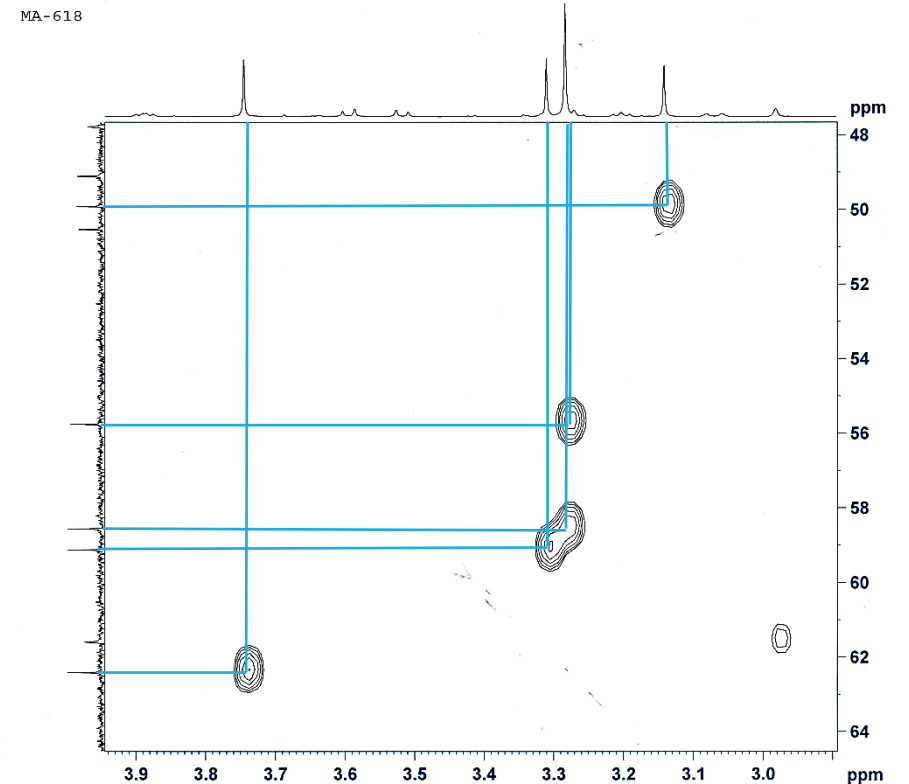
Aconitine was refluxed in methanol (at 65 °C for 6 hrs), purified by HPLC using C18 Gemini column 5 $\mu$  250 x 10 mm (mobile phase: acetonitrile : water with 0.3% conc. NH<sub>3</sub> [9:1 v/v], 3 mL/min; sample 100  $\mu$ L, UV detection at  $\lambda$  = 232 nm), R<sub>t</sub> 8 min. 10 mg were purified and NMR data were recorded in CDCl<sub>3</sub> using a Bruker 500 MHz NMR machine.

# NMR data of aconitine artefact

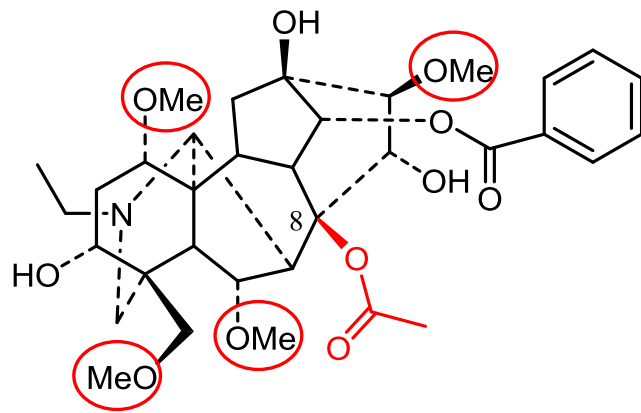
The  $^{13}\text{C}$ -NMR data revealed the presence of 33 carbon atoms while the DEPT experiment showed the presence of 5 methylenes, and 6 quaternary carbons.

The NMR data revealed the presence of *N*-ethyl fragment, 5 methoxyl groups and absence of acetoxy group.

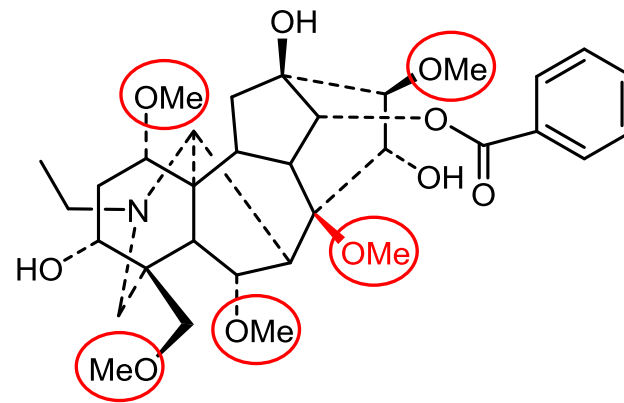
Quaternary C-8 resonates at  $\delta\text{C}$  82.4 in 14-*O*-benzoyl-8-*O*-methylnaconine and at  $\delta\text{C}$  92.2 in (8-acetoxy)aconitine.



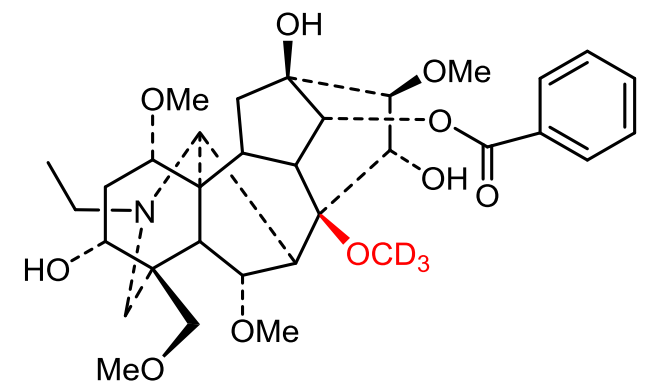
# Aconitine and its artefacts



**aconitine**



**artefact**



**deuterated artefact**

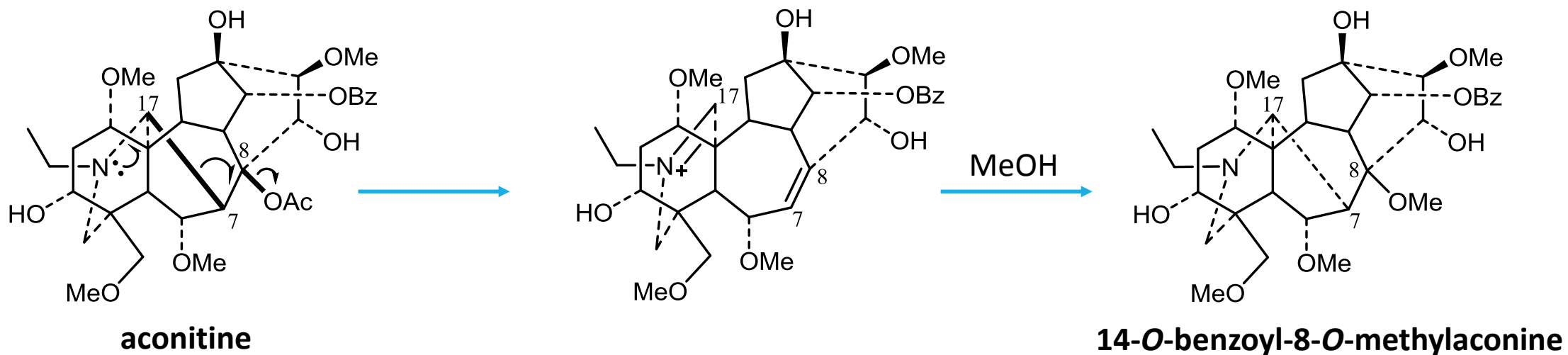
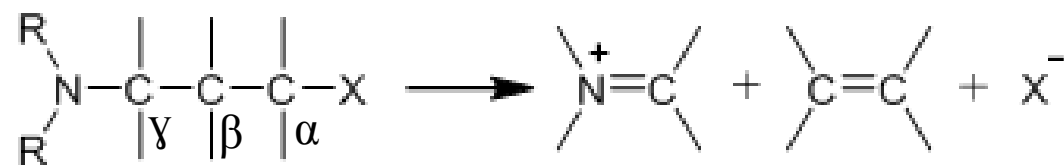
## Aconitine reacted with different alcohols

High Resolution Time-of-Flight mass spectra were obtained on a Bruker Daltonics micrOTOF spectrometer using electrospray ionisation (ESI).

Alcohol	Artefact	m/z (found)	Molecular formula
methanol	14- <i>O</i> -benzoyl-8- <i>O</i> -methylaconine	618.3291 Da	$C_{33}H_{48}NO_{10}$ requires 618.3278
$d_4$ -methanol	14- <i>O</i> -benzoyl-8- <i>O</i> -trideuteriated methylaconine	621.3490 Da	$C_{33}H_{45}D_3NO_{10}$ requires 621.3461
ethanol	14- <i>O</i> -benzoyl-8- <i>O</i> -ethylaconine	632.3459 Da	$C_{34}H_{50}NO_{10}$ requires 632.3429
$d_6$ -ethanol	14- <i>O</i> -benzoyl-8- <i>O</i> -pentadeuteriated ethylaconine	637.3759 Da	$C_{34}H_{45}D_5NO_{10}$ requires 637.3743

# Grob-type fragmentation

It takes place when the free electron pair of the nitrogen atom and C<sub>α</sub>-X bond are oriented anti-parallel to the C<sub>β</sub>-C<sub>γ</sub> bond which undergoes cleavage.



# Conclusions

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Alcoholic solvents should be avoided in any extraction procedure for aconitine and diester-diterpenoid alkaloids. Isolation of 8-*O*-alkylated diterpenoid alkaloids is most probably due to artefact formation. Medicinal formulations such as tinctures that involve extracts in ethanol are likely to suffer extensive degradation of diester alkaloids during preparation and/or storage and, on this occasion, be less toxic than the parent alkaloids.

# Let us meet again..

WE WELCOME YOU TO OUR FUTURE CONFERENCES OF OMICS  
INTERNATIONAL

2<sup>ND</sup> INTERNATIONAL CONFERENCE AND EXPO  
ON

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DRUG DISCOVERY & DESIGNING

ON

OCTOBER -31 NOVEMBER-02, 2016 AT ISTANBUL, TURKEY

[HTTP://DRUG-DISCOVERY.PHARMACEUTICALCONFERENCES.COM/](http://drug-discovery.pharmaceuticalconferences.com/)