## About OMICS Group

OMICS Group International is an amalgamation of Open Access publications and worldwide international science conferences and events. Established in the year 2007 with the sole aim of making the information on Sciences and technology 'Open Access', OMICS Group publishes 500 online scholarly journals in all open access aspects of Science, Engineering, Management and Technology journals. OMICS Group has been instrumental in taking the knowledge on Science & technology to the doorsteps of Research ordinary men and women. Scholars, Students, Libraries, Educational Institutions, Research centers and the industry are main stakeholders that benefitted greatly from this knowledge dissemination. OMICS International also organizes 500 International conferences annually across the globe, where knowledge transfer takes place through debates, round table discussions, poster presentations, workshops, symposia and exhibitions.

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OMICS International is a pioneer and leading science event organizer, which publishes around 500 open access journals and conducts over 400 Medical, Clinical, Engineering, Life Sciences, Pharma scientific conferences all over the globe annually with the support of more than 1000 scientific associations and 30,000 editorial board members and 3.5 million followers to its credit.

OMICS International has organized 500 conferences, workshops and national symposiums across the major cities including San Francisco, Las Vegas, San Antonio, Omaha, Orlando, Raleigh, Santa Clara, Chicago, Philadelphia, Baltimore, United Kingdom, Valencia, Dubai, Beijing, Hyderabad, Bengaluru and Mumbai. Comparative Metabolic Profiling of Halotolerant Bacterial Strains and Identification of Novel-Species Specific Metabolites

## G. Jayaraman

VIT University, Tamil Nadu, India



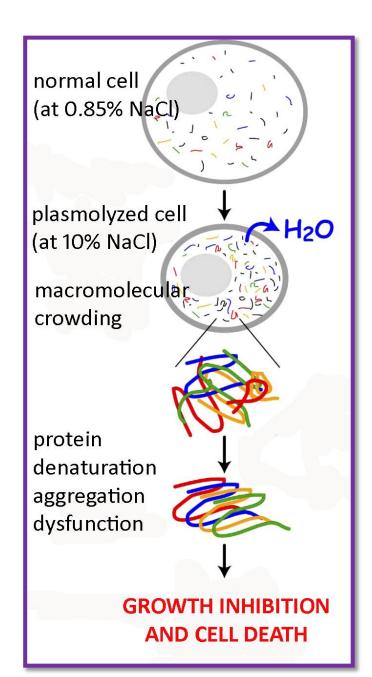
4<sup>th</sup> International Conference and Exhibition on Metabolomics & Systems Biology Philadelphia, USA

## The Research Focus

Saline environments - marine water, salt lakes, brines, salterns, saline soils, salted foods, etc.

Halophilic and halotolerant bacteria have developed two types of strategies (Galinski and Truper, 1994)

- **1.** Salt-in strategy K<sup>+</sup>, Cl<sup>-</sup>
- 2. Organic solute-in strategy polar neutral/zwitterionic organic compounds



# The 'Bacterial' Challenge

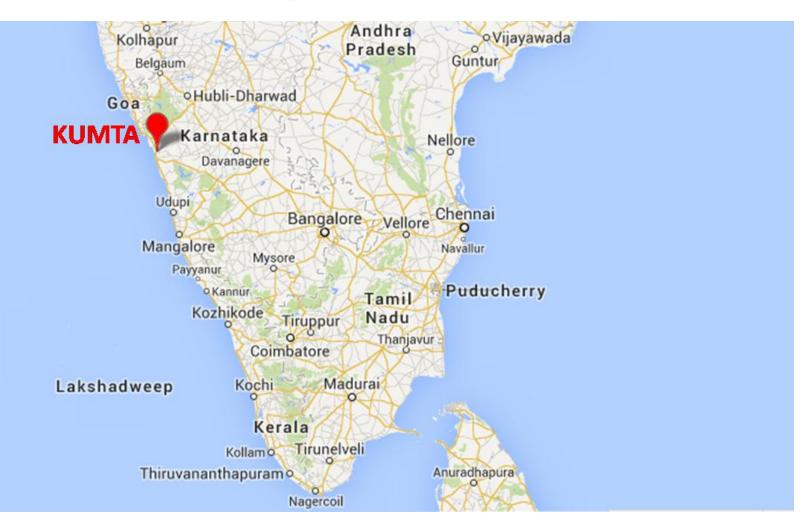
**Protein stability** – one of the most **challenging problems in halotolerant** bacteria during osmotic stress due to

- reduced water activity
- macromolecular crowding

**Compatible solutes help in protecting the protein stability and function under such stress conditions** – survival of these organisms

Maintaining protein stability during **various biotechnological processes** – protein expression, purification, formulations, storage, harsh conditions employed in various processes – food and chemical processing, detergents - a challenging and daunting task

# The Starting Point



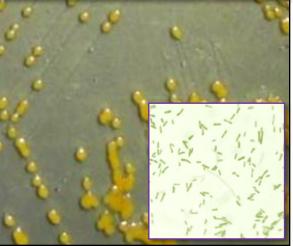


## The Isolates

# Halomonas hydrothermalis<br/>VITP09 (FJ743438)Bacillus<br/>VITP04Image: Strate in the strate in the

#### Virgibacillus dokdonensis VITP14 (HQ929427)

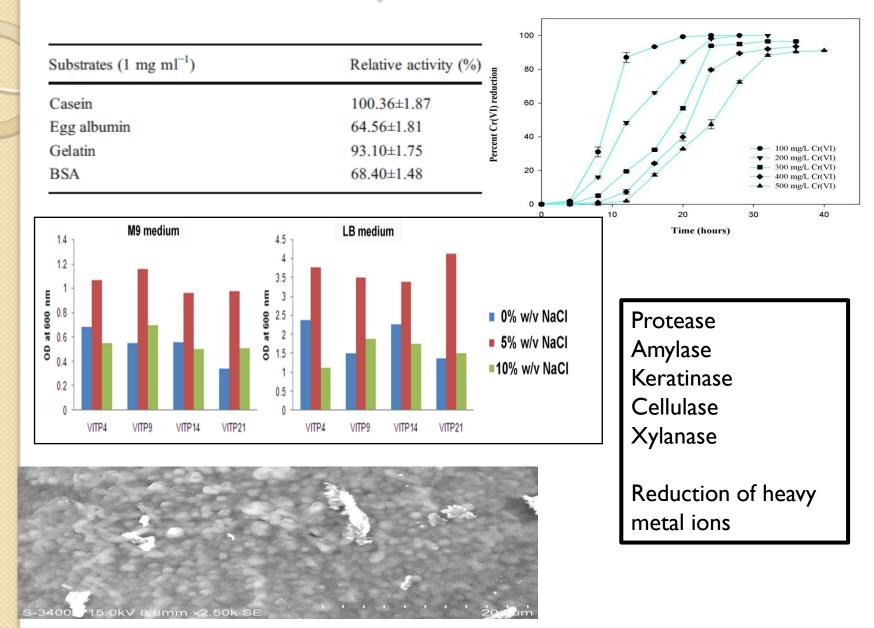
#### Bacillus aquimaris VITP04 (FJ687490)



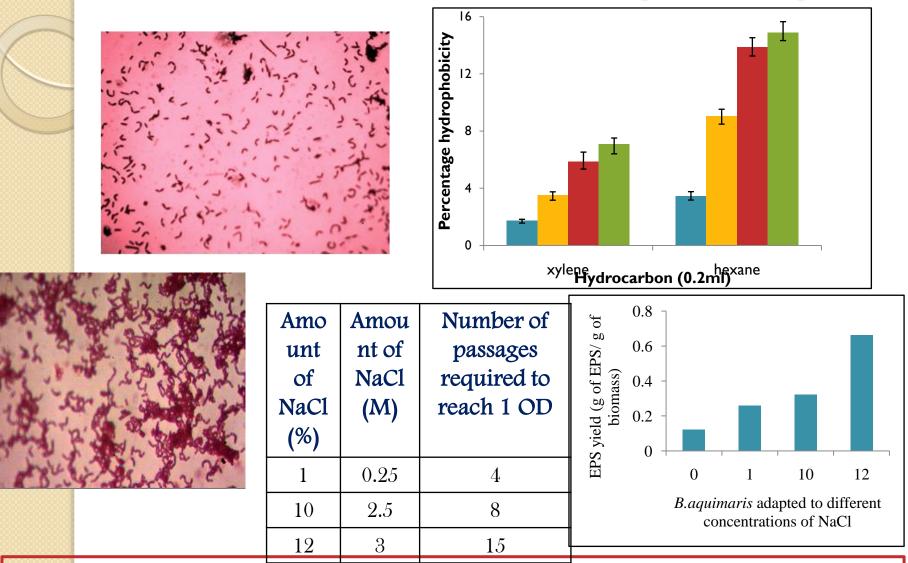
#### Planococcus maritimus VITP21 (HQ829429)



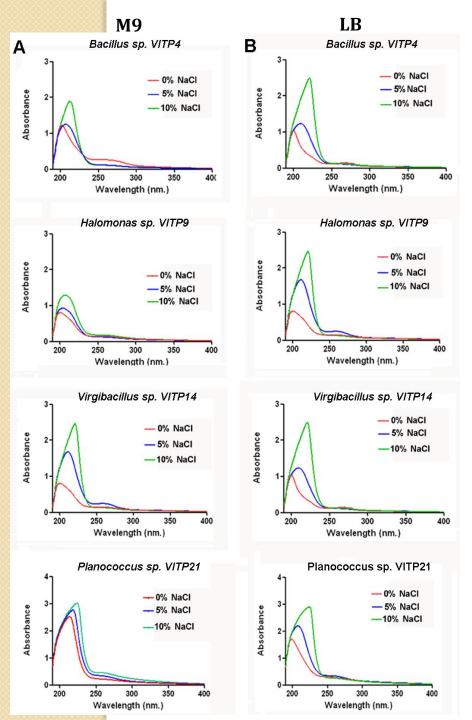
## Products & Adaptability



# Natural & Induced Adaptability



With shift in temperature and pH optimum and also increased stability under saline conditions

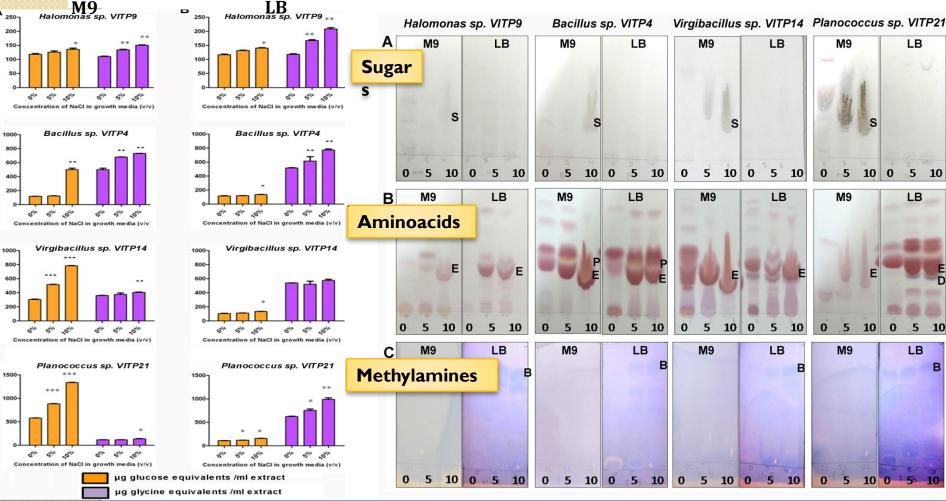


## SALT STRESS INDUCED BIOCHEMICAL CHANGES

Changes in the intracellular organic solute pools analyzed – UV spectra of the extracts (200 – 400 nm) - varied with respect to the salt concentration, growth media and the type of species.

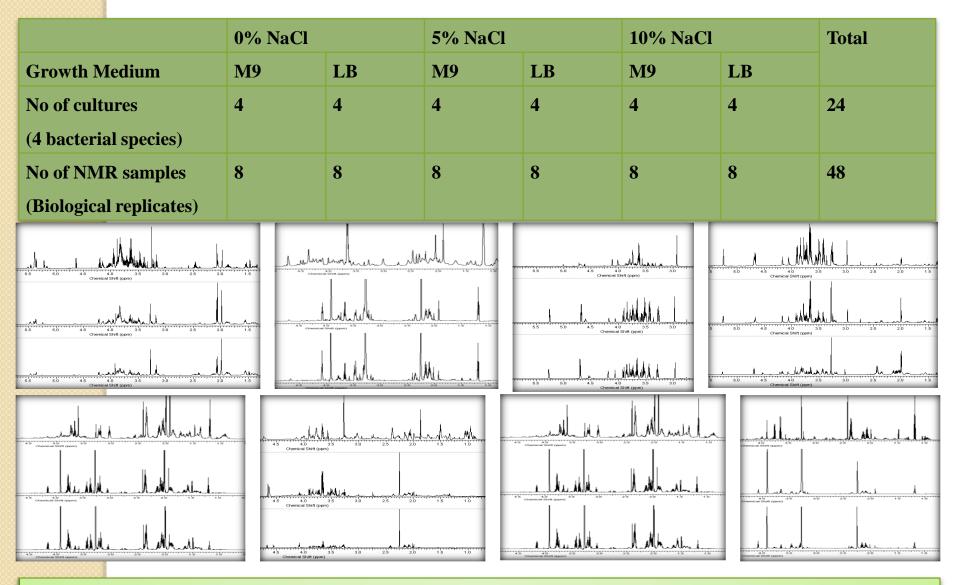
Could be assumed that the composition of the intracellular organic solutes changes with increasing salt concentrations, type of growth media and bacterial species.

#### SALT STRESS INDUCED BIOCHEMICAL CHANGES



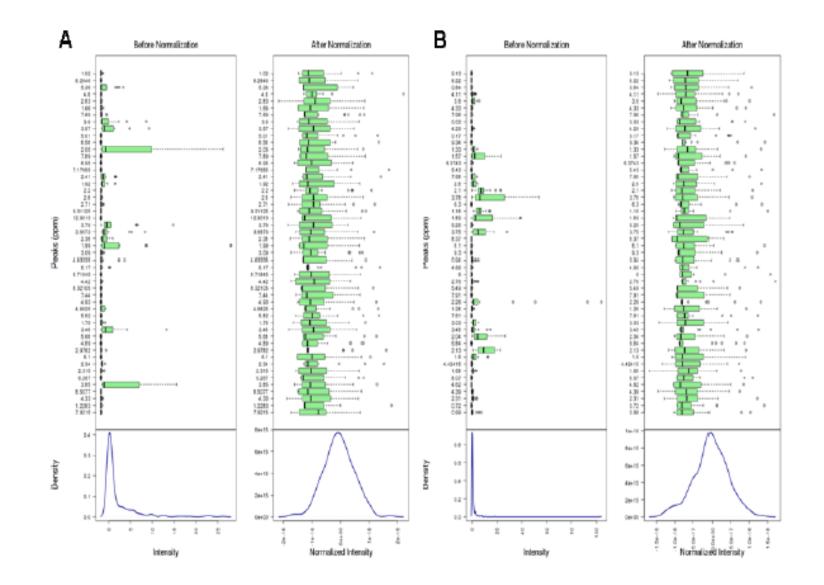
- Significant increase in intracellular levels of sugars and amino acids with increasing NaCl concentration in the growth medium
- Intracellular solute profile depended on the growth medium and species
- glutamic acid (E), proline (P), aspartic acid (D), betaine (B) identified to increase with increasing salt concentration
- Unidentified sugars and amino acids were also detected

#### SPECTRAL COMPLEXITY

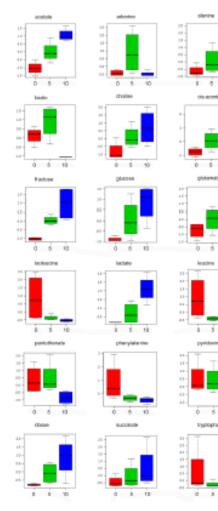


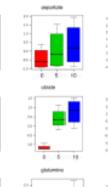
METABOANALYST 2.0 was used to analyse the data

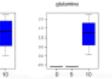
## The Data Normalisation

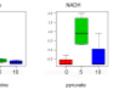


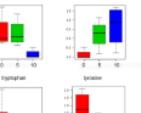
# **Concentration of Selected Metabolites**

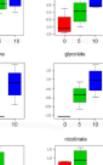


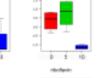


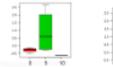


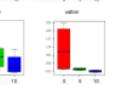


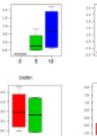






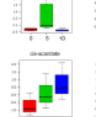


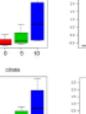




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over web



















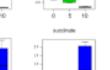












































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## Pathway Analysis

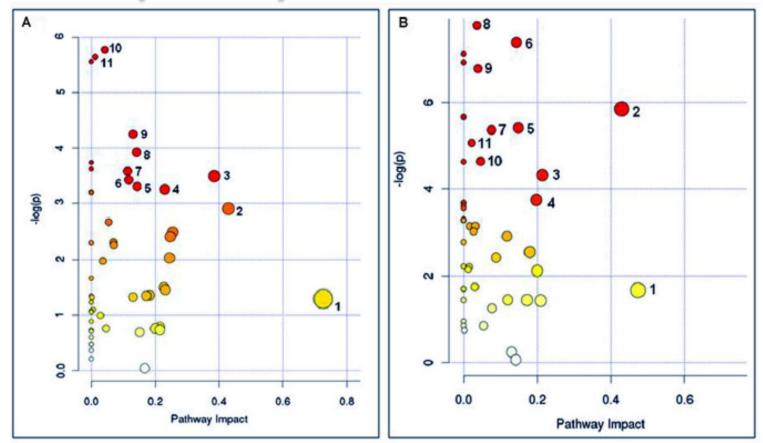
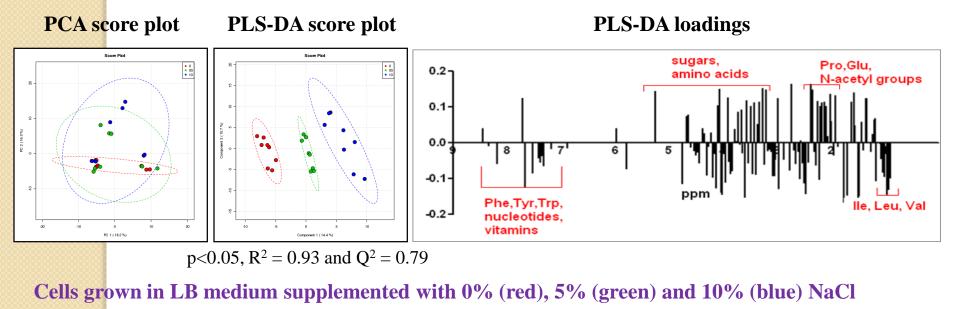


Figure 4.8 Summary of the pathway analysis for the salt-stress response of the cells grown in minimal medium (A) and complex medium (B). All the matched pathways are represented as circles. The color intensity and the radius of the circle indicate the p value and pathway impact value respectively. Key : 1. Alanine, aspartate and glutamate metabolism, 2. Pyruvate metabolism, 3. Tricarboxylic acid metabolism, 4. Glyoxylate metabolism, 5. Glycolysis metabolism, 6. Pentose phosphate pathway, 7. Valine, leucine and isoleucine metabolism, 8. Phenylalanine, tyrosine and tryptophan metabolism, 9. Pantothenate metabolism, 10. Biotin metabolism, 11. Glycerophospholipid, metabolism

#### CHEMOMETRIC ANALYSIS

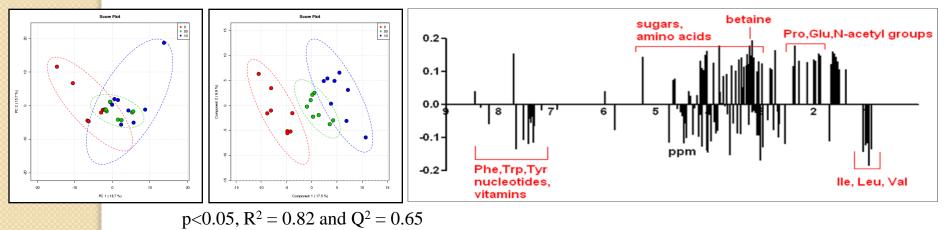
Cells grown in M9 medium supplemented with 0% (red), 5% (green) and 10% (blue) NaCl



PCA score plot

#### PLS-DA score plot

**PLS-DA** loadings

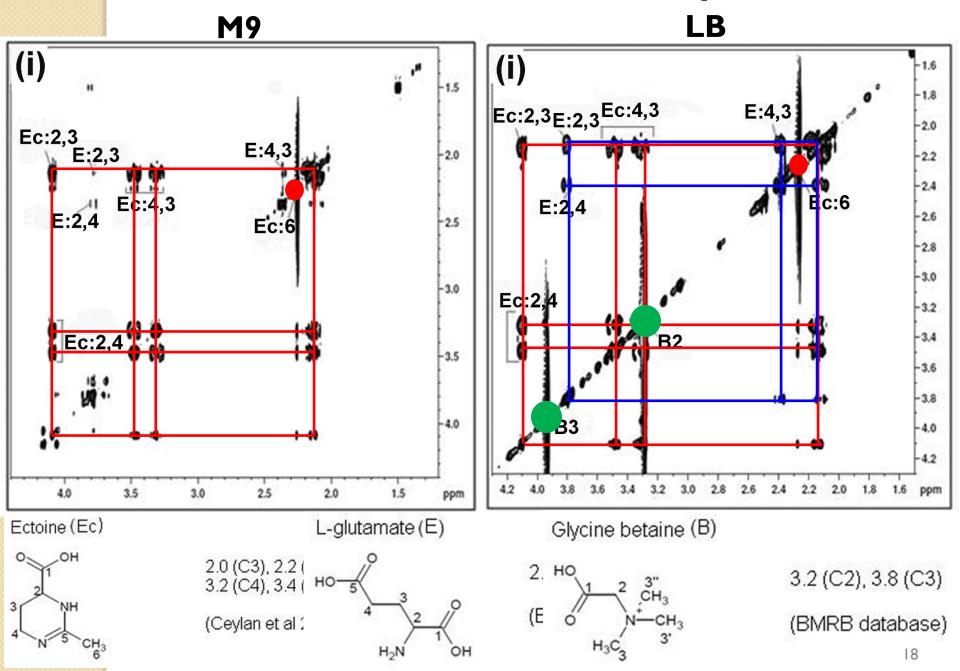


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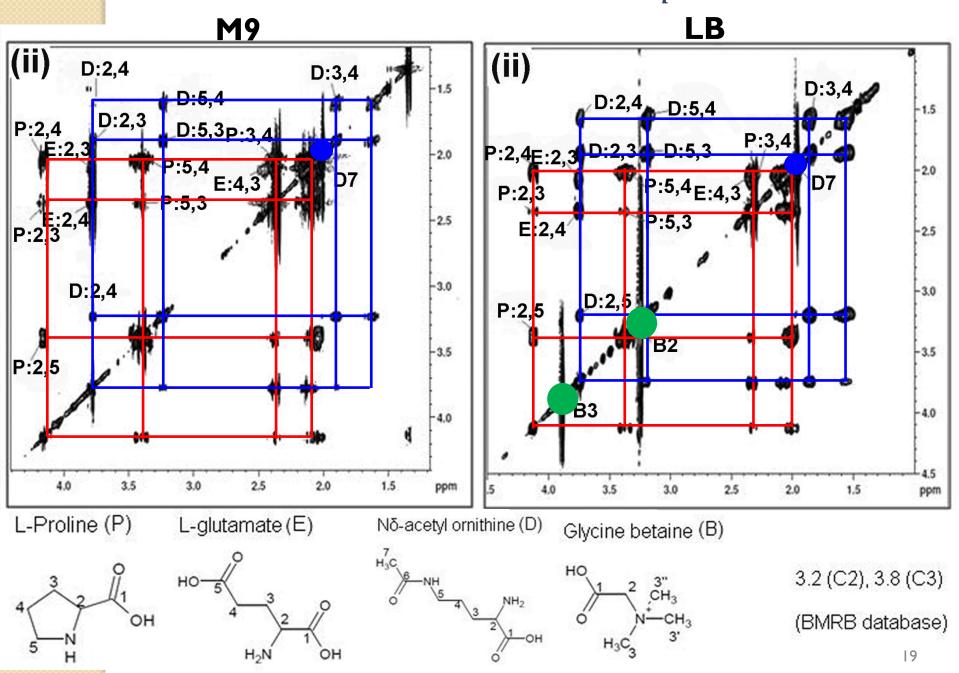
#### STRESS RESPONSIVE METABOLITES

Putative metabolites		Related pathway				
Up-regulated						
Amino acids Sugars	Aspartate	Alanine, aspartate and glutamate metabolism				
	Alanine	Alanine, aspartate and glutamate metabolism				
	Glutamine	Alanine, aspartate and glutamate metabolism				
	Glucose	Glycolysis metabolism, Pentose phosphate pathway				
	Fructose	Glycolysis metabolism				
	Ribose	Pentose phosphate pathway				
Organic acids	Pyruvate	Pyruvate metabolism, Tricarboxylic acid metabolism, Glycolysis metabolism, Glyoxylate metabolism				
	Citrate	Tricarboxylic acid metabolism, Glyoxylate metabolism				
	Succinate	Tricarboxylic acid metabolism, Glyoxylate metabolism				
	Cis-aconitate	Tricarboxylic acid metabolism, Glyoxylate metabolism				
	Formate	Pyruvate metabolism, Glyoxylate metabolism				
	Acetate	Pyruvate metabolism, Glycolysis metabolism				
	Lactate	Pyruvate metabolism				
	Glycolate	Glyoxylate metabolism				
Down-regulated						
Amino acids	Phenyl alanine	Phenylalanine, tyrosine and tryptophan metabolism				
	Tyrosine	Phenylalanine, tyrosine and tryptophan metabolism				
	Tryptophan	Phenylalanine, tyrosine and tryptophan metabolism				
	Valine	Valine, leucine and isoleucine metabolism				
	Isoleucine	Valine, leucine and isoleucine metabolism				
	Leucine	Valine, leucine and isoleucine metabolism				
Vitamins	Pantothenate	Pantothenate metabolism				
	Biotin	Biotin metabolism 7				

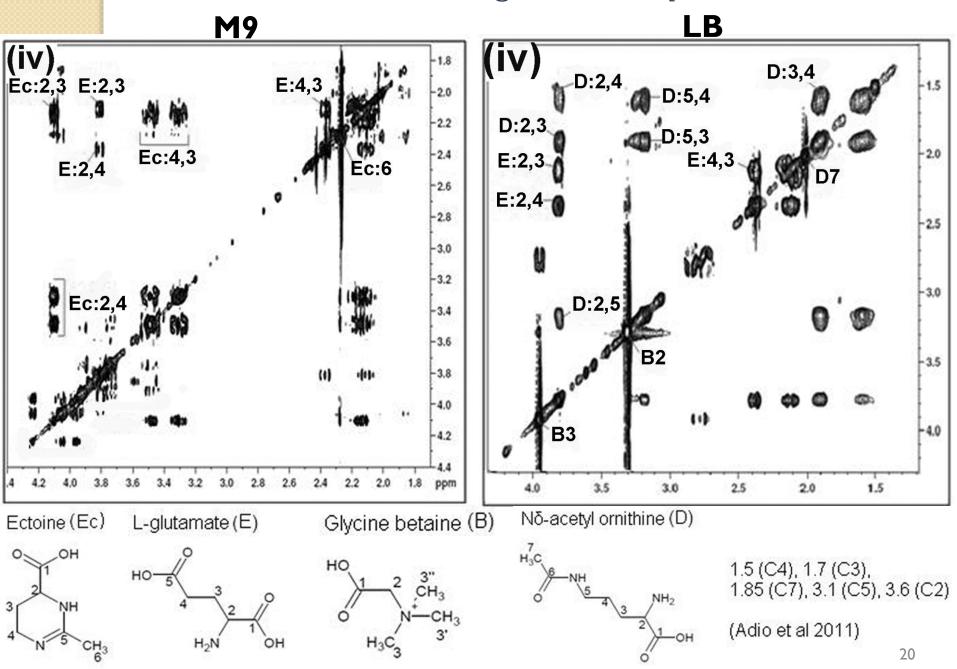
**TOCSY SPECTRUM OF** *Halomonas sp.* VITP9



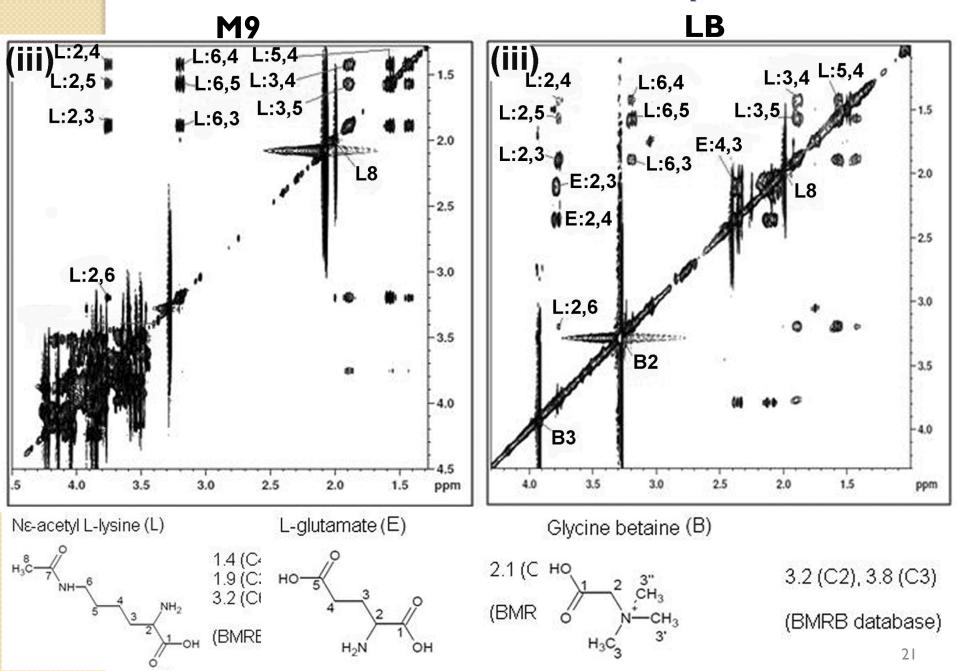
TOCSY SPECTRUM OF Bacillus sp. VITP4



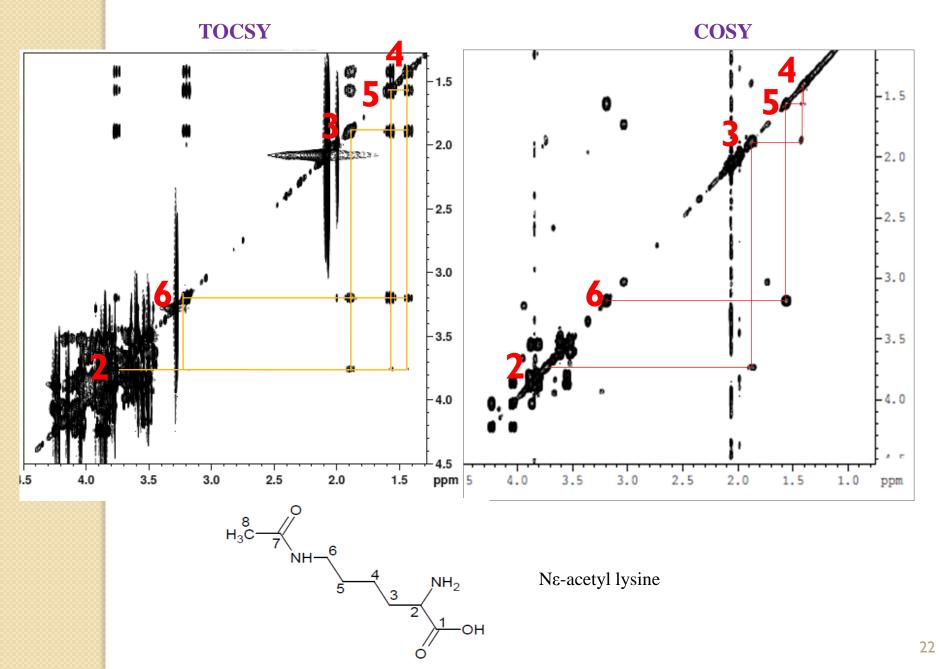
TOCSY SPECTRUM OF Virgibacillus sp. VITP14



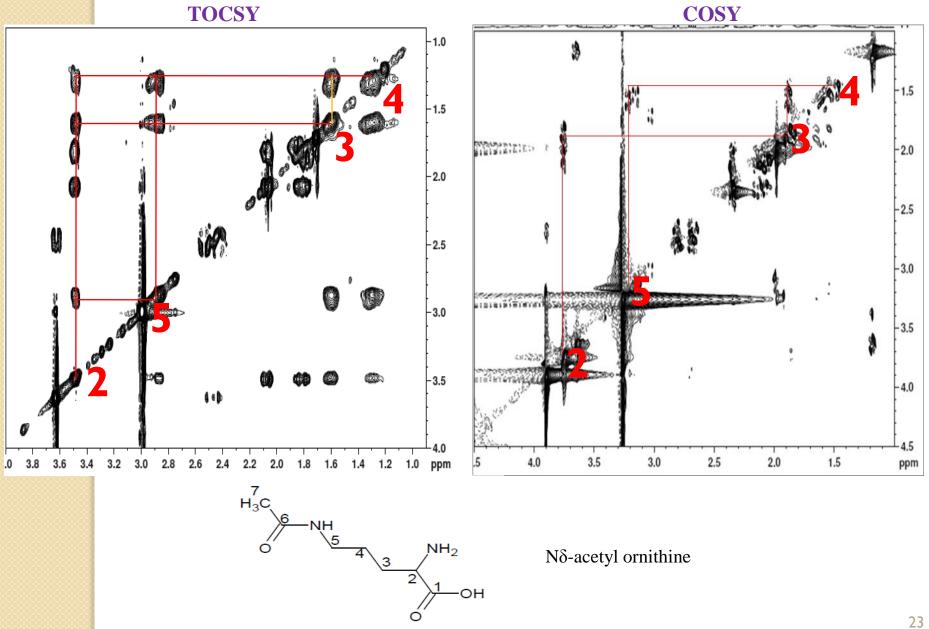
**TOCSY SPECTRUM OF** *Planococcus sp.* VITP21



#### Nε-acetyl lysine



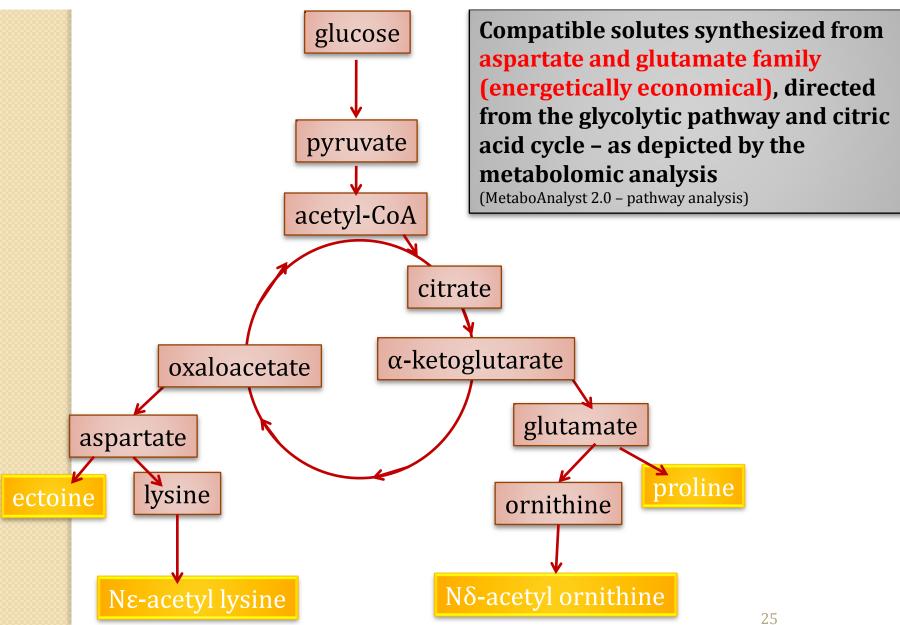
#### $N\delta$ -acetyl ornithine



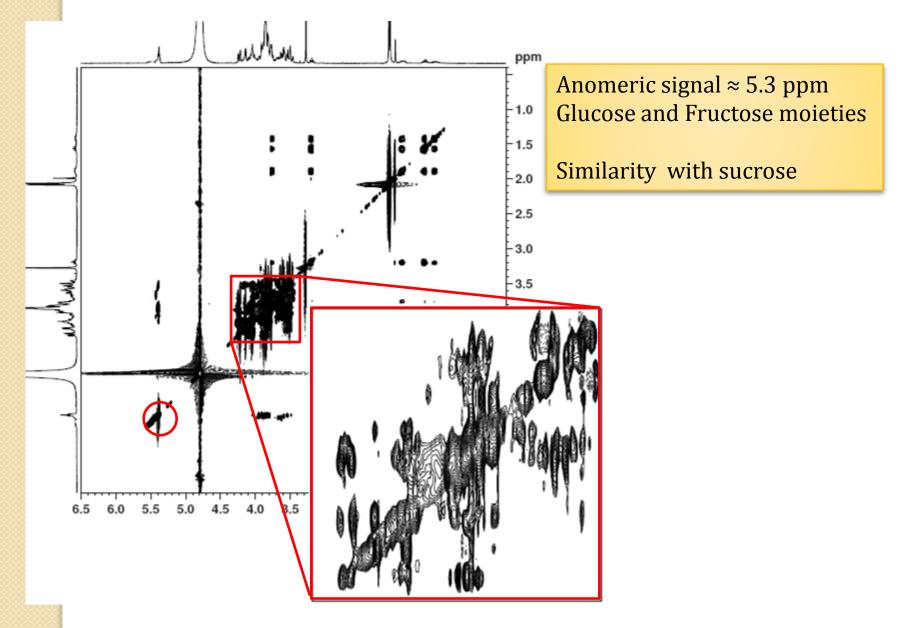
#### DISTRIBUTION OF OSMOLYTES IDENTIFIED WITHIN THE STRAINS

Bacterial strain	Minimal medium	At 0% w/v NaCl	At 5% w/v NaCl (mM)	At 10% w/v NaCl (mM)	Complex medium	At 0% w/v NaCl	At 5% w/v NaCl (mM)	At 10% w/v NaCl
Suain		(mM)				(mM)		(mM)
Halomonas					Glycine betaine	0	100.36±1.	137.52±3.
sp. VITP9	Ectoine	0	41.56±2.1	96.95±2.7	Ectoine	0	31.56±1.5	83.50±2.8
	Glutamate	0.88±0.2	26.51±2.1	54.62±6.2	Glutamate	1.95±0.1	8.60±0.9	14.90±0.5
	Aspartate	0	0.13±0.0	0.21±0.0	Aspartate	0.24±0.0	0.39±0.0	0.48±0.1
Bacillus sp.					Glycine betaine	0.33±0.3	65.57±2.2	83.75±1.7
VITP4	Proline	1.34±0.4	45.61±6.6	112.07±4.1	Proline	0	36.18±2.5	60.80±1.
	Nδ-acetyl ornithine	$0.40 \pm 0.0$	4.16±1.1	24.70±4.6	$N\delta$ -acetyl ornithine	0.62±0.1	28.20±1.7	36.45±0.7
	Ornithine	0.13±0.0	0.23±0.0	0.31±0.0	Ornithine	0	3.23±0.0	5.32±0.0
	Glutamate	1.34±0.0	66.01±1.4	98.59±1.9	Glutamate	18.64±2.3	36.62±2.3	44.71±0.4
	Aspartate	0	0.19±0.0	0.42±0.1	Aspartate	0	0.89±0.1	3.65±0.1
Virgibacillus	Ectoine	0	32.95±7.1	69.44±7.8	Glycine betaine	2.32±0.5	62.91±4.3	112.44±0.
sp. VITP14					Nδ-acetyl ornithine	2.25±0.3	6.25±1.5	12.79±2.5
					Ornithine	0.02±0.0	3.45±0.0	7.18±0.2
	Glutamate	7.35±2.0	26.85±1.3	64.65±4.7	Glutamate	11.62±0.5	21.37±1.9	37.42±2.2
	Aspartate	0.04±0.0	0.10±0.0	0.145±0.0	Aspartate	4.11±0.6	8.38±0.8	10.89±1.0
Planococcus					Glycine betaine	2.07±0.8	78.80±2.5	113.32±0.
sp.VITP21					Proline	0	4.36±0.9	36.05±1.4
	Nɛ-acetyl lysine	3.45±0.1	45.12±1.5	107.49±2.1	Nɛ-acetyl lysine	3.05±0.2	11.60±1.9	31.29±1.0
	Lysine	0.53±0.0	4.46±0.1	8.53±1.0	Lysine	3.04±0.1	5.34±0.5	10.32±0.5
	Glutamate	1.30±0.1	13.40±0.5	27.10±2.5	Glutamate	9.62±2.2	21.37±1.9	36.92±1.5
	Aspartate	0.76±0.5	1.38±0.4	2.86±0.2	Aspartate	5.91±0.0	12.81±0.0	15.94±0.1

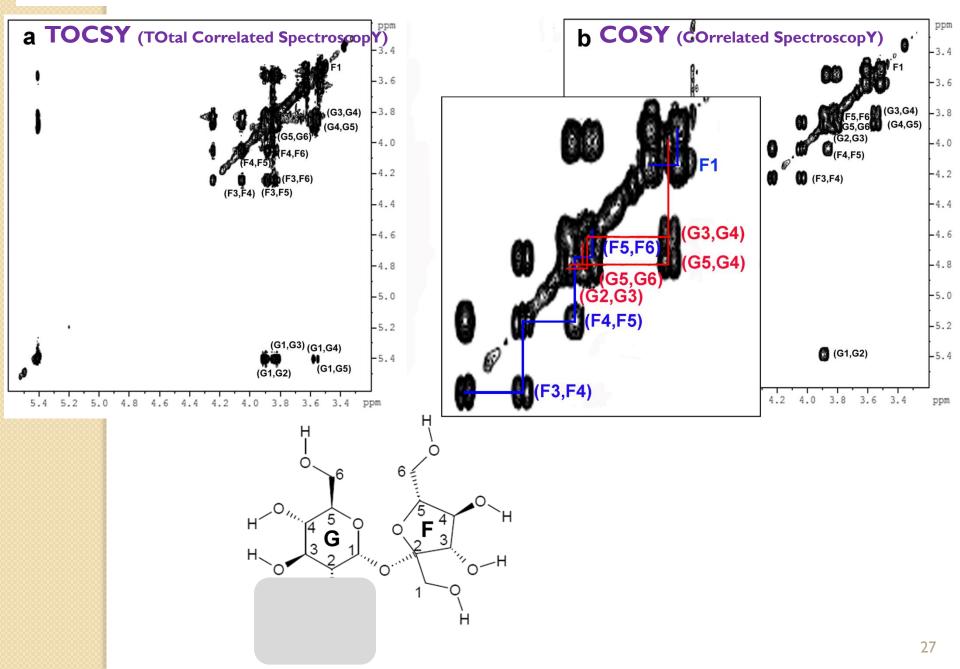
## METABOLIC PATHWAY ANALYSIS – SYNTHESIS OF NITROGENOUS OSMOLYTES



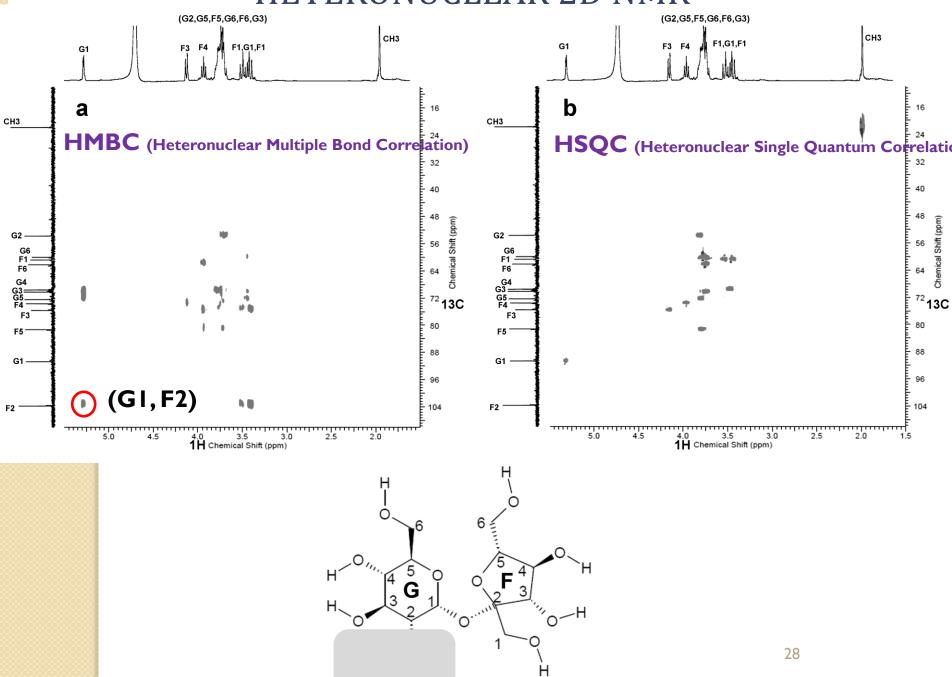
#### UNKNOWN SUGAR-OSMOLYTE FROM *Planococcus sp.* VITP21



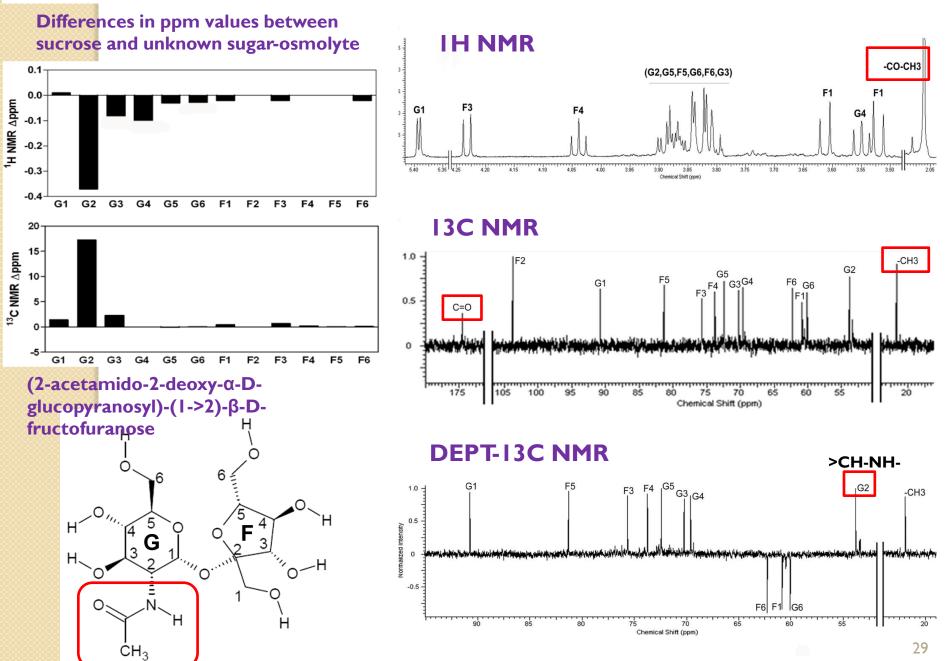
#### HOMONUCLEAR 2D NMR



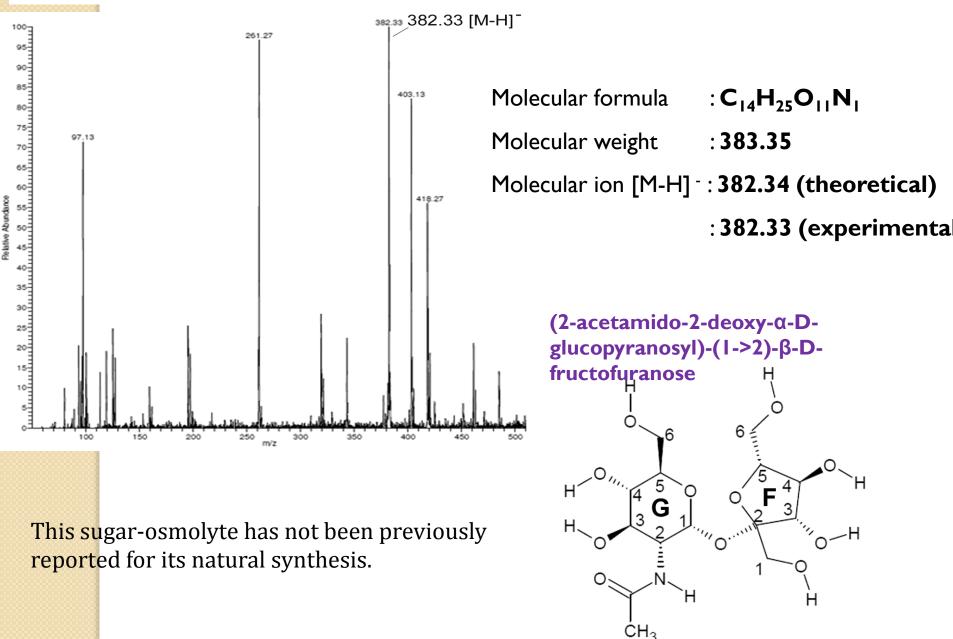
**HETERONUCLEAR 2D NMR** 



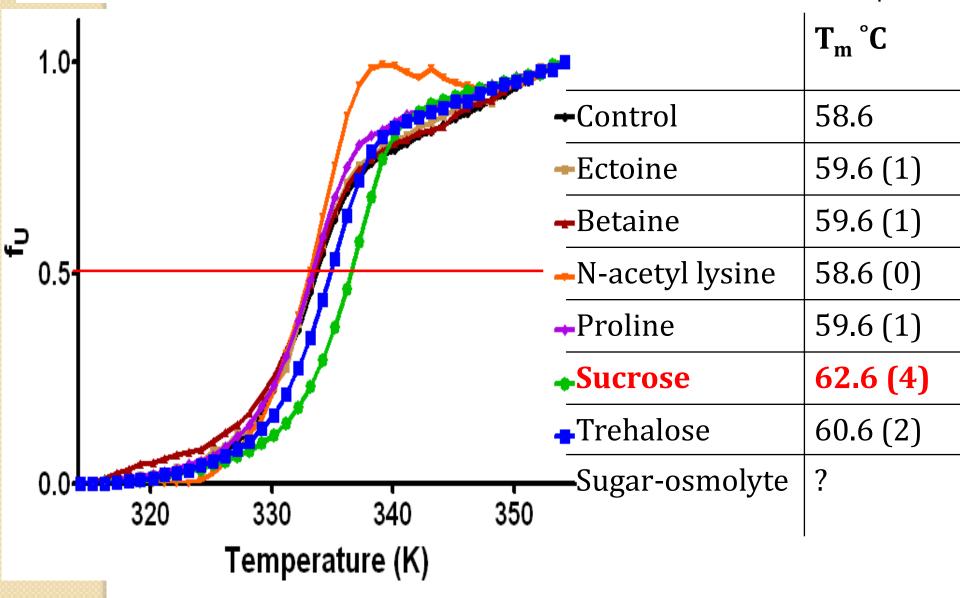
#### STRUCTURE ELUCIDATION OF THE UNKNOWN SUGAR-OSMOLYTE



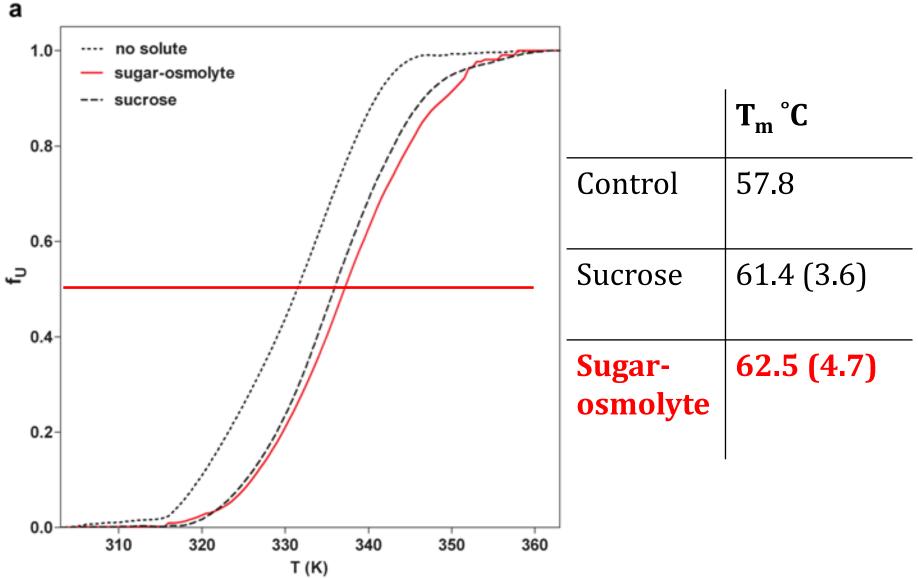
#### SUPPORTING INFORMATION – ESI-MS



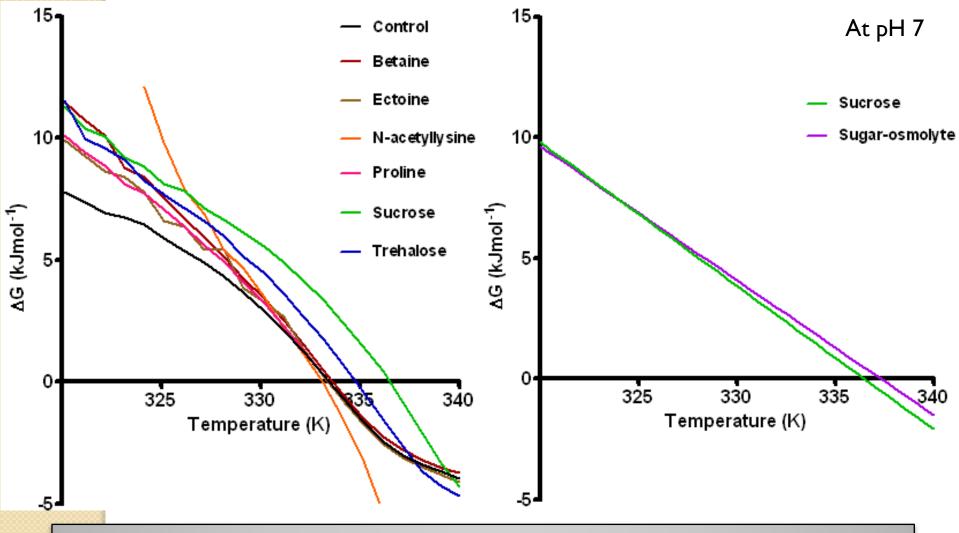
#### EFFECT OF THE SOLUTES ON PROTEIN STRUCTURAL STABILITY – FLUORESCENCE At pH 7



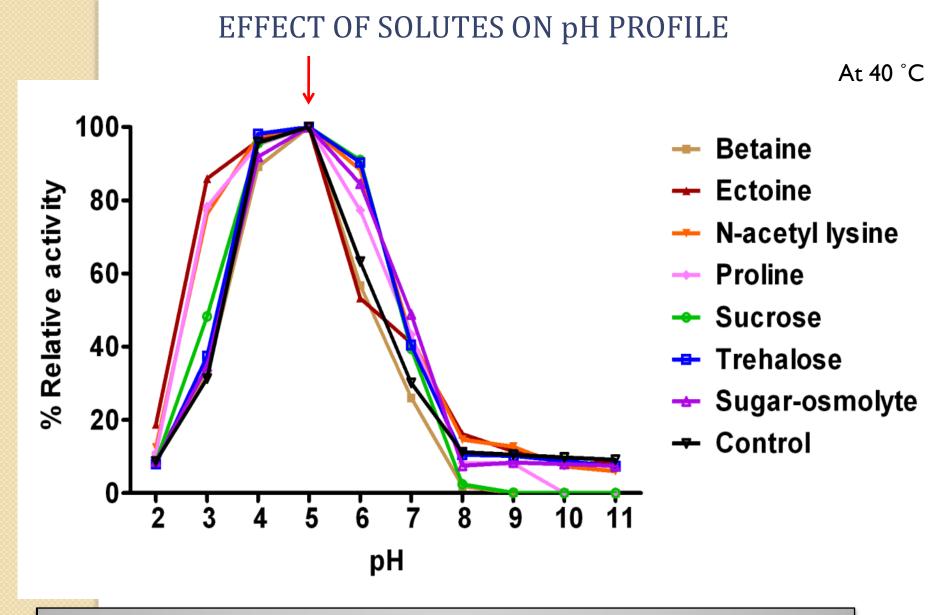
#### EFFECT OF THE SUGAR-OSMOLYTE ON PROTEIN STRUCTURAL STABILITY – CIRCULAR DICHROISM At pH 7



**EFFEC**T OF THE SOLUTES ON GIBBS FREE ENERGY OF UNFOLDING



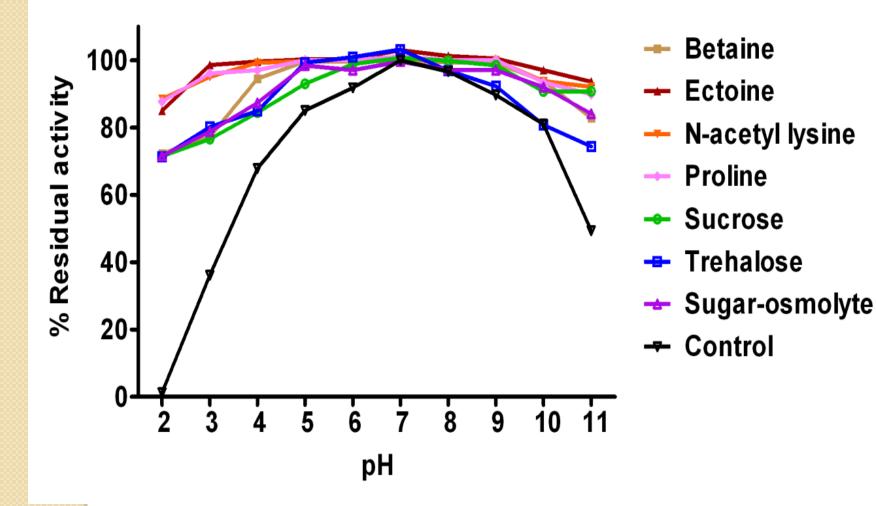
Osmolytes modulated protein function by manipulating their stability – increased the  $T_m$  and  $\Delta G_U$ 



**Optimum activity at pH 5** 

#### EFFECT OF SOLUTES ON pH STABILITY

At 40 °C



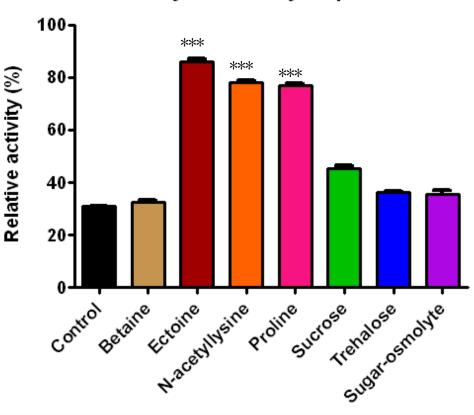
Solutes increased stability at all pH ranges

#### STRUCTURAL STABILITY AT pH 5

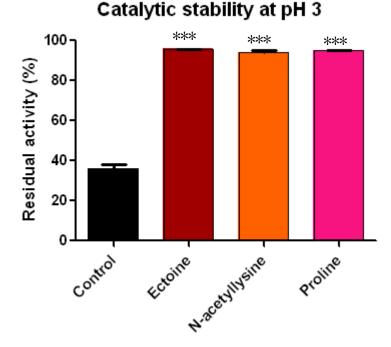
	T <sub>m</sub> °C	T <sub>agg</sub> °C
Control	50.5	42.1
Ectoine	52.5 (2)	49.7 (7.6)
Betaine	52.5 (2)	45.4 (3.3)
N-acetyl lysine	51.5 (1)	44.9 (2.8)
Proline	51.5 (1)	44.1 (2)
Sucrose	59.6 (9)	50.8 (8.7)
Trehalose	53.5 (2.9)	45.7 (3.6)
Sugar-osmolyte	60.9 (10.4)	46.2 (4.1)

Improved the structural stability - shifting the N↔D equilibrium towards left by delaying unfolding - increase in T<sub>m</sub> by inhibiting aggregation – increase in T<sub>agg</sub>

#### IMPROVED ACTIVITY AND STABILITY AT ACIDIC pH 3



Catalytic activity, catalytic stability and structural stability are improved by solutes –N-acetyl lysine

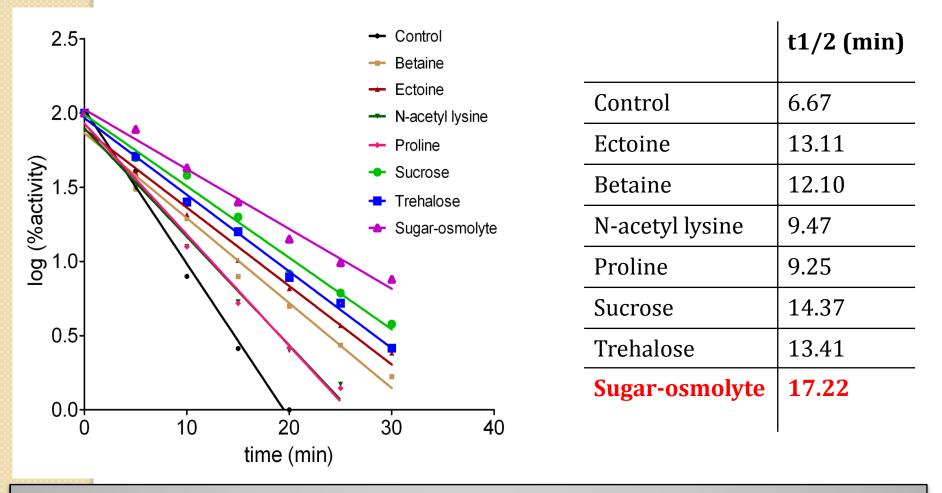


#### Structural stability at pH 3

	, <b>,</b> ,	
	Tm °C	
Control	30.5	
Ectoine	40.5 (10)	
N-acetyl lysine	41.5 (11)	
Proline	34.5 (4)	
	37	

Catalytic activity at pH 3

#### OPERATIONAL STABILITY – INACTIVATION KINETICS AT 50°C AND pH 5



• Proof-of-concept for osmolyte induced protein stability under denaturing stress conditions

• Such comparative studies result in the identification of suitable solutes for improving the activity and stability of the protein of interest 38

## SALIENT FEATURES OF THE STUDY

- 1. Synthesis / accumulation of species specific and growth media dependent compatible solutes to achieve halotolerance
- 2. Changes in the levels of metabolites of the aspartate and glutamate metabolic pathway to restore homeostasis
- Two strains (*Planococcus maritimus* VITP21 and *Bacillus aquimaris* VITP4) capable of synthesizing rarely occurring N-acetylated diamino acids, Nε-acetyl lysine and Nδ-acetyl ornithine were reported.
- 4. A novel sugar osmolyte, (2-acetamido-2-deoxy- $\alpha$ -D-glucopyranosyl)-(1 $\rightarrow$ 2)- $\beta$ -D-fructofuranose, synthesized de-novo by *Planococcus maritimus* VITP21 was discovered.
- **5.** These compatible solutes could enhance the stability of proteins

#### Acknowledgement

#### **Scholars**

Pooja

Nidhya

Jabeen Thaz

Lavanya

#### **Facilities**

TIFR, Mumbai

IISc, Bangalore

SIF, VIT

**VIT University** 

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